# Chapter 12 Regulating Services of Bivalve Molluscs in the Context of the Carbon Cycle and Implications for Ecosystem Valuation



R. Filgueira, T. Strohmeier, and Ø. Strand

Abstract The role of marine bivalves in the  $CO_2$  cycle has been commonly evaluated as the balance between respiration, shell calcium carbonate sequestration, and  $CO_2$  release during biogenic calcification; however, this individual-based approach neglects important ecosystem interactions that occur at the population level, e.g. the interaction with phytoplankton populations and benthic-pelagic coupling, which in turn can significantly alter the  $CO_2$  cycle. Therefore, an ecosystem approach that accounts for the trophic interactions of bivalves, including the role of dissolved and particulate organic and inorganic carbon cycling, is needed to provide a rigorous assessment of the role of bivalves as a potential sink of  $CO_2$ . Conversely, the discussion about this potential role needs to be framed in the context of non-harvested vs. harvested populations, given that harvesting represents a net extraction of matter from the ocean. Accordingly, this chapter describes the main processes that affect  $CO_2$ cycling and discuss the role of non-harvested and harvested bivalves in the context of sequestering carbon. A budget for deep-fjord waters is presented as a case study.

Abstract in Chinese 摘要:海水双壳贝类在二氧化碳循环中的作用通常根据 基于呼吸作用、钙化作用和钙化期间二氧化碳释放进行评价。 然而,这种基 于个体的评估方法并没有考虑种群水平的贝类与生态系统的相互作用。例 如,贝类与浮游植物种群和底栖生物的相互作用,这种相互作用可以明显改变 CO<sub>2</sub>循环过程。因此,需要建立一套综合考虑溶解有机碳、溶解无机碳、颗粒 有机碳、颗粒无机碳等碳存在形态的生态系统方法来评估双壳贝类潜在的碳 汇作用。然而,关于这种潜在作用的讨论需要在区分自然种群和养殖种群的 情况下进行,因为养殖种群的最终收获其实是从海水中进行相关营养成分的 净提取。因此,本章介绍了影响CO<sub>2</sub>循环的主要过程,并讨论了自然和养殖的 双壳贝类在碳移除过程中的作用。在挪威峡湾内的一个双壳贝类养殖区的碳 收支会作为一个案例研究进行展示。

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关键词 双壳贝类 • 二氧化碳 • 碳循环 • 碳交易系统

## 12.1 Introduction

Bivalves are soft-bodied organisms protected by an external shell consisting of two hinged valves. The ratio shell:tissue in terms of weight is different across species and is habitat dependent within species (e.g. Newell and Hidu 1982; Rodhouse et al. 1984; MacDonald and Thompson 1985; Penney et al. 2008). For example, mussels cultured in suspended structures tend to have lighter shells than those in natural populations, which could be related to the feeding conditions in aquaculture facilities promoting faster growth and thinner shells (Aldrich and Crowley 1986), but also to the reduced predation pressure (Lowen et al. 2013). The shells of cultured bivalves can generally be considered residues although they are sometimes used as by-products in construction and agriculture (e.g. Rodríguez Álvaro et al. 2014; Varhen et al. 2017). Taking into account the global annual production of cultured bivalves is ~14 x10<sup>6</sup> tons, including clams, cockles, oysters, mussels and scallops (www.fao.org reporting 2015 data) and assuming an average contribution of shell to total body weight of 50% (general ballpark figure given that this varies greatly between species), shell represents a residue (potential by-product) of  $\sim 7 \times 10^6$  tons, of which 95% is calcium carbonate.

The shell is an exoskeleton that offers protection against predators and adverse environmental conditions. Adductor muscles are attached to the shell providing the animal with the capability to close their valves, isolating the internal tissues from the environment, although the effectiveness varies across species. In the case of scallops, the rapid contraction of the adductor muscle forces the valves to quickly squeeze the intervalvar fluid, which creates a water jet that propels the scallop, providing them with swimming capabilities (Guderley and Tremblay 2016). The different shell shapes across bivalve species allowed this class of molluscs to colonize a variety of habitats (Stanley 1970). Marine bivalves are widely distributed from tropical to boreal waters, and can be found inhabiting a variety of substrates, ranging from rocky to soft bottoms, infaunal and epifaunal. Most marine bivalves are suspension-feeders and can reach high densities in the wild, e.g. oyster reefs or mussel beds. At high density, they are ecosystem engineers (sensu Jones et al. 1994). Bivalves can modify the physical environment, for example by preventing erosion (Jones et al. 1994). They can also modify the available resources for other species, by controlling phytoplankton populations and/or altering nutrient cycling (Mann and Powell 2007; Filgueira et al. 2015). Consequently, the effects of bivalves on biogeochemical cycles goes beyond the individual scale. Accordingly, an ecosystem scale approach in which these feedbacks are included becomes imperative when studying the implications of marine bivalves in biogeochemical cycles.

The role of bivalves as ecosystem engineers and the need for an ecosystem approach become even more relevant when bivalves are cultured at high densities. Although the same ecosystem process can be conceptually applied to wild and cultured populations, the higher densities in aquaculture sites can significantly alter the magnitude of biogeochemical fluxes. For example, although cultured bivalves can exert a bottom-up nutrient control in stimulating primary production (Cranford et al. 2007; Jansen 2012), this positive effect is density dependent, with a resulting high bivalve biomass causing a reduction in primary production (Burkholder and Shumway 2011; Smaal et al. 2013). Given their ideal growing conditions, growth rates of cultured populations are usually higher than for wild populations; however, the most critical aspect of cultured bivalves is that their biomass is extracted from the ocean, a relevant consideration when comparing the role of wild versus cultured populations in biogeochemical cycles. The shells of wild bivalves will eventually dissolve in seawater, but those of cultured bivalves may end up on land. Note that some wild populations may also end up on land when they are commercially exploited (e.g. mussel or scallop dredging). Therefore, in this chapter bivalves will be considered according to two main categories: non-harvested (wild populations that are not harvested) and harvested (cultured and wild populations that are harvested). Separation of non-harvested and harvested populations is critical when evaluating the role of bivalves from each group in the CO<sub>2</sub> cycle and, in general, when valuing ecosystem services.

The goal of this chapter is to describe the role of bivalves in the  $CO_2$  cycle with special emphasis on the specific role of their shells and the implications for ecosystem services valuation. To achieve this, the chapter has been structured accordingly:

- The role of calcifying organisms in the CO<sub>2</sub> budget.- which describes the chemistry of shell formation.
- The influence of organic carbon on CO<sub>2</sub> fluxes.- which describes the main processes involving organic carbon that are relevant to the CO<sub>2</sub> cycle.
- Ecosystem services of non-harvested and harvested populations.- which describes the implications of harvesting bivalves as a food source in the context of a holistic valuation of ecosystem services.
- Case study Norwegian cultured mussels.- in which the rationale described in previous sections is applied to the case of Norwegian cultured mussels.
- Conclusions.- which summarizes the most relevant findings of the chapter.

# 12.2 The Role of Calcifying Organisms in the CO<sub>2</sub> Budget

Calcifying organisms are directly involved in two processes that release  $CO_2$ . First,  $CO_2$  is released via the catabolism of ingested organic matter:

$$CH_2O + O_2 \rightarrow CO_2 + H_2O \tag{12.1}$$

and, second, it is released via calcium carbonate (CaCO<sub>3</sub>) formation by biogenic calcification:

$$Ca^{2+} + 2HCO_3^{-} \leftrightarrow CaCO_3 + CO_2 + H_2O$$
 (12.2)

This release of CO<sub>2</sub> also induces shifts in the carbonate system:

$$\operatorname{CO}_2 + \operatorname{H}_2 \operatorname{O} \leftrightarrow \operatorname{H}_2 \operatorname{CO}_3 \leftrightarrow \operatorname{H}^+ + \operatorname{HCO}_3^- \leftrightarrow 2\operatorname{H}^+ + \operatorname{CO}_3^{2-}$$
 (12.3)

These processes depend on environmental conditions such as pH, alkalinity, salinity, and temperature (Millero 1995; Lerman and Mackenzie 2005; Dickson 2010; Mackenzie and Andersson 2013).

The balance between the  $CO_2$  released in respiration and biogenic calcification and the net C sequestered as calcium carbonate have been used to evaluate the role of bivalves in the  $CO_2$  cycle. The available studies in which these processes have been quantified for bivalves is reviewed in Table 12.1. The units from the different studies have been converted to g C m<sup>-2</sup> y<sup>-1</sup> for comparative purposes (conversion factors: 12 g C in 100 g CaCO<sub>3</sub>; 12 g C in 1 mol CO<sub>2</sub>). With the exception of the estimations from Hily et al. (2013), all other studies suggest that sequestration minus biocalcification and respiration is negative (Table 12.1), which suggests that bivalves are net generators of  $CO_2$ . Hily et al. (2013) suggested that under specific environmental conditions Crassostrea gigas and Mytilus edulis can sequester carbon effectively after accounting for biocalcification and respiration. The disagreement between Hily et al. (2013) and the other studies (Table 12.1) seems to be related to the respiration flux in Hily et al. (2013) which is especially obvious when comparing the ratio between sequestration and respiration. The respiration values in Hily et al. (2013) are extremely low compared to the other studies (Table 12.1) when considering the carbon that is sequestered in the shell. This is even more striking given the fact that most of these studies, including Hily et al. (2013), use the same empirical equation proposed by Schwinghamer et al. (1986) to estimate respiration. Nevertheless, aside from Hily et al. (2013), the level at which bivalves release  $CO_2$  is species dependent, with a net carbon release ranging from 0.35 to 2.45 gC m<sup>-2</sup>year<sup>-1</sup> per 1 gC m<sup>-2</sup>year<sup>-1</sup> (Table 12.1). The results of several studies (see Table 12.1) demonstrate that bivalves are  $CO_2$  generators when the balance strictly focuses on this inorganic form of carbon at the individual level.

Solely from the individual perspective, it makes sense that a filter feeder is a net generator of CO<sub>2</sub>. The deposition of calcium carbonate generates a small net sequestration explicitly resulting from individual biocalcification given that the precipitation of 1 mol of CaCO<sub>3</sub> releases approximately 0.6 mol of CO<sub>2</sub> (Ware et al. 1992). But this net sequestration  $(1.0-0.6 = 0.4 \text{ mol of CO}_2 \text{ per mol of CaCO}_3)$  is not enough to compensate the CO<sub>2</sub> that is released due to the catabolism of organic matter. Nevertheless, scaling these numbers up from the individual to the ecosystem level is not a trivial task. In a controversial paper, Tang et al. (2011) proposed that bivalve (and seaweed) aquaculture could increase atmospheric CO<sub>2</sub> absorption within coastal ecosystems. These authors did not account for the release of CO<sub>2</sub> via

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	Sequestration	Biocalcification	Respiration	Balance	Balance/	
Species (Habitat)	gC m <sup>-2</sup> year <sup>-1</sup>	Sequestration	References			
Potamocorbula amurensis	23.9 <sup>a</sup>	18.0	37.0	-31.1	-1.30	Chauvaud et al. (2003)
Mytilus edulis (sheltered)	3.8	2.3 <sup>a</sup>	1.9	-0.4	-0.09	Hily et al. (2013)
Mytilus edulis (semiexposed)	129.2	77.4 <sup>a</sup>	44.3	7.6	0.06	Hily et al. (2013)
Mytilus edulis (exposed)	45.0	27.0 <sup>a</sup>	19.6	-1.6	-0.03	Hily et al. (2013)
Crassotrea gigas (sheltered)	286.8	172.0 а	11.9	103.0	0.36	Hily et al. (2013)
Chlamys farreri	78.1	54.0	71.7	-47.6	-0.61	Jiang et al. (2014)
Crassostrea gigas	15.5 <sup>a</sup>	11.1	32.7	-28.3	-1.83	Lejart et al. (2012)
Ruditapes philippinarum	98.2	66.7	272.4	-241.0	-2.45	Mistri and Munari (2012)
Arculata senhousia	46.0	11.7	50.4	-16.1	-0.35	Mistri and Munari (2013)
Mytilus galloprovincialis	1639.2	1041.6	2253.6	-1656.0	-1.01	Munari et al. (2013)

Table 12.1 Carbon fluxes in different bivalve species: sequestration (carbon content in the shell), biocalcification (carbon released during biogenic calcification), respiration (carbon released through respiration of organic matter), balance (sequestration minus biocalcification and respiration), ratio balance/sequestration, and bibliographic references

<sup>a</sup>shell dissolution is included in this term

respiration in their budget (see Mistri and Munari 2013; Munari et al. 2013) but they argued for the inclusion of some relevant ecosystem effects when scaling up from the individual to the ecosystem level. For example, Tang et al. (2011) suggested that in a strongly autotrophic system, CO<sub>2</sub> released by carbonate precipitation may be used by photosynthetic organisms, resulting in a lower transfer of CO<sub>2</sub> from water to the atmosphere. They also suggested that removing shells from the oceans presents a long-term carbon sink. The slow dissolution of shells in the oceans, e.g. ~29 years for a 4-year old oyster excluding abrasion effects from waves and dissolution after burial (Suykens et al. 2011), provides a buffering capacity of respiratory acids to the environment (Waldbusser et al. 2013). Consequently, this removal can cause a loss of alkalinity regeneration and buffering of metabolic acids, which could affect ecosystem functioning (Waldbusser et al. 2013). These effects on water chemistry highlight that a simple multiplicative extrapolation from the individual to the ecosystem level oversimplifies the role of bivalves in the ecosystem. As stated by Lejart et al. (2012), the contribution of C. gigas to total carbon fluxes should be estimated for the entire community and not just for oysters. In addition, as stated by Waldbusser et al. (2013), the final destination of the shells can be relevant for ecosystem functioning and consequently has a feedback on the bivalves themselves. Clearly an integrated approach is required in which the ecosystem as well as anthropogenic aspects are simultaneously considered.

## **12.3** The Influence of Organic Carbon on CO<sub>2</sub> Fluxes

The strong coupling between inorganic and organic carbon cycles is fundamental for scaling up from individual to population fluxes. This is even more critical in aquaculture sites, where bivalve populations are artificially maintained at generally high densities. The ecosystem role, and implications on the  $CO_2$  cycle, of dense bivalve populations can be very complex due to cascading effects, e.g. indirect effects on fish species via zooplankton consumption (Gibbs 2007; Kluger et al. 2017). Only the direct bivalve ecophysiological processes will be discussed in this chapter. The five main, direct ecophysiological processes of bivalves within the carbon cycle are: (1) respiration, which implies a net release of  $CO_2$  (discussed above); (2) biocalcification, which involves a net sequestration of carbon (discussed above); (3) food ingestion; (4) rejection of uningested food; and (5) egestion of unabsorbed food. In addition, an indirect link with the carbon cycle is carried out by excreted nutrients (Fig. 12.1). Although ingestion, rejection, egestion, and excretion are not directly involved in the inorganic carbon cycle, they are key processes for phytoplankton dynamics, which in turn play a key role in the  $CO_2$  cycle.

Bivalve <u>ingestion</u> may cause a direct top-down control on zooplankton (Maar et al. 2008) and phytoplankton populations (Dame 1996; Dame and Prins 1998; Newell 2004; Petersen et al. 2008; Huang et al. 2008). The net effects are strongly dependent on bivalve biomass and its relation to local environmental conditions, mainly water residence time and phytoplankton production rates (Dame and Prins



Fig. 12.1 Ecosystem approach to carbon cycling (continuous and dashed lines for organic and inorganic carbon, respectively) and feedbacks of mussel aquaculture on the pool of inorganic nutrients (dotted line). (Adapted for C from Cranford et al. 2012)

1998), which represent the renewal of planktonic resources driven by allochthones and autochthonous processes, respectively. If filtration capacity dominates over renewal, planktonic communities could be negatively affected (e.g. Heral 1993; Prins et al. 1998; Maar et al. 2007, 2010). This effect on planktonic biomass could have a direct effect on CO<sub>2</sub> dynamics although secondary local drivers could also exert a significant influence on the net fluxes. For example, in nutrient-limited systems, the reduced phytoplankton population could accelerate its turnover rate by using the additional available nutrients, which in turn could result in the same levels of CO<sub>2</sub> fixation as for a larger population (Newell 2004). Contrarily, in light-limited systems, the increase in filtration pressure usually causes a decrease in phytoplankton biomass and primary production (Cloern et al. 2007; Smayda 2008). This effect can be relaxed if filtration activity is sufficient to increase water clarity and light penetration (Cerco and Noel 2007; Schröder et al. 2014), which could stimulate phytoplankton growth and consequently CO<sub>2</sub> fixation. In addition to the changes in biomass, the structure of phytoplankton communities could also be affected due to the increasing retention efficiency from small to large particles (Jacobs et al. 2015; Cranford et al. 2016). This differential retention efficiency may benefit the relative abundance of the smallest planktonic species (e.g. Vaquer et al. 1996; Smaal et al. 2013; Froján et al. 2014); however, this is a site-specific effect, as demonstrated by Sonier et al. (2016), who could not find any changes in the ratio picoplankton:nanoplankton in a densely cultured site in Atlantic Canada. In any case, the potential alteration of a phytoplankton community could have an effect on CO<sub>2</sub> fluxes.

During the feeding process, phytoplankton and particulate organic matter are consolidated into pseudofaeces (rejected uningested material) and faeces (egested unabsorbed material). These biodeposits sink to the seafloor and their fates are highly dependent upon local environmental conditions (Carlsson et al. 2009, 2010; Jansen 2012). The hydrodynamic regime is relevant not only for determining the horizontal advection of the biodeposits (Pearson and Black 2001; Grant et al. 2005), but also for their potential disaggregation (Driscoll 1970). The remineralization of the biodeposits begins in the water column and consequently the amount of organic matter that reaches the bottom is dependent on water depth. This vertical flux is critical for pelagic-benthic coupling and consequently for CO<sub>2</sub> dynamics. For example, in shallow systems, biodeposits accumulated on the seafloor are exposed to very dynamic conditions in which resuspension and mixing can play important roles in determining remineralization rates or organic matter (Findlay and Watling 1997). In contrast, in deep fjord-type systems, sedimentation of biodeposits could transfer carbon to deep waters, potentially reaching the sediment (Sepúlveda et al. 2005), which can be considered as a carbon sequestering compartment. In addition to hydrodynamics and depth, other local conditions such as grain size, temperature, dissolved oxygen, presence/absence of seagrass, infauna, etc. determine the assimilative capacity of the benthos (Kusuki 1981; Souchu et al. 2001; Mitchell 2006). These local processes, in conjunction with bay-scale aspects such as terrestrial organic inputs and stoichiometry of nutrient inputs, define bay-scale dynamics and ultimately ocean-atmosphere CO<sub>2</sub> fluxes (Laruelle et al. 2010; Bauer et al. 2013).

The remineralization of biodeposits on the seafloor enhances the fluxes of nutrients under highly dense bivalve populations (e.g. Carlsson et al. 2009; Alonso-Pérez et al. 2010). In addition, bivalve ammonia <u>excretion</u> constitutes another source of nitrogen that can be directly used by phytoplankton (e.g. Smaal and Prins 1993; Sara 2007). Nitrogen is probably the most limiting nutrient in coastal marine ecosystems in the temperate zone (Howarth and Marino 2006). Therefore, in nutrientlimited systems, bivalve ammonia excretion can enhance primary production (Smaal 1991; Prins et al. 1995, 1998; Pietros and Rice 2003). This bottom-up control on phytoplankton populations has been demonstrated for aquaculture sites emplaced in nutrient-limited systems such as in Grande-Entrée Lagoon (Canada, Trottet et al. 2008) or Narragansett Bay (USA, Oviatt et al. 2002). Bottom-up control effectively accelerates phytoplankton turnover and primary production rates, which directly increase the net  $CO_2$  fixation via photosynthesis, thereby accelerating carbon assimilation into the biosphere.

# 12.4 Ecosystem Services of Non–Harvested and Harvested Populations

The chemical and ecological aspects discussed above can be directly applied to both non-harvested and harvested populations; however, the final destination of the bivalve is a critical aspect that needs to be considered when valuing ecosystem services. For example, as stated above, the final destination of the shells can be relevant for water chemistry and consequently for ecosystem functioning (Waldbusser et al. 2013). In the case of non-harvested populations, the shells remain in the ocean but the final destinations of harvested bivalves are diverse, from waste to building materials (e.g. Rodríguez Álvaro et al. 2014; Varhen et al. 2017), agricultural usage or the production of lime (calcium oxide CaO), which could be used to remove phosphates from rural watersheds (Abevnaike et al. 2011). This difference is fundamental for the shell, but it is even more critical when the meat of the bivalve is part of the equation. In the case of non-harvested bivalves, the tissue will become part of the food web via predation and decomposition after death. Food provision via the meat of harvested bivalves is the primary goal of culturing bivalves. Although these differences are meaningless when discussing the role of bivalves as a whole in the  $CO_2$  cycle, they become very important when valuing ecosystem services. Therefore, in the case of non-harvested bivalves that are not harvested to provide food, the analysis of their role on the CO<sub>2</sub> cycle should only include the chemical and ecological aspects discussed above. In the case of harvested bivalves, however, a clear distinction between the tissue, which is the main product of this economic activity, and the shell, which usually is considered waste, can be made when valuing their ecosystem services.

In the most extreme scenario, it can be argued that the shell has no marketable value and should be considered waste. In that situation, the carbon sequestered in the shell could be used to valorize the waste and create a by-product for carbon sequestration. Consequently, in that scenario all the CO<sub>2</sub> released from biocalcification and respiration should be accounted towards the CO<sub>2</sub> budget of the product, the meat. This would result in valuing the waste (shell) as a by-product that constitutes a net sink of carbon independent of the CO<sub>2</sub> released during the biocalcification and respiration. An alternative, and probably more logical, accountability would be to split the  $CO_2$  fluxes towards shell and meat as a function of the biological processes involved in their formation. This implies splitting all the ecosystem fluxes and respiration among shell and meat as a function of their energetic demand. Splitting the energetic demand of a bivalve between shell and meat is not straightforward. It is commonly accepted that most of the energy is allocated towards maintenance, tissue growth and reproduction rather than shell growth. Nevertheless, the exact fraction of total energy that is invested in shell growth is unknown in part because any estimation is highly dependent on local environmental conditions. For example, habitat (Fig. 12.2, Rodhouse et al. 1984), feeding conditions (Aldrich and Crowley 1986), hydrodynamics (Steffani and Branch 2003) and predation pressure (Lowen et al. 2013) can all affect the energy allocation towards shell.

The lack of specific studies on energy allocation and the effects of local conditions on growth investment becomes a serious limitation when trying to split carbon fluxes between shell and tissue. The available data are limited to the estimations by Hawkins and Bayne (1992) who suggested that *Mytilus edulis* could spend more than 20% of the energy that is available for growth on shell formation. This matches with the calculations of Duarte et al. (2010), who indirectly estimated that *Mytilus galloprovincialis* could invest 20–28% of the energy that is available for growth in



Fig. 12.2 Allocation of carbon in wild and cultured Mytilus edulis. (From Rodhouse et al. 1984)

shell formation. It is important to highlight that these estimations establish the energy that is available for growth as a bottom line for the calculations, in other words, the available energy after paying maintenance, digestion/absorption and growth costs (Scope For Growth, Winberg 1960). The shell does not require any maintenance costs, with the exception of repairing mechanical damage, and consequently allocating 20–28% (based on Duarte et al. 2010) of the total  $CO_2$  fluxes towards shell would overestimate the energetic requirements of shell growth. Accordingly, for the following estimated calculations, 10% has been assumed as the percentage of the total energetic demands that is allocated towards shell (with the remaining 90% allocated towards maintenance and tissue growth).

As explained above, all the processes in the full ecosystem approach towards the quantification of CO<sub>2</sub> fluxes should be split between tissue and shell according to this 10/90% estimation. For these preliminary calculations and for simplicity, the following calculations have included only biocalcification and respiration in the CO<sub>2</sub> budget, following the approach presented in Table 12.1. Accordingly, the respiration values provided in Table 12.1 have been re-calculated in Table 12.2 by considering only 10% of the total respiration, which would represent the CO<sub>2</sub> flux that corresponds to the shell energetic requirements. The datasets from Hily et al. (2013) have been removed from this table due to the uncertainties highlighted above. Splitting respiration provides a general budget for shell CO<sub>2</sub> fluxes (Table 12.2) rather than for the whole individual (Table 12.1). According to the Table 12.2 calculations and in the context of harvested bivalves, the shells, which are waste of an industrial process, could be considered net sinks of  $CO_2$  and consequently valorized as by-products. It should be re-emphasized that this reasoning is based on the assumption that humans culture bivalves with the aim of producing food and not sequestering  $CO_2$  and consequently, from the perspective of ecosystem services the CO<sub>2</sub> generated through respiration should be split between meat and shell.

The next logical question is: is this sequestered carbon relevant from a global perspective? As stated above, cultured bivalves produce  $\sim 7 \times 10^6$  tons of shell per year. Taking into account that 95% is calcium carbonate, and 12% of that is carbon, shells contain  $8 \times 10^5$  tons of carbon per year. Assuming that shell growth demands 10% of total energy and the net sequestration of carbon in the shell is ~21% (aver-

balance/sequestration, and bibl	liographic referenc	es				
	Sequestration	Biocalcification	Respiration	Balance	Balance/	
Species	$g C m^{-2} y ear^{-1}$	g C m <sup>-2</sup> year <sup>-1</sup>	g C m <sup>-2</sup> year <sup>-1</sup>	g C m <sup>-2</sup> year <sup>-1</sup>	Sequestration	References
Potamocorbula amurensis	23.9	18.0	3.7	2.2	0.09	Chauvaud et al. (2003)
Chlamys farreri	78.1	54.0	7.2	16.9	0.22	Jiang et al. (2014)
Crassostrea gigas	15.5	11.1	3.3	1.1	0.07	Lejart et al. (2012)
Ruditapes philippinarum	98.2	66.7	27.2	4.2	0.04	Mistri and Munari (2012)
Arculata senhousia	46.0	11.7	5.0	29.3	0.64	Mistri and Munari (2013)
Mytilus galloprovincialis	1639.2	1041.6	225.4	372.2	0.23	Munari et al. (2013)

 

 Table 12.2
 Carbon fluxes for shells in different bivalve species: sequestration (carbon content in the shell), biocalcification (carbon released during biogenic calcification), respiration (10% of carbon released through respiration of organic matter), balance (sequestration minus biocalcification and respiration), ratio

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age value of Balance/Sequestration column in Table 12.2),  $1.71 \times 10^5$  tons of carbon, or  $6.3 \times 10^5$  tons of CO<sub>2</sub> equivalent, per year are effectively sequestered in shells of cultured bivalves. In economic terms, the impact highly depends upon the carbon initiative that values a ton of CO<sub>2</sub>, which can range from US\$131 in the Swedish carbon tax, to US\$1 in Mexico (World Bank et al. 2016). Assuming an average value of US\$24 per ton of CO<sub>2</sub> (average for Denmark, France, United Kingdom, British Columbia and Ireland; World Bank et al. 2016) the global value of the carbon effectively sequestered in shells of cultured bivalves is ~15.7 million US\$ per year. This amount represents less than 0.01% of the total bivalve aquaculture value.

# 12.5 Case-Study: Norwegian Cultured Mussels

Marine carbon burial is the main natural mechanism of long-term organic carbon sequestration (Berner 1982; Hedges et al. 1997). Fjords are deep, glacially carved estuaries situated at high latitudes. Smith et al. (2015) estimated that  $18 \times 10^6$  tons carbon is buried in fjord sediment each year. This is equivalent to 11% of the annual marine carbon burial globally, and makes the fjord organic carbon burial rate 100 times more efficient than the global ocean average, per unit area. As stated above, local conditions are critical for the implications of cultured bivalves on the CO<sub>2</sub> cycle. The estimation of the CO<sub>2</sub> budget of mussel (*Mytilus edulis*) farming in a Norwegian fjord has been selected as a case-study to guide the application of the rationale described in this study to a cultured system. It is important to emphasize that due to the effects of the cultured species and local conditions, the following calculations cannot be extrapolated to other bivalves or locations.

The CO<sub>2</sub> budget is based on the life history of a 2-year old mussel, the typical lifespan of mussels in suspended culture in Norway. The mussel is harvested before reproduction, and obtains a dry shell weight (DSW) of 4.8 g and a dry tissue weight of 1.0 g. It is assumed that 85% of farmed mussels will be harvested and 15% will fall off their ropes during strong winds and wave action, farm operation, density control/thinning, harvest and predation (Strohmeier et al. 2008). As a consequence of low food quantity and high food quality, mussels have not been reported to reject uningested food in Norwegian fjords (Strohmeier et al. 2015). Following the rationale described above, the CO<sub>2</sub> fluxes were split between the shell and the tissue according to their presumed energetic demand.

#### 12.5.1 Respiration

Throughout their life history, mussels consume oxygen and release  $CO_2$  as a result of the catabolism of organic matter. The oxygen required for the mussel growth has been estimated by allometric scaling (Bayne and Widdows 1978; Thompson 1984;



Smaal et al. 1997) and a Dynamic Energy Budget (DEB) model parameterized for Norwegian mussels (Rosland et al. 2009), using a seasonal time series of mussel growth and ecophysiology that included respiration data (Strohmeier 2009; Strohmeier et al. 2015). The results indicated a cumulative oxygen consumption of 4.5 to 8.8 g. Assuming a respiratory quotient towards herbivory (0.85, Galtsoff 1964), this results in a cumulative release of 0.12 to 0.23 mol or 5.3 to 10.3 g CO<sub>2</sub>. Splitting the CO<sub>2</sub> fluxes between shell and tissue according to the 10/90% outlined above, the allocated catabolism of the shell represents 0.5 to 1 g CO<sub>2</sub> (mean 0.75, Fig. 12.3) and the catabolism of the tissue from 4.8 to 9.3 g CO<sub>2</sub> (mean 7.05, Fig. 12.3).

#### 12.5.2 The Shell

The deposit of CaCO<sub>3</sub> in a 4.8 g mussel shell sequesters 0.55 g carbon or 2.0 g CO<sub>2</sub>. The flux of CO<sub>2</sub> due to shell formation to land (harvest) and seabed (fall off) is 1.7 and 0.3 g CO<sub>2</sub>, respectively (Fig. 12.3). The amount of CO<sub>2</sub> released during biocalcification for the same individual is 1.2 g. Therefore, the net sequestration in the shell is 0.8 g CO<sub>2</sub>. Including the associated cost of respiration (10%) to the net sequestration of CO<sub>2</sub> in the shell results in a balance (sequestration minus biocalcification and respiration) in the range - 0.2 to 0.3 g CO<sub>2</sub>, which accounts for all the relevant fluxes at the individual level needed to define the CO<sub>2</sub> budget (e.g. Table 12.1). Under the assumption that 85% of the mussels are harvested and 15% fall off, the mean balance indicates a net flux of 0.04 g CO<sub>2</sub> to land (harvest) and 0.01 g CO<sub>2</sub> to the seabed (fall off).

## 12.5.3 The Tissue

The carbon content of the tissue shows seasonal variation, with a mean value of 0.44 g C per gram of dry weight (range 0.40–0.47, Jansen 2012). The mean carbon content of a 1 g of mussel's tissue in terms of dry weight is thereby 0.44 g, corresponding to 1.61 g CO<sub>2</sub>. The flux of CO<sub>2</sub> to land (harvest) and seabed (fall off) is thus 1.36 and 0.24 g CO<sub>2</sub>, respectively (Fig. 12.3). Inclusion of the associated cost of catabolism (90%) results in a net balance (sequestration minus catabolism) in the range from -7.7 to -3.2 g CO<sub>2</sub>. The mean balance indicates a net flux of -4.9 g CO<sub>2</sub> to land (harvest) and -0.5 g CO<sub>2</sub> to seabed (fall off).

# 12.5.4 Egestion of Unabsorbed Food

The cumulative mass and carbon content of fecal pellets has been estimated based on Jansen et al. (2012) and the DEB model (Rosland et al. 2009). The results indicate that a mussel egests 12.9–13.7 g faeces over the 2 year period in terms of dry weight. The fecal pellets comprise a C fraction of 13.5% (Jansen 2012). The cumulative egestion is thereby 1.7–1.9 g C or 6.4–7.0 g CO<sub>2</sub>. Faeces contain fresh biological material, and may be used as a food source by other organisms until they are buried in the sediment. Faeces were assumed to enter the pelagic environment after being "trapped" on the mussel collectors for a brief period of time, then they sink towards the seabed (Jansen et al. 2012). The sinking velocity of fecal pellets, obtained for mussels grazing on natural seston, has been reported at 3.9 mm s<sup>-1</sup> or 337 m d<sup>-1</sup> (Carlsson et al. 2010). Overall, the residence time for faeces in the pelagic environment was set to two days, representing the average depth of a Norwegian fjord of about 300 m.

Faeces contain a labile faction that can be fully catabolized in oxygenated water on a timescale ranging from 5 to 15 days, depending on the season (Jansen 2012). Here a constant decay and 10 days to fully catabolize the fecal matter is assumed. Taking into account the average sinking time of 2 days, 80% of the faeces will reach the seabed (mean 1.4 g C or 5.4 g CO<sub>2</sub>), while 20% will be metabolized in the pelagic environment (0.4 g C or 1.3 g CO<sub>2</sub>). Splitting these fluxes according to the associated energy demand of shell (10%) and tissue (90%), 0.1 and 1.2 g CO<sub>2</sub> of the fecal matter will be respired in the pelagic environment, and 0.5 and 4.9 g CO<sub>2</sub> will reach the sediment for shell and tissue, respectively (Fig. 12.3). In deep anoxic fjords a high carbon burial rate can be expected, and in this budget it is assumed that the carbon that reaches the seabed is not metabolized further.

### 12.5.5 General Budget in the Context of Ecosystem Services

The balance for the shell and tissue was estimated separately as: +burial of fall off mussel +burial of faeces +harvest of mussel –biocalcification –respiration –faeces respired in the water column. The balance for shell is +0.45 g CO<sub>2</sub> (+0.3 + 0.5 + 1.7–1.2 –0.75 –0.1) suggesting that mussel shell of cultured mussels in a 2 year cycle in a Norwegian fjord can be considered a net sink of CO<sub>2</sub>, assuming that the harvested shells are disposed of in a way that can be considered sequestered material, e.g. concrete. The balance for tissue is -3.11 g CO<sub>2</sub> (+0.24 + 4.9 + 0–0 –7.05 –1.2). Note that for tissue, CO<sub>2</sub> towards the term 'harvest of mussel' has not been included in the budget. This flow of CO<sub>2</sub> is assumed to be consumed and respired in the short term and consequently not sequestered in the long term. Accounting shell and tissue together, this budget confirms that mussels are, as expected, net sources of CO<sub>2</sub>.

This budget includes the traditional fluxes of respiration and biocalcification (e.g. Table 12.1), but also an additional direct ecosystem flux, the egestion of unabsorbed food. Given that in Norwegian waters the rejection of uningested food is negligible, the impact of bivalve ingestion and ammonia excretion would be the only two additional processes to assess for a holistic ecosystem approach to the  $CO_2$  budget. The  $CO_2$  fluxes have been split according to the biological process involved in the formation of the shell and tissue, based on their anticipated energetic demands (see text above). In valuing the ecosystem service of mussel farming in the carbon cycle a distinction has been made between the shell (waste) and the tissue (food). Following this rationale, the goods and services of mussel farming in deep fjords includes the valorization of the shells as a net sink of  $CO_2$ .

The more holistic ecological approach reveals a previously unaccounted for, yet significant indirect carbon sequestration by deposition of mussel faeces in sediment. Given the assumption that all mussel faeces are buried in deep fjords, the sediment may sequester more than 60% of the total  $CO_2$  respired. In environments comprising high food quantities, mussels can produce a significant amount of pseudofaeces (rejected uningested material) in addition to fecal matter and thereby increase the organic flux to the seabed (Galimany et al. 2013). If this particulate matter, faeces and pseudofaeces, sinks into an environment where it is not further catabolized, then a net  $CO_2$  sequestering from mussel farming is plausible. This may serve as an example to encourage an ecosystem approach towards the quantification of bivalve  $CO_2$  fluxes.

A typical Norwegian mussel farm produces a volume from 50 to 150 tons each year equating to a mean farm production of about 6.25  $\times 10^6$  mussels. Valuing the ecosystem service of mussel farming in the carbon cycle, the shells sequester 2.8 tons of CO<sub>2</sub>, or 146 US\$ per year (assuming US\$53 per ton of CO<sub>2</sub> in Norway according to World Bank et al. (2016)).

## 12.6 Conclusions

As expected and as proved in the literature (Table 12.1), given their nature as primary consumers, bivalves release  $CO_2$ . The sequestration of  $CO_2$  in the shell is not enough to compensate the release generated during the respiration of organic matter. Note that the use of the term "production" was avoided within the manuscript in the context of bivalves "producing" CO<sub>2</sub>. This has been done intentionally to avoid negative connotations associated with being a CO<sub>2</sub> generator. As discussed, all primary consumers release  $CO_2$  that was captured by primary producers. Accordingly, a better term could be "recycling" CO<sub>2</sub> rather than a term that suggests the production of new CO2. In any case, in the context of ecosystem services, there are two fundamental aspects that should be also taken into account when estimating a CO<sub>2</sub> budget: the consideration of ecosystem processes when scaling individual fluxes to the population level (e.g. Lejart et al. 2012), and the final destination of the bivalves (e.g. Waldbusser et al. 2013), that is, bivalves harvested for food production or nonharvested bivalves. Ecosystem processes involving bivalves are relevant and alter the CO<sub>2</sub> cycle via filtration and/or nutrient cycling (Lejart et al. 2012). Consequently, they should be considered when the CO<sub>2</sub> budget is calculated for bivalve populations. When valuing ecosystem services, it has been recognized that humans harvest bivalves to provide food and consequently shells should be considered waste. Accordingly, a different CO<sub>2</sub> budget should be calculated for product (tissue) and waste (shell).

Under these considerations, bivalve shells can be considered net sinks of  $CO_2$ and consequently provide additional ecosystem services besides the food provided by the tissue. A full life cycle analysis should be performed to account for the emissions required to properly dispose of the shells. The 0.45 g  $CO_2$  sequestered by the shell of each cultured mussel in Norway is hardly significant taking into account that a regular car produces more than 100 g  $CO_2$  per km. For example, since 2015 European Union law requires that new cars do not emit more than an average of 130 g  $CO_2$  per km, with a target of 95 g  $CO_2$  per km by 2021 (European Commission, Climate Action). Even when these numbers are extrapolated to the global scale, a conservative extrapolation of the individual bivalve budget to the global production would result in a sequestration of  $6.3 \times 10^5$  tons of  $CO_2$  per year, ~15.7 × 10<sup>6</sup> US\$/ year. In different units, this is equivalent to the annual emissions of 242,307 cars driving an average of 20,000 km each. Although this is far from solving a global problem, everything counts. In addition, it is important to re-emphasize that this comes at no cost or effort given that bivalves are cultured to produce food.

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