



## Inorganic carbon and nutrient dynamics in the marginal ice zone of the Barents Sea: Seasonality and implications for ocean acidification

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### ABSTRACT

The Barents Sea is a productive Arctic shelf sea experiencing Atlantification with ocean warming, sea ice loss and increased influence of Atlantic Water. The impact of these changes on inorganic carbon and nutrient dynamics and ocean acidification are yet to be fully understood. Seasonal variability and drivers of inorganic carbon and nutrients were determined from south of the Polar Front at 76 °N in the Barents Sea to 82 °N in the Nansen Basin, Arctic Ocean, encompassing Atlantic and Arctic regimes in summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (May 2021). Summer sea-ice meltwater was the main driver reducing total alkalinity ( $A_T$ ) and dissolved inorganic carbon ( $C_T$ ), corresponding to 55 % and 81 % of the total changes  $\Delta C_T$  and  $\Delta A_T$ , respectively, in the surface layer. Primary production reduced  $C_T$  (37 % of  $\Delta C_T$ ) and nutrients, particularly in ice-free waters, increasing calcium carbonate ( $\text{CaCO}_3$ ) saturation for aragonite ( $\Omega_{\text{aragonite}}$ ) from 1.3 in spring to 2.3 in summer. Net community production (NCP) in the upper 50 m was highest ( $1.3 \text{ mol C m}^{-2}$ ) in the Atlantic Water regimes and lowest ( $0.4 \text{ mol C m}^{-2}$ ) in the ice-covered Nansen Basin in summer. Mixing, organic matter remineralisation and CO<sub>2</sub> uptake enriched  $A_T$ ,  $C_T$  and nutrients to pre-condition the water column from winter to early spring. Formation and dissolution of  $\text{CaCO}_3$  from shells and ikaite in sea ice (5–13 % of  $\Delta A_T$ ) represented a minor  $A_T$  source from winter to spring. Highest NCP by spring ( $0.3 \text{ mol C m}^{-2}$ ) occurred in a transient Atlantic-like regime over the shelf slope. Atmospheric CO<sub>2</sub> uptake contributed up to 37 % of  $\Delta C_T$  and the region was an annual CO<sub>2</sub> sink. Sea-ice processes and deep winter convection lowered  $A_T$  and  $\Omega_{\text{aragonite}}$  (1.12–1.14) to enhance the risk of acidification over the shelf. Future Atlantification may increase biological production, reduce meltwater dilution effects and counteract acidification in the Barents Sea. The observations reveal the importance of seasonal and spatial studies to capture variability in inorganic carbon and nutrient cycling, driven by numerous processes coupled to the ice-ocean system that may exacerbate or alleviate ocean acidification in Arctic shelf seas.

### 1. Introduction

The Arctic Ocean and the seasonally ice-covered shelf seas are particularly vulnerable to ocean acidification as the solubility of carbon dioxide (CO<sub>2</sub>) increases in cold water and fresher polar waters have lower total alkalinity ( $A_T$ ), which is the natural buffer against acidity (Zeebe and Wolf-Gladrow, 2001). Ocean acidification refers to the decrease in carbonate ions in seawater that lowers the degree of saturation ( $\Omega$ ) of calcium carbonate ( $\text{CaCO}_3$ ) minerals (Orr et al., 2005; Doney et al., 2009). If  $\Omega$  becomes less than 1 this is an indication of vulnerability of the  $\text{CaCO}_3$  shells to dissolution (Orr et al., 2005; Doney

et al., 2009). Ocean acidification in the Arctic is driven by uptake of atmospheric CO<sub>2</sub> that increases the oceanic dissolved inorganic carbon ( $C_T$ ) and freshwater inputs from melting sea ice, glacial meltwater, precipitation and river runoff (Chierici and Fransson, 2009; Fransson et al., 2013). The freshwater-driven dilution in  $A_T$  and subsequent reduction in surface water  $\Omega$  occurs possibly faster than reductions in  $\Omega$  due to anthropogenic CO<sub>2</sub> uptake (Chierici and Fransson, 2009; Yamamoto-Kawai et al., 2009; Zhang et al., 2020).

Variability of oceanic  $C_T$  and  $A_T$  is driven by seasonal physical, biogeochemical and biological processes that influence the state of ocean acidification (indicated by changing  $\Omega$ ). Changes in salinity from

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freshwater inputs, advection and mixing of different water masses reduce  $C_T$  and  $A_T$  through dilution and increase  $C_T$  and  $A_T$  through concentration (Friis et al., 2003), with competing effects on  $\Omega$  (Chierici and Fransson, 2009; Yamamoto-Kawai et al., 2009; Zhang et al., 2020; Jones et al., 2021). Primary production and biological uptake of  $C_T$  is accompanied by small increases in  $A_T$ , which increase  $\Omega$ , as concomitant uptake of nitrate with  $H^+$  conserves electroneutrality (Brewer and Goldman, 1976; Zeebe and Wolf-Gladrow, 2001). Oppositely, organic matter respiration/remineralisation increases  $C_T$  and slightly decreases  $A_T$ , with the effect of lowering  $\Omega$ . Wind-induced mixing with subsurface high- $CO_2$  waters can lead to reductions in surface water  $\Omega$  in the absence of biological  $CO_2$  uptake (Chierici et al., 2011; Fransson et al., 2017; Jones et al., 2021). Air-sea  $CO_2$  fluxes influence oceanic  $C_T$  only. The seasonal retreat of the sea ice exposes surface waters to the atmosphere and enhances air-sea  $CO_2$  fluxes, which can increase  $\Omega$  through  $CO_2$  release or decrease  $\Omega$  through  $CO_2$  uptake (Fransson et al., 2013). Dissolution of  $CaCO_3$  produces two bicarbonate ions and consumes  $CO_2$ , thereby increasing  $A_T$  by a factor of two relative to the increase in  $C_T$  (Zeebe and Wolf-Gladrow, 2001). Formation of  $CaCO_3$  affects  $A_T$  and  $C_T$  in the opposite direction by the same 2:1 ratio. In addition to biotic  $CaCO_3$  calcite and aragonite, abiotic  $CaCO_3$  ikaite ( $CaCO_3 \cdot 6H_2O$ ) can be found in sea ice under specific thermo-haline conditions (Dieckmann et al., 2010). During sea ice formation, ikaite precipitation can occur in the brine and the minerals can be trapped within the sea ice as the brine is rejected to the underlying seawater. This process partitions  $A_T$  (as  $CaCO_3$ ) in the ice and  $C_T$  (as dissolved  $CO_2$  from the ikaite precipitation reaction) in the brine. During sea ice melt, ikaite is released and dissolves in the meltwater to provide a minor source of  $A_T$  and increasing  $\Omega$  in surface waters influenced by seasonal sea-ice melt (Rysgaard et al., 2007; Fransson et al., 2013; 2017).

The Barents Sea is an Atlantic Water inflow shelf to the Arctic Ocean, where the advection of Atlantic-sourced waters supplies the region with heat, salt,  $C_T$  and nutrients (Reigstad et al., 2002; Carmack et al., 2006; Chierici et al., 2019). Atlantic Water also has a high  $A_T$  content and has been reported to act as a buffer to acidification through entrainment into the surface layer (Chierici et al., 2011; Jones et al., 2021). The southern part of the Barents Sea is characterised by Atlantic regimes due to inflowing Atlantic Water and is seasonally ice free (Loeng, 1991; Reigstad et al., 2002) where the Polar Front marks the transition between Atlantic and Arctic regimes (Oziel et al., 2016). The northern part of the Barents Sea is influenced by the Atlantic Water inflow in the West Spitsbergen Current along the northern shelf and the cold and low-saline Polar Water and locally formed and advected sea ice from the Arctic Ocean (Smedsrud et al., 2013; Renner et al., 2018; Lundegaard et al., 2022). Arctic waters consist of remnants of winter cooling, convection and brine from sea ice that form a halocline separating a lower salinity surface layer affected by sea-ice melt from subsurface Atlantic Water (Onarheim et al., 2014; Lundegaard et al., 2022). Sea-ice and ice-ocean processes contribute to dense Barents Sea water formation and provide a mechanism of  $CO_2$  transport across the shelf and outflow into the Arctic Ocean (Skagseth et al., 2020; Rogge et al., 2023). This Barents Sea is an important region for mixing and cooling of the Atlantic Water, which influences the local sea ice cover and the extent of the marginal ice zone (MIZ).

Phytoplankton blooms are seasonal features in the MIZ of the Barents Sea, closely coupled to light availability, sea-ice extent and supply of inorganic nutrients (Reigstad et al., 2002; Tremblay et al., 2015; Assmy et al., 2017). The spring blooms result in strong uptake and seasonal cycling of inorganic nutrients and  $C_T$  through production of organic matter that drives atmospheric  $CO_2$  uptake in surface waters over the Barents Sea shelf (Fransson et al., 2001; Carmack et al., 2006; Henley et al., 2020). Further blooms in summer and autumn can contribute significantly to primary production and carbon cycling if nutrients are resupplied through upwelling, mixing with subsurface waters and remineralisation of organic matter in the presence of sufficient light (Reigstad et al., 2002; Oziel et al., 2016). Nutrient supply from Atlantic-

sourced waters to the surface layer mainly occurs in autumn and winter through vertical mixing, upwelling and tidal forcing, and in spring and summer through episodic wind-induced mixing (Randelhoff et al., 2015; Oziel et al., 2016; Randelhoff et al., 2018; Renner et al., 2018).

The Barents Sea is experiencing a rapid shift in oceanic regime due to amplification of anthropogenic climate change through warming, loss of sea ice and increased influence of Atlantic Water (Onarheim et al., 2014; Polyakov et al., 2017). The inflowing Atlantic Water has warmed in recent decades (Beszczynska-Möller et al., 2012; Skagseth et al., 2020; Smedsrud et al., 2022) and resulted in Atlantification along the inflow pathways with a shift to more Atlantic-like regimes (Árthun et al., 2012; Onarheim et al., 2014; Polyakov et al., 2017; Lind et al., 2018). Atlantification has led to expansions of temperate species northwards and changes in the ecosystem (Oziel et al., 2016; Neukermans et al., 2018; Ingvaldsen et al., 2021) and increased Atlantic Water influence has been shown to drive acidification in subsurface and intermediate waters of the Arctic Ocean (Ericson et al., 2014; Ulfbo et al., 2018). The lowering of  $\Omega$  for biotic  $CaCO_3$  minerals calcite ( $\Omega_{\text{calcite}}$ ) and aragonite ( $\Omega_{\text{aragonite}}$ ) through ocean acidification has important consequences for marine calcifiers (Orr et al., 2005). Calcifying phytoplankton, such as the coccolithophore *Emiliania huxleyi*, and zooplankton, such as the pteropod *Limacina helicina*, are key producers of calcite and aragonite, respectively, and play important roles in biogeochemical cycling and carbon fluxes in the Barents Sea (Hegseth and Sundfjord, 2008; Zamelczyk et al., 2021). Enhanced supply of nutrients with reduced sea-ice cover may increase primary production and biological carbon uptake (Randelhoff et al., 2018; Arrigo and van Dijken, 2015; Tremblay et al., 2015; Chierici et al., 2019) coupled to inputs of  $A_T$  that buffer against acidification (Jones et al., 2021). However, increased sea-ice melt increases stratification and reduces mixing, which may impede nutrient resupply from subsurface waters and limit primary production and  $C_T$  uptake (Slagstad et al., 2015; Fransson et al., 2017; Chierici et al., 2019). A shift in phytoplankton species such as from diatom- to *Phaeocystis* dominated blooms under changing sea-ice conditions is predicted to reduce carbon uptake and have impacts on the Arctic food web (Carmack et al., 2006; Wassmann et al., 2006). Furthermore, reduced  $A_T$  through meltwater dilution and limited mixing will suppress the biologically- and biogeochemically-driven enhancement of  $\Omega$  (Chierici and Fransson, 2009; Yamamoto-Kawai et al., 2009; Chierici et al., 2011; Zhang et al., 2020; Jones et al., 2021).

Climate-driven changes in sea ice cover, ice-ocean processes and Atlantification have implications for biogeochemical cycling through a balance of processes that influence nutrients,  $C_T$  and  $A_T$ , which may act synergistically to enhance or alleviate the risk of ocean acidification in the Barents Sea and Arctic Ocean. However, there is currently limited understanding of the seasonal variability of inorganic carbon and nutrient cycling in the Barents Sea, particularly between Atlantic and Arctic regimes, which is essential to better understand the current and future ocean acidification state in a region experiencing environmental change. The aim of this study is to determine the seasonal dynamics and drivers of inorganic carbon and nutrient cycling and ascertain the implications for ocean acidification in the Barents Sea. This study presents carbonate chemistry and inorganic nutrient data in the full water column from summer (August 2019), winter (December 2019), winter/spring (2021) and spring (2021) across the Barents Sea shelf and slope and into the deep Nansen Basin of the Arctic Ocean. The changes in the upper ocean are linked to seasonal sea-ice cover, ice meltwater inputs, advection and mixing of different water masses, primary production/respiration, calcium carbonate processes and air-sea  $CO_2$  exchange, across Atlantic and Arctic regimes. The results are discussed within context of the state of ocean acidification during the transitions from polar summer to winter and winter to spring in a region of the Barents Sea and Arctic Ocean already experiencing amplification of anthropogenic climate change.

## 2. Methods

### 2.1. The Barents sea seasonal sampling

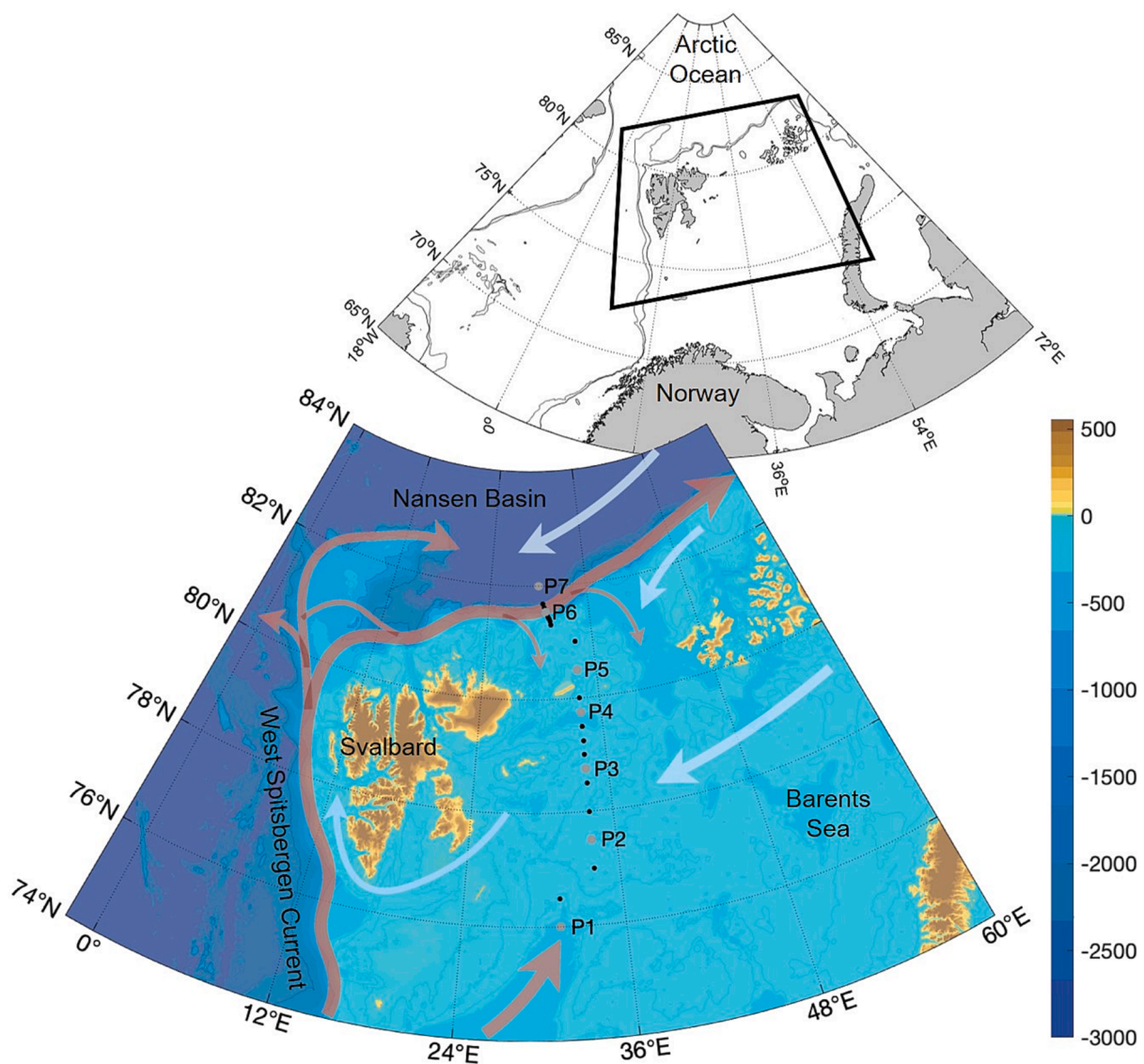
Seasonal sampling was carried out in the northwestern Barents Sea along a latitudinal transect 28.8–34 °E, ~76–82 °N extending from south of the Polar Front, across the shelf and to the Nansen Basin of the Arctic Ocean (Fig. 1). Fieldwork campaigns were conducted onboard RV *Kronprins Haakon* within the framework of the Nansen Legacy project where a series of 25 hydrographic stations that included 7 process (P) stations were occupied (Table 1). The transect encompassed a natural environmental gradient that covered Atlantic and Arctic regimes of the Barents Sea shelf and slope and deep Nansen basin, and the marginal ice zone, which is defined here as the area having a monthly average minimum of 15 % sea ice concentration. Details on sampling procedures can be found in Nansen Legacy (2022).

The surveys were carried out to capture seasonal (quarterly, Q) conditions in the Barents Sea; late summer 8–23 August 2019 (Q3),

winter 3–13 December 2019 (Q4), winter/spring 4–17 March 2021 (Q1), spring 30 April – 13 May 2021 (Q2). Due to the COVID-19 pandemic, the planned successive completion of the seasonal cruises was not possible and therefore Q1 and Q2 were postponed from 2020 to 2021. This introduces additional interannual variability within the seasonal signals, which will be discussed in section 3.7. Therefore, data will be discussed in terms of key patterns during each season and the changes taking place between the consecutive seasonal transitions, i.e. summer to winter 2019 and winter to spring 2021.

### 2.2. Sea-ice conditions

Daily sea ice concentration data derived from AMSR-2 was downloaded from the Institute of Environmental Physics, University of Bremen, Germany (Spren et al., 2008), available at <https://seaice.uni-bremen.de/sea-ice-concentration-amsr-eamsr2/>.



**Fig. 1.** Maps of Atlantic Arctic Ocean region the Barents Sea showing main regional circulation and topographic features. Red arrows depict the Atlantic Water inflow in the West Spitsbergen Current in eastern Fram Strait and the Atlantic Water pathways along the northern Svalbard shelf. Blue arrows depict the general southward flow of Arctic waters and imported sea ice, and the westward flowing Coastal Current around Svalbard. Location of the hydrographic stations are marked (black dots) and the main process (P) stations are labelled (grey dots) along the transect in the Barents Sea. Colour scale is depth (m) with isobars marked at 500, 1000, 1500, 2000, 2500 and 3000 m from bathymetric data retrieved from the International Bathymetric Chart of the Arctic Ocean IBCAO v3 (Jakobsson et al., 2012). Matlab toolbox `m_map` was used. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

**Table 1**

Main hydrographic process (P) stations and sampling dates during summer (August 2019; Q3), winter (December 2019; Q4), winter/spring (March 2021; Q1) and spring (April–May 2021; Q2).

| Station | Location             | Latitude<br>(°N) | Longitude<br>(°E) | Bottom depth<br>(m) | Season/ date |        |               |        |
|---------|----------------------|------------------|-------------------|---------------------|--------------|--------|---------------|--------|
|         |                      |                  |                   |                     | summer       | winter | winter/spring | spring |
| P1      | South of Polar Front | 76.00            | 31.22             | 325                 | 08–08        | 12–12  | 05–03         | 30–04  |
| P2      | Shelf                | 77.50            | 33.99             | 190                 | 12–08        | 10–12  | 07–03         | 02–05  |
| P3      | Shelf                | 78.75            | 34.00             | 305                 | 13–08        | 9–12   | 08–03         | 03–05  |
| P4      | Shelf                | 79.69            | 34.23             | 332                 | 14–08        | 8–12   | 11–03         | 05–05  |
| P5      | Shelf                | 80.50            | 33.99             | 158                 | 16–08        | 6–12   | 12–03         | 07–05  |
| P6      | Slope                | 81.59            | 31.52             | 840                 | 18–08        | 5–12   | 16–03         | 10–05  |
| P7      | Nansen Basin         | 81.93            | 29.14             | 3120                | 21–08        | 2–12   | 17–03         | 13–05  |

### 2.3. Hydrography and water masses

Water column hydrographic measurements were obtained using a conductivity-temperature-depth (CTD) sensor system (Sea-Bird SBE 911 + ) mounted onto a General Oceanics rosette equipped with 24 10 L Niskin bottles. Seawater was typically collected at 12 depths (e.g., 10, 20, 30, 50, 100, 150, 200, then every 100 m to about 10 m above the seafloor) covering the full water column at each hydrographic station (Table 1). The CTD system was deployed through the moonpool of the vessel during Q4, Q1 and Q2 and thus the upper 10 m could not be sampled due to interference by the ship. Therefore, the upper 15 m was selected as the surface reference depth to ensure consistency between the seasons. The CTD data are reported in Gerland (2022), Ludvigsen (2022), Reigstad (2022), Søreide (2022).

Water masses in the Barents Sea and Nansen Basin (Table 2) were classified based on the definitions given in Sundfjord et al. (2020). Atlantic Water is the relatively warm and saline water mass and defined by conservative temperature ( $T_C$ ) > 2 °C and absolute salinity ( $S_A$ ) ≥ 35.06 g kg<sup>-1</sup>. Modified Atlantic Water is a cooler variety of Atlantic Water. Polar Water is the cold and low saline surface and halocline layer.

**Table 2**

Seasonal water mass, inorganic nutrients and carbonate system characteristics using bounds of conservative temperature ( $T_C$ , °C), absolute salinity ( $S_A$ ; g kg<sup>-1</sup>) and potential density ( $\sigma_\theta$ , kg m<sup>-3</sup>) from Sundfjord et al. (2020). Average and standard deviation (in parentheses) are shown for nitrate (NO<sub>3</sub>, μmol kg<sup>-1</sup>); phosphate (PO<sub>4</sub>, μmol kg<sup>-1</sup>); silicic acid (Si(OH)<sub>4</sub>, μmol kg<sup>-1</sup>); dissolved inorganic carbon (C<sub>T</sub>, μmol kg<sup>-1</sup>); total alkalinity (A<sub>T</sub>, μmol kg<sup>-1</sup>); the A<sub>T</sub>:C<sub>T</sub> ratio (in situ values of A<sub>T</sub> relative to C<sub>T</sub>); aragonite saturation state (Ω<sub>aragonite</sub>) in summer (2019, Q3), winter (2019, Q4), winter/spring (2021, Q1) and spring (2021, Q2). No CBSDW was identified in winter/spring and spring 2021 (–).

| Water mass                                | Season (year, cruise) | NO <sub>3</sub>        | PO <sub>4</sub>        | Si(OH) <sub>4</sub>    | A <sub>T</sub>         | C <sub>T</sub>         | A <sub>T</sub> :C <sub>T</sub> | Ω aragonite |
|---|-----------------------|------------------------|------------------------|------------------------|------------------------|------------------------|--------------------------------|-------------|
|   |                       | μ mol kg <sup>-1</sup> | μ mol kg <sup>-1</sup> | μ mol kg <sup>-1</sup> | μ mol kg <sup>-1</sup> | μ mol kg <sup>-1</sup> |                                |             |
| Warm Polar Water (wPW)                    | Summer                | 4.61 (5.17)            | 0.37 (0.31)            | 2.29 (2.10)            | 2276 (27)              | 2110 (44)              | 1.08 (0.01)                    | 1.76 (0.26) |
|   | Winter                | 8.84 (1.54)            | 0.64 (0.10)            | 4.25 (1.12)            | 2291 (12)              | 2151 (17)              | 1.07 (0.01)                    | 1.57 (0.10) |
|   | Winter/spring         | 10.67 (0.71)           | 0.74 (0.06)            | 5.32 (0.86)            | 2302 (7)               | 2173 (12)              | 1.06 (0.01)                    | 1.44 (0.11) |
| Polar Water (PW)                          | Spring                | 10.56 (1.09)           | 0.72 (0.07)            | 5.44 (1.05)            | 2296 (6)               | 2167 (10)              | 1.06 (0.00)                    | 1.43 (0.09) |
|   | Summer                | 6.35 (3.46)            | 0.52 (0.29)            | 3.04 (1.33)            | 2259 (39)              | 2113 (68)              | 1.07 (0.02)                    | 1.57 (0.28) |
|   | Winter                | 5.79 (2.19)            | 0.48 (0.13)            | 2.85 (1.26)            | 2266 (26)              | 2135 (44)              | 1.06 (0.01)                    | 1.49 (0.15) |
| Atlantic Water (AW)                       | Winter/spring         | 8.38 (0.88)            | 0.62 (0.05)            | 3.89 (0.70)            | 2282 (6)               | 2162 (9)               | 1.06 (0.00)                    | 1.38 (0.06) |
|   | Spring                | 7.98 (1.25)            | 0.59 (0.07)            | 4.13 (0.63)            | 2281 (8)               | 2159 (14)              | 1.06 (0.00)                    | 1.39 (0.08) |
|   | Summer                | 11.08 (2.43)           | 0.74 (0.08)            | 4.87 (1.11)            | 2303 (4)               | 2160 (8)               | 1.07 (0.01)                    | 1.54 (0.11) |
| Modified Atlantic Water (mAW)             | Winter                | 9.37 (1.57)            | 0.65 (0.13)            | 4.39 (0.82)            | 2304 (3)               | 2161 (3)               | 1.07 (0.00)                    | 1.63 (0.05) |
|   | Winter/spring         | 11.65 (0.12)           | 0.78 (0.01)            | 5.75 (0.00)            | 2308 (4)               | 2171 (0)               | 1.06 (0.00)                    | 1.40 (0.03) |
|   | Spring                | 9.80 (0.86)            | 0.69 (0.04)            | 5.00 (0.18)            | 2311 (1)               | 2164 (2)               | 1.07 (0.00)                    | 1.64 (0.01) |
| Intermediate Water (IW)                   | Summer                | 11.63 (0.66)           | 0.79 (0.02)            | 5.23 (0.89)            | 2303 (5)               | 2176 (13)              | 1.06 (0.01)                    | 1.34 (0.07) |
|   | Winter                | 11.58 (0.26)           | 0.83 (0.04)            | 6.67 (1.06)            | 2303 (5)               | 2176 (13)              | 1.06 (0.01)                    | 1.34 (0.06) |
|   | Winter/spring         | 11.55 (1.01)           | 0.81 (0.04)            | 5.76 (1.33)            | 2307 (3)               | 2179 (7)               | 1.06 (0.00)                    | 1.40 (0.10) |
| Cold Barents Sea Dense Water (CBSDW)      | Spring                | 10.79 (1.27)           | 0.73 (0.08)            | 5.78 (1.00)            | 2307 (4)               | 2171 (5)               | 1.06 (0.00)                    | 1.47 (0.12) |
|   | Summer                | 10.39 (0.58)           | 0.75 (0.04)            | 5.56 (0.79)            | 2297 (2)               | 2197 (6)               | 1.05 (0.00)                    | 1.18 (0.05) |
|   | Winter                | 10.93 (0.00)           | 0.84 (0.00)            | 7.45 (0.00)            | 2303 (0)               | 2215 (0)               | 1.04 (0.00)                    | 1.13 (0.00) |
| Eurasian Basin Deep Water (EBDW)          | Winter/spring         | 13.42 (0.00)           | 0.89 (0.00)            | 9.14 (0.16)            | 2306 (2)               | 2180 (6)               | 1.06 (0.00)                    | 1.16 (0.02) |
|   | Spring                | 13.32 (0.08)           | 0.89 (0.00)            | 8.81 (0.08)            | 2299 (4)               | 2169 (3)               | 1.06 (0.00)                    | 1.18 (0.05) |
|   | Summer                | 8.55 (0.48)            | 0.65 (0.04)            | 5.53 (0.88)            | 2300 (3)               | 2189 (10)              | 1.05 (0.00)                    | 1.23 (0.07) |
| Depth > 500 m                             | Winter                | 8.83 (0.41)            | 0.69 (0.00)            | 6.30 (0.12)            | 2298 (4)               | 2196 (6)               | 1.05 (0.00)                    | 1.24 (0.02) |
|   | Winter/spring         | -                      | -                      | -                      | -                      | -                      | -                              | -           |
|   | Spring                | -                      | -                      | -                      | -                      | -                      | -                              | -           |
| –1.1 °C < T <sub>C</sub> ≤ 0.0 °C         | Summer                | 14.32 (0.59)           | 0.95 (0.04)            | 10.62 (1.45)           | 2298 (3)               | 2165 (10)              | 1.06 (0.01)                    | 0.99 (0.14) |
|   | Winter                | 13.15 (0.40)           | 0.93 (0.03)            | 10.76 (1.24)           | 2303 (3)               | 2164 (5)               | 1.06 (0.00)                    | 1.06 (0.15) |
|   | Winter/spring         | 14.44 (0.13)           | 0.96 (0.01)            | 11.78 (0.49)           | 2308 (4)               | 2169 (6)               | 1.06 (0.00)                    | 1.00 (0.11) |
| S <sub>A</sub> > 35.06 g kg <sup>-1</sup> | Spring                | 14.34 (0.25)           | 0.96 (0.02)            | 11.22 (0.47)           | 2302 (3)               | 2163 (4)               | 1.06 (0.00)                    | 0.99 (0.12) |

Warm Polar Water is formed by solar heating of Polar Water and mixing with Atlantic-derived waters. Intermediate Water is typically found deeper than the Atlantic-influenced water masses with lower salinity. Cold Barents Sea Dense Water (CBSDW) is defined here as  $T_C \leq -1.1$  °C and is found only in the Barents Sea. Both Intermediate Water and CBSDW indicate the presence of strong winter convection due to cooling and brine release during sea-ice formation. North of the Barents Sea continental slope, Eurasian Basin Deep Water (EBDW) has an overlapping definition with Intermediate Water but includes waters deeper than 500 m north of the continental slope.

### 2.4. Carbonate chemistry

Seawater samples for carbonate chemistry analyses (C<sub>T</sub>, A<sub>T</sub> and pH<sub>T</sub>) were drawn from the Niskin bottles into 250 mL borosilicate glass bottles. During summer 2019 (Q3), winter/spring 2021 (Q1) and spring 2021 (Q2), C<sub>T</sub> and A<sub>T</sub> samples were preserved with 50 μL of saturated mercuric chloride solution and stored at ~ 4 °C for post-cruise analysis at the Institute of Marine Research, Tromsø, following Dickson et al. (2007). During winter 2019 (Q4), C<sub>T</sub> and pH samples were analysed

onboard at  $\sim 25^\circ\text{C}$  within approximately 24 h and  $A_T$  values were calculated using the  $C_T$  and pH measurements in the CO2SYS model.

Determination of  $C_T$  was carried out by gas extraction of acidified (10 %  $\text{H}_3\text{PO}_4$ ) samples followed by coulometric titration and photometric detection (Johnson et al., 1987) using a Versatile Instrument for the Determination of Titration carbonate (VINDTA 3D, Marianda, Germany). The  $A_T$  was determined by potentiometric titration with 0.1 M hydrochloric acid in a semi-open cell using a Versatile Instrument for the Determination of Titration Alkalinity (VINDTA 3S, Marianda, Germany). Measurements of  $C_T$  and  $A_T$  were corrected against Certified Reference Materials (CRM, provided by A. G. Dickson, Scripps Institution of Oceanography, USA). The measurements precision, as determined from the average standard deviation of replicate analyses, was within  $\pm 2 \mu\text{mol kg}^{-1}$ . Measurements for pH on the total hydrogen ion scale ( $\text{pH}_T$ ) were carried out during the winter 2019 (Q4) cruise using spectrophotometric determination (Agilent 8453 Diode-array) according to Clayton and Byrne (1993) and the purified indicator dye, *metacresol purple* (2 mM). 3 mL of the sample was mixed with 35  $\mu\text{L}$  of the indicator. The  $\text{pH}_T$  of the indicator was measured during each day using a 0.2 mm quartz cuvette. Correction for the perturbation of the indicator  $\text{pH}_T$  was performed according to Chierici et al. (1999). The precision for  $\text{pH}_T$  was determined from the average standard deviation for triplicate analyses as  $\pm 0.001$  units. The  $\text{pH}_T$  data were used in combination with the  $C_T$  data to calculate the winter 2019  $A_T$  values using the chemical speciation model CO2SYS, as detailed in section 2.8.

## 2.5. Inorganic nutrients

Seawater samples for the determination of the concentration of inorganic macronutrients nitrate ( $\text{NO}_3$ ), phosphate ( $\text{PO}_4$ ) and silicic acid ( $\text{Si}(\text{OH})_4$ ) were drawn from the Niskin bottles into 20 mL pre-rinsed HDPE vials, preserved with 250  $\mu\text{L}$  chloroform and stored at  $4^\circ\text{C}$ . Post-cruise analysis was performed using a spectrophotometric method following standard procedures (Grasshoff et al., 2009) at the Institute of Marine Research, Bergen, Norway (Gundersen, et al., 2022). The analyser was validated by routine measurements of reference seawater from Ocean Scientific International Ltd., UK. The detection limits were  $0.5 \mu\text{mol kg}^{-1}$  for  $\text{NO}_3$ ,  $0.06 \mu\text{mol kg}^{-1}$  for  $\text{PO}_4$  and  $0.7 \mu\text{mol kg}^{-1}$  for  $\text{Si}(\text{OH})_4$ . The data are reported in (Chierici et al., 2021a; Chierici et al., 2021b; Jones et al., 2022a; Jones et al., 2022b).

## 2.6. Seasonal drivers of $C_T$ and $A_T$ in the upper ocean

Changes in the carbonate system ( $C_T$  and  $A_T$ ) for each season were estimated by integrating the difference between the surface (upper 15 m) concentrations and concentrations at 120 m depth, which was below the seasonal mixed layer (Fig. S1-S2). The total change in  $C_T$  ( $\Delta C_T$ ) and  $A_T$  ( $\Delta A_T$ ) can be partitioned into the main physical, biological and biogeochemical drivers (e.g. Chierici et al., 2011; Fransson et al., 2013; Fransson et al., 2017; Jones et al., 2021) shown in Equations (1) and (2): salinity changes ( $\Delta C_{T \text{ sal}}$ ,  $\Delta A_{T \text{ sal}}$ ) from freshwater inputs, advection and mixing of different water masses, biological production of organic matter and remineralisation/respiration ( $\Delta C_{T \text{ bio}}$ ,  $\Delta A_{T \text{ bio}}$ ) and the formation and dissolution of biotic (aragonite, calcite) and abiotic (ikaite) calcium carbonate ( $\Delta C_{T \text{ CaCO}_3}$ ,  $\Delta A_{T \text{ CaCO}_3}$ ).

$$\Delta C_{T(\text{surface}-120\text{m})} = \Delta C_{T \text{ sal}} + \Delta C_{T \text{ bio}} + \Delta C_{T \text{ CaCO}_3} + \Delta C_{T \text{ flux+res}} \quad (1)$$

$$\Delta A_{T(\text{surface}-120\text{m})} = \Delta A_{T \text{ sal}} + \Delta A_{T \text{ bio}} + \Delta A_{T \text{ CaCO}_3} \quad (2)$$

Changes in  $C_T$  and  $A_T$  due to salinity changes were estimated by firstly carrying out salinity normalisation of  $C_T$  and  $A_T$  ( $C_{T \text{ sal}}$  and  $A_{T \text{ sal}}$ ) using a non-zero freshwater endmember ( $C_T^{S=0} < 0$ , therefore set to 0;  $A_T^{S=0} = 340 \mu\text{mol kg}^{-1}$ ) according to equation (3) (Friis et al. 2003), with in situ salinity ( $S$ ) and a reference salinity ( $S_{\text{ref}}$ ) of 34.9 (absolute salinity). The  $\Delta C_{T \text{ sal}}$  and  $\Delta A_{T \text{ sal}}$  were determined from  $C_{T \text{ sal}}$  and  $A_{T \text{ sal}}$  by integration

between the surface and 120 m depth, as above, and then subtracting these values from the total ( $\Delta C_T$  and  $\Delta A_T$ ).

$$X_{T \text{ sal}} = ((X_T - X_T^{S=0}/S)S_{\text{ref}}) + X_T^{S=0} \quad (3)$$

Photosynthetic uptake of  $C_T$  ( $\Delta C_{T \text{ bio}}$ ; Eqn. 4) was determined from changes in salinity-normalised  $\text{NO}_3$  (using the traditional method in Friis et al. (2003)), integrated between the surface and 120 m depth, and applying the Redfield C:N stoichiometric ratio for C:N of 106:16 = 6.6 (Redfield et al. 1963). The Redfield ratio was selected as being representative of average C:N uptake ratios suitable for biological production estimates in Atlantic and Arctic waters in this region (Frigstad et al., 2014; Chierici et al., 2019; Jones et al., 2021).

$$\Delta T_{T \text{ bio}(\text{surface}-120\text{m})} = \Delta \text{NO}_3 \text{ sal}(\text{surface}-120\text{m}) \times 6.6 \quad (4)$$

Biological uptake of  $1 \mu\text{mol kg}^{-1}$  of  $C_T$  during primary production is accompanied by a decrease of  $16/106 = \sim 0.15 \mu\text{mol kg}^{-1} \text{NO}_3$ , which causes an increase in  $A_T$  of  $\sim 0.15 \mu\text{mol kg}^{-1}$  through the uptake of  $\text{NO}_3$  with  $\text{H}^+$  to conserve electroneutrality (Brewer and Goldman, 1976). The value for  $\Delta A_{T \text{ bio}}$  is therefore estimated from  $\Delta C_{T \text{ bio}}$  by applying  $0.15 \mu\text{mol } A_T \text{ kg}^{-1}$  per  $1 \mu\text{mol } C_T \text{ kg}^{-1}$  removed during biological production of organic matter (Eqn. (5)). Conversely, remineralisation of organic matter from heterotrophic respiration releases increases  $C_T$  and  $\text{NO}_3$  and decreases  $A_T$  with same ratios.

$$\Delta A_{T \text{ bio}(\text{surface}-120\text{m})} = \Delta C_{T \text{ bio}(\text{surface}-120\text{m})} - 0.15 \quad (5)$$

The  $\Delta C_{T \text{ CaCO}_3}$  term accounts for the formation or dissolution of any  $\text{CaCO}_3$  mineral (biotic aragonite and calcite; abiotic ikaite). Dissolution of  $\text{CaCO}_3$  produces two bicarbonate ions and consumes one mol of  $\text{CO}_2$ , thereby increasing  $A_T$  by a factor of two relative to the increase in  $C_T$  (Zeebe and Wolf-Gladrow, 2001). Formation of  $\text{CaCO}_3$  affects  $A_T$  and  $C_T$  negatively by the same 2:1 ratio. Firstly, residual changes in  $A_T$  between the surface and 120 m depth are determined by subtracting  $\Delta A_{T \text{ sal}}$  and  $\Delta A_{T \text{ bio}}$  from  $\Delta A_T$ , which are ascribed to  $\text{CaCO}_3$  processes, and then  $\Delta C_{T \text{ CaCO}_3}$  calculated (Eqn. (6)).

$$\Delta C_{T \text{ CaCO}_3(\text{surface}-120\text{m})} = \Delta A_{T \text{ CaCO}_3(\text{surface}-120\text{m})} 0.5 \quad (6)$$

The  $\Delta A_{T \text{ CaCO}_3}$  and  $\Delta C_{T \text{ CaCO}_3}$  terms refer to changes due to  $\text{CaCO}_3$  processes but cannot distinguish which  $\text{CaCO}_3$  mineral (aragonite, calcite, ikaite) is involved. As  $\Delta A_{T \text{ CaCO}_3}$  is determined from residual changes from the other  $\Delta A_T$  terms (Eqn. (2)) it may include  $A_T$  contributions from other minor sources e.g. mineral weathering, which is assumed to be insignificant in this region (Jones et al., 2021) and errors associated with the other terms.

The  $\Delta C_{T \text{ flux+res}}$  term is the difference between  $\Delta C_T$  and the sum of the other terms ( $\Delta C_{T \text{ sal}} + \Delta C_{T \text{ bio}} + \Delta C_{T \text{ CaCO}_3}$ ), which accounts for changes in  $C_T$  due to air-sea  $\text{CO}_2$  exchange ( $\Delta C_{T \text{ flux}}$ , section 2.9; uptake or release of 1 mol of  $\text{CO}_2$  only affects  $C_T$  by an increase or decrease of 1 mol, respectively), residual changes from other processes such as input of  $\text{CO}_2$ -rich brines, and errors associated with the other  $\Delta C_T$  terms.

## 2.7. Net community production

Net community production (NCP;  $\text{mol C m}^{-2}$ ) was estimated at the process (P) stations from changes in salinity-normalised nitrate concentrations ( $\Delta \text{NO}_3 \text{ sal}$ ;  $\mu\text{mol kg}^{-1}$ ) during the seasonal transitions throughout the growing season: winter/spring (March 2021) and spring (April-May 2021) and summer (August 2019). It was assumed that the replete  $\text{NO}_3$  concentrations in the well-mixed water column in winter/spring (March 2021) were representative of pre-bloom conditions in order to estimate NCP by summer of a different year, i.e. August 2019. The  $\Delta \text{NO}_3 \text{ sal}$  was integrated in the upper 50 m of the water column and converted to carbon using C:N stoichiometric ratio 106:16 (Redfield et al., 1963), as described above for determining  $\Delta C_{T \text{ bio}}$  in the upper

120 m. The upper 50 m depth range was selected for the NCP estimates as this depth was below the typical mixed layer depth and within the nitracline in summer for this region (Koenig et al., this issue b; Renner et al., this issue), and for comparisons to other NCP estimates in the Atlantic Water inflow and Barents Sea (Assmy et al., 2017; Fransson et al., 2017; Chierici et al., 2019).

## 2.8. Ocean acidification state

Values of  $\Omega_{\text{aragonite}}$  and  $\Omega_{\text{calcite}}$  and the  $A_T:C_T$  ratio (in situ values of  $A_T$  relative to  $C_T$ ) can be used as indicators of the ocean acidification state in relation to changes in  $C_T$  ( $\Delta C_T$ ) and  $A_T$  ( $\Delta A_T$ ) in the upper 120 m driven by the processes discussed in section 2.6. Changes in  $C_T$  and  $A_T$

affect  $\Omega$ ; increased  $C_T$  and decreased  $A_T$  suppresses  $\Omega$  and decreased  $C_T$  and increased  $A_T$  enhances  $\Omega$ . The  $\Omega$  was calculated from input pairs of  $C_T$  and  $A_T$  (Q3, Q1, Q2) and  $C_T$  and pH (Q4) with in situ temperature, salinity, pressure and concentrations of  $\text{PO}_4$  and  $\text{Si}(\text{OH})_4$  using the chemical speciation model CO2SYS (Pierrot et al., 2006). The carbonic acid dissociation constants of Mehrbach et al. (1973), as refitted by Dickson and Millero (1987), were used with the bisulfate dissociation constant from Dickson (1990) and the total boron concentration of Lee et al. (2010). The aragonite and calcite stoichiometric solubility constants of Mucci (1983) were used with the pressure corrections of Millero (1979) and the  $\text{CaCO}_3$  and salinity ratio of Riley and Tongudai (1967). When  $\Omega < 1$  the seawater is undersaturated with respect to  $\text{CaCO}_3$  and  $\text{CaCO}_3$  minerals are vulnerable to dissolution. Discussions on

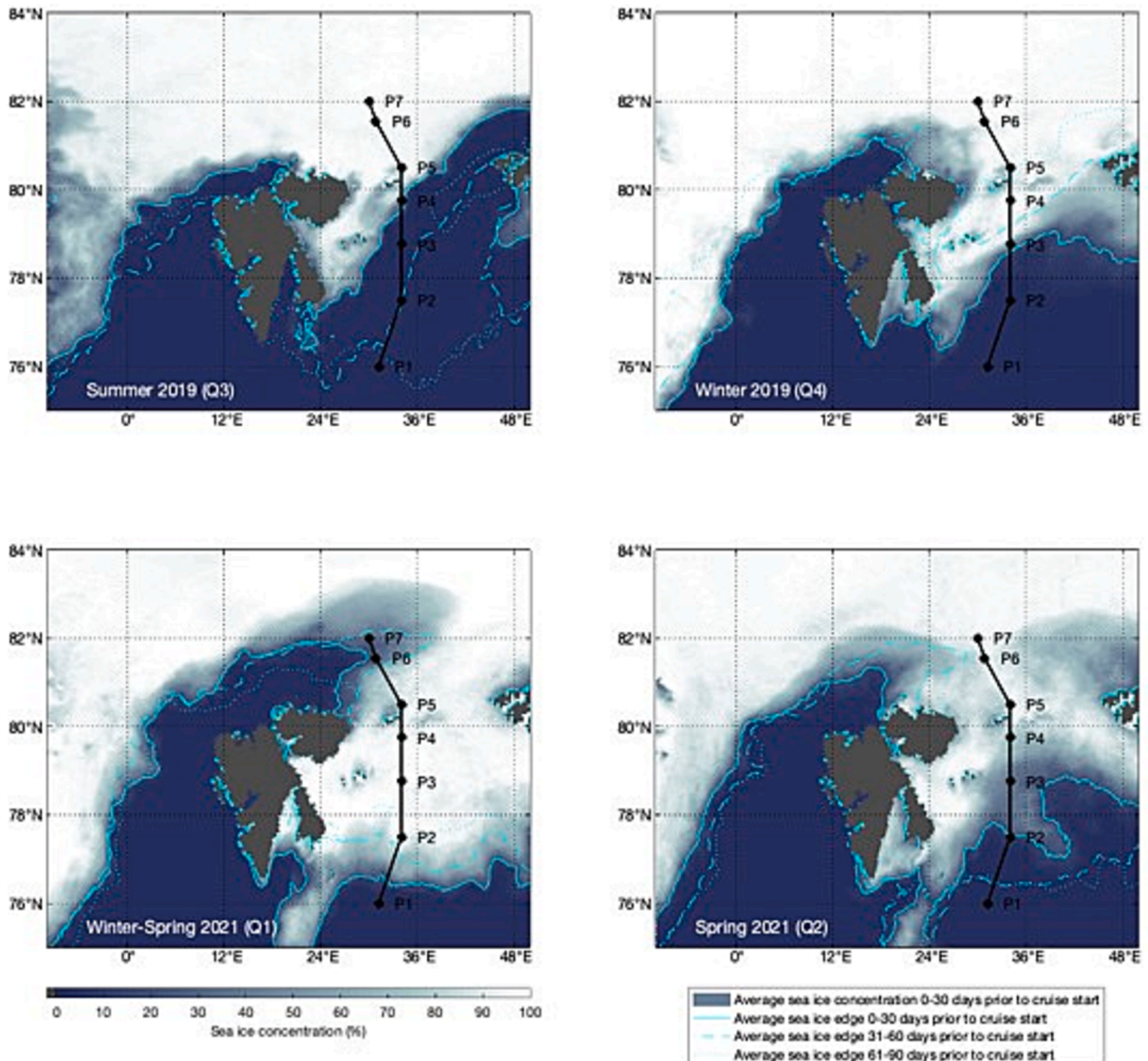


Fig. 2. Sea-ice conditions during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April-May 2021). The locations of the main process (P) stations are shown. Average sea ice concentration (%) is shown (shading) with average location of the ice edge at 0–30 days (solid cyan line), 31–60 days (dashed cyan line) and 61–90 days (dotted cyan line) indicated. Sea ice concentration data derived from AMSR-2 was downloaded from the Institute of Environmental Physics, University of Bremen, Germany (Spren et al., 2008). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

the state of ocean acidification will focus on  $\Omega$ aragonite as aragonite is the more unstable biotic  $\text{CaCO}_3$  mineral.

## 2.9. Air-sea $\text{CO}_2$ fluxes

Air-sea  $\text{CO}_2$  fluxes were calculated from the air-sea gradient in the fugacity of  $\text{CO}_2$  ( $\Delta f_{\text{CO}_2}$  sea-air) following the procedure described in Jones et al. (2021). The  $f_{\text{CO}_2}$  in surface water was calculated from input pairs of  $C_T$  and  $A_T$  (Q3, Q1, Q2) and  $C_T$  and pH (Q4) with using the chemical speciation model CO2SYS as described in section 2.8. The atmospheric mixing ratio of  $\text{CO}_2$  ( $x_{\text{CO}_2}$ ) was obtained from the Zeppelin Observatory, Spitsbergen (78.907°N, 11.888°E) and was converted to  $f_{\text{CO}_2}$  following Dickson et al. (2007) and Ericson et al., 2023, and references therein. Seasonal mean wind speeds (mean value of observations from 3 months prior to each seasonal cruise) were obtained from the Verlegenhukken weather station, Spitsbergen (80.056°N, 16.243°E), from <https://seklima.met.no/observations/>. Sea ice concentrations at the location of each hydrographic station (Fig. 2) were estimated from daily sea ice concentration data, as reported in Van Engeland et al. (2023), and were used to determine ice-scaled  $\text{CO}_2$  fluxes (assuming 100 % sea ice cover limits significant air-sea  $\text{CO}_2$  exchange).

A negative air-sea  $\text{CO}_2$  flux indicates oceanic uptake of  $\text{CO}_2$ , increasing  $C_T$  in the surface layer, and that the ocean is a sink for atmospheric  $\text{CO}_2$ . To account for the effects of  $\text{CO}_2$  uptake on changes in  $C_T$  in the surface layer ( $\Delta C_{T \text{ flux}}$ ) the corresponding total change in  $C_T$  (mol  $\text{m}^{-2}$  per season) was calculated using the daily ice-scaled fluxes summed for the 3 months (average 90 days) to represent each season. The  $\Delta C_{T \text{ flux}}$  in the surface layer is compared to integrated changes in the upper 120 m due to air-sea  $\text{CO}_2$  fluxes and residual effects ( $\Delta C_{T \text{ flux+res}}$ ), as described in section 2.6.

## 2.10. Uncertainties in analytical techniques and calculations

Uncertainties associated with the total  $\Delta C_T$  (Eqn. (1) and  $\Delta A_T$  (Eqn. (2) were based on the analytical precision of the measurements ( $\pm 2 \mu\text{mol kg}^{-1}$ ) and estimated (square-root-of-sum-of-squares of the absolute errors) as  $\pm 3 \mu\text{mol kg}^{-1}$ . Similarly, uncertainties in  $\Delta C_{T \text{ sal}}$  and  $\Delta A_{T \text{ sal}}$  were  $\pm 3 \mu\text{mol kg}^{-1}$ . Uncertainties in  $\Delta C_{T \text{ bio}}$  and  $\Delta A_{T \text{ bio}}$  were estimated from the combination of analytical precision for  $\text{NO}_3$  and deviations in the C:N ratio (Redfield et al., 1963; Frigstad et al., 2014). An upper bound of the error for  $\Delta \text{NO}_3 \text{ sal}$  was  $\pm 0.1 \mu\text{mol kg}^{-1}$ , corresponding to errors  $C_{T \text{ bio}} \pm 1 \mu\text{mol kg}^{-1}$  and variability in C:N of  $\pm 1 \mu\text{mol kg}^{-1}$ , yielding composite uncertainties for  $\Delta C_{T \text{ bio}} \pm 2 \mu\text{mol kg}^{-1}$  and  $\Delta A_{T \text{ bio}} \pm 1 \mu\text{mol kg}^{-1}$ . Upper bound uncertainties in  $\Delta A_{T \text{ CaCO}_3}$  and  $\Delta C_{T \text{ CaCO}_3}$  were estimated from combined uncertainty as  $\pm 4 \mu\text{mol kg}^{-1}$ . Associated errors on  $\Delta C_{T \text{ flux+res}}$  are estimated as  $\pm 4 \mu\text{mol kg}^{-1}$ .

## 3. Results

### 3.1. Sea ice and hydrography

The northern Barents Sea and Nansen Basin were ice covered during all seasons (Fig. 2), varying from minimum ice cover in summer (August 2019) to maximum ice cover in winter (March 2021) with the ice edge indicating the MIZ. The variations in conservative temperature and absolute salinity (Fig. 3) and potential density characterises (Table 2) identified key water masses in the water column (Fig. 4). During all seasons, Atlantic Water, warm Polar Water and modified Atlantic Water dominated the water column south of the Polar Front at P1 (Fig. 3–4). Shoaling of isotherms and isohalines to the surface marked the influence of Atlantic Water along the northern shelf, which was most pronounced in winter (December 2019). Eurasian Basin Deep Water (EBDW) was identified during each season at about 1500–3500 m in the Nansen Basin. The definition of EBDW overlaps with that of Intermediate Water (IW) in Sundfjord et al. (2020), which will be referred to hereafter as IW/EBDW and was found at 800–1500 m depth in the Nansen Basin. The

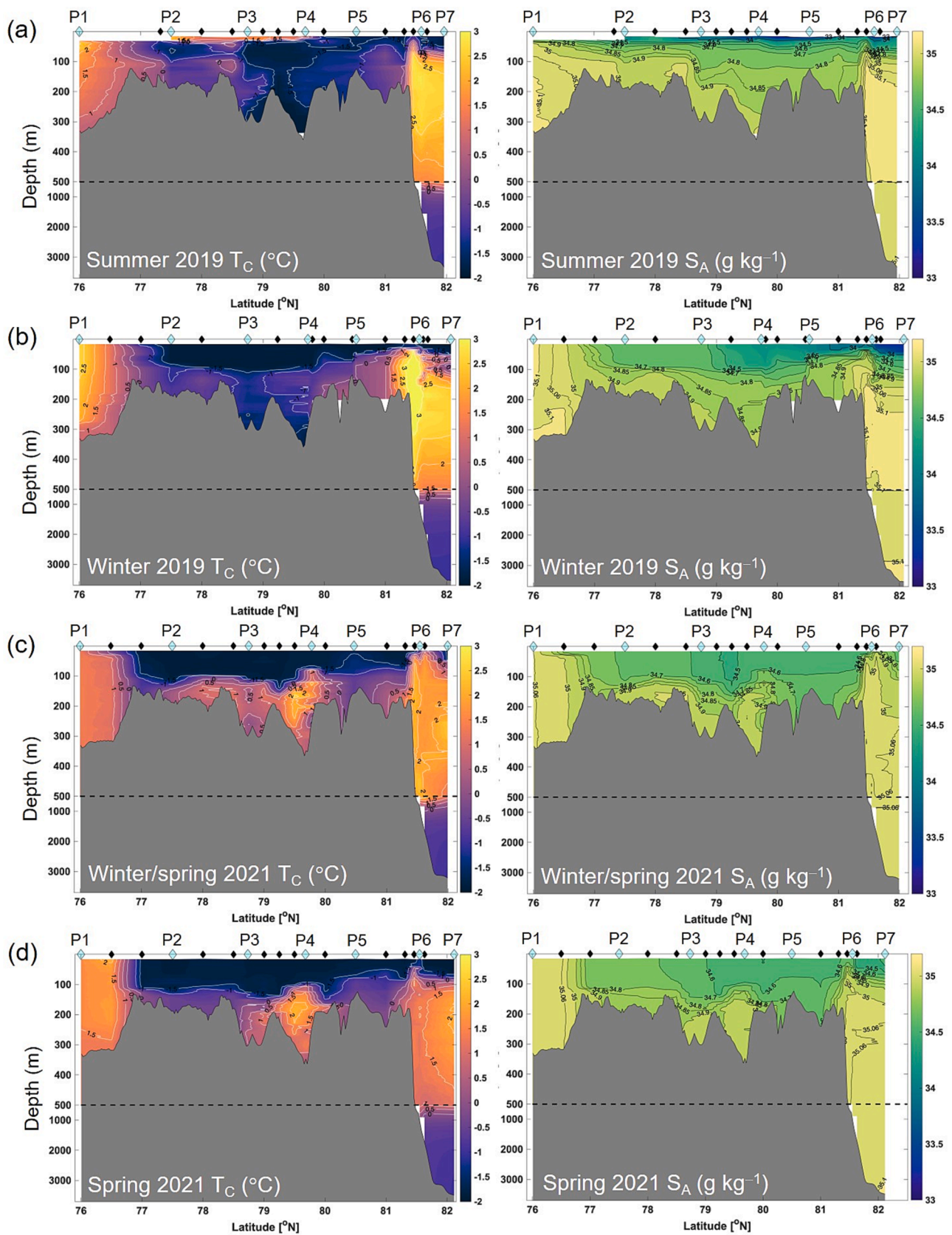
IW/EBDW likely has Intermediate Water as a source water mass but has been isolated from the atmosphere and accumulated over multiple years or decades whereas Intermediate Water on the Barents Sea shelf exhibited seasonal signals from wintertime ventilation and accumulation of saline brines from forming sea ice.

In summer (August 2019), sea ice was present across the shelf of the Barents Sea with the ice edge located north of 79.75°N and east of 34°E between P4 and P5 (Fig. 2). Warm Polar Water occupied the upper layer (10–45 m depth) from P2 to P4 with an underlying layer of Polar Water (Fig. 3a, 4). In the deepest layers, Intermediate Water was found at P2 and cold, dense CBSDW was found across the central shelf from P3 to P4 (Fig. 4). The whole water column around P5 was dominated by Polar Water. Around P6 on the shelf and northwards to P7 in the Nansen Basin, colder and fresher Polar Water defined the surface layer, with underlying warm Polar Water, Atlantic Water and modified Atlantic Water at increasing depth. By winter (December 2019), sea ice cover extended southwards with the ice edge located at around P3 (Fig. 2). The water column at P1 was warmer and more saline compared with summer, with modified Atlantic Water found in the deepest layer (Fig. 3b, 4). The ice-covered water column north of P2 was predominantly colder and fresher Polar Water. Warm Polar Water re-emerged at around P5 and towards P6 in the deeper layers. Atlantic Water was identified by a more pronounced warm core between ~ 150 and 210 m depth from P6 to P7.

Winter/spring (March 2021) was characterised by the seasonal maximum in sea ice cover, with the ice edge located between P1 and P2 (Fig. 2). The ice-free and well mixed water column at P1 was occupied by modified Atlantic Water (Fig. 3c, 4). Polar Water with underlying warm Polar Water characterised the water column from P2 to around P4. The whole water column at P5 was dominated by Polar Water. Below the surface Polar Water layer north of P5, warm Polar Water and modified Atlantic Water were identified in mid (50–200 m) and deep (490–840 m) depth ranges, respectively. North of the shelf break, Atlantic Water was only found at P6 at around 500 m depth. Intermediate Water occupied the deep (1000–3370 m) layers from P6 to P7. No Intermediate Water or CBSDW was detected over the Barents Sea shelf. Recent decreasing and increasing sea ice cover was evident between P6 and P7, which coincided with warm Polar Water intruding the surface layer (Fig. 3c). By spring (April–May 2021), the onset of seasonal sea ice retreat was evident by the more northerly location of the ice edge around P2 (Fig. 2). P1 remained well mixed with modified Atlantic Water (Fig. 3d). The seasonal sea ice melt influenced the water column with cold and fresh Polar Water at all depths north of the Polar Front to P3 (Fig. 3d, 4). Across the shelf from P3 to around P6, warm Polar Water was found in the deeper layers, with the exception of P5 where Polar Water was present at all depths (same as observed during summer and winter/spring). North of P6, modified Atlantic Water was found in deeper (750–900 m) layers and no Atlantic Water was identified north of the shelf break. No Intermediate Water or CBSDW was detected over the Barents Sea shelf in 2021.

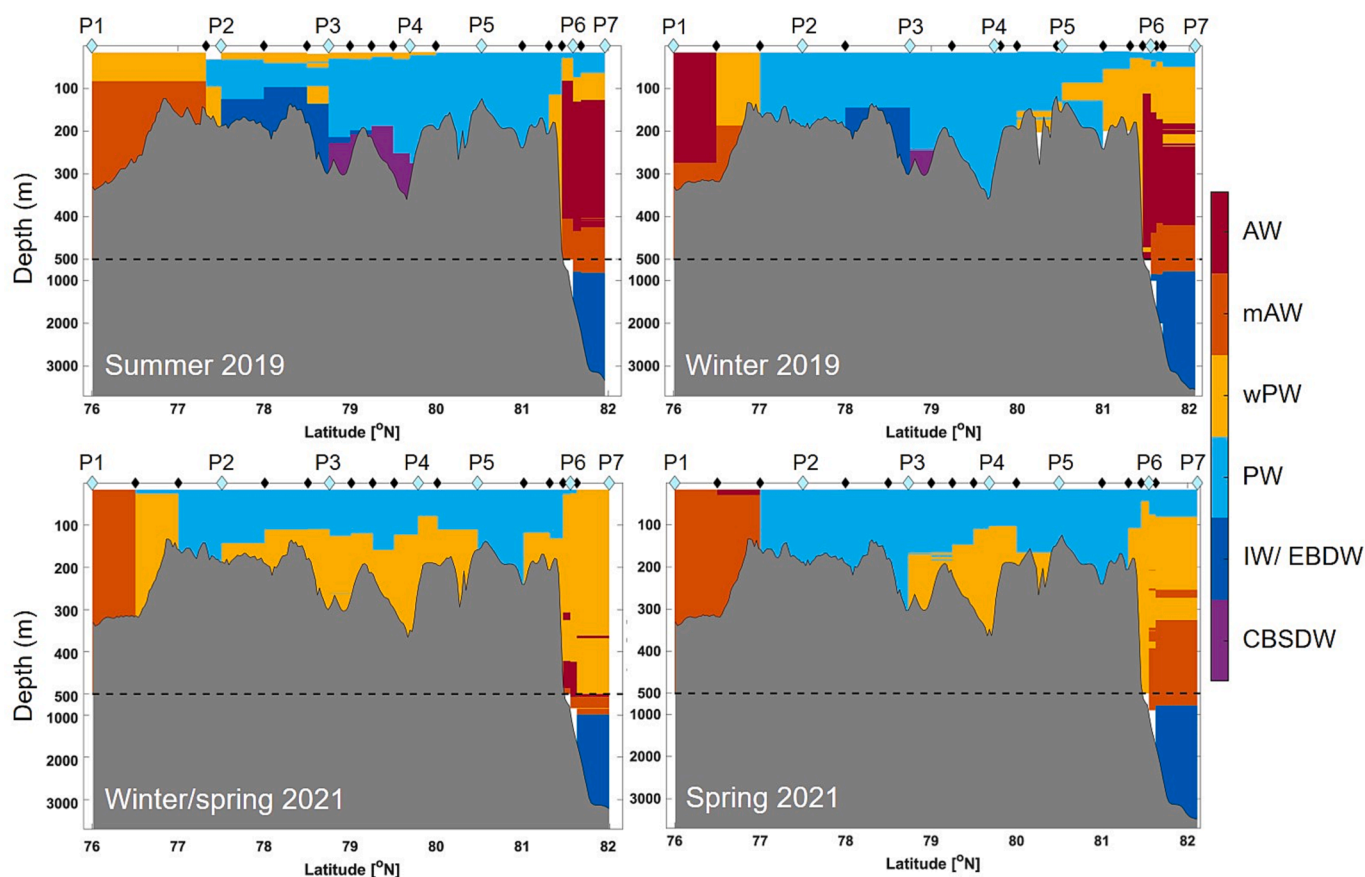
### 3.2. Inorganic nutrients and carbonate chemistry

In summer (August 2019), depleted  $\text{NO}_3$ ,  $\text{PO}_4$  and  $\text{Si(OH)}_4$  characterised much of the upper layers of Polar Water and warm Polar Water (Fig. 5, S1). Concentrations increased with depth to Intermediate Water overlying the shelf (150–200 m depth) and IW/EBDW in the deep Nansen Basin, where highest  $\text{NO}_3$  and  $\text{Si(OH)}_4$  concentrations of ~ 15 and ~ 12  $\mu\text{mol kg}^{-1}$ , respectively, were found (Table 2). Lowest  $A_T$  (2134  $\mu\text{mol kg}^{-1}$ ) and  $C_T$  (1930  $\mu\text{mol kg}^{-1}$ ) were found in the surface Polar Water layer (Fig. 6). Values of  $A_T$  and  $C_T$  increased with depth to highest  $A_T$  (2317  $\mu\text{mol kg}^{-1}$ ) in Atlantic Water (at P1) and highest  $C_T$  (2203  $\mu\text{mol kg}^{-1}$ ) in Intermediate Water (at P2) over the shelf (Table 2). Deepest layers over parts of the shelf that were occupied by CBSDW (around P3 and P4) showed reductions in  $\text{NO}_3$ , and similar values of  $\text{Si(OH)}_4$  and  $A_T$ , and increased  $C_T$ , relative to the overlying Polar Water



**Fig. 3.** Transect section plots of conservative temperature ( $T_C$ ;  $^{\circ}\text{C}$ ) and absolute salinity ( $S_A$ ;  $\text{g kg}^{-1}$ ) during (a) summer (August 2019), (b) winter (December 2019), (c) winter/spring (March 2021) and (d) spring (April-May 2021). The locations of the hydrographic stations (black diamonds) and process (P) stations (blue diamonds) are marked and labelled. The y-axis is expanded in the upper 500 m. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)





**Fig. 4.** Water mass distribution during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021) colour-coded to identify Warm Polar Water (wPW), Polar Water (PW), Atlantic Water (AW), Modified Atlantic Water (mAW), Intermediate Water (IW), Cold Barents Sea Dense Water (CBSDW) and Eurasian Basin Deep Water (IW/ EBDW). The locations of the hydrographic stations (black diamonds) are marked and process (P) stations (blue diamonds) are marked and labelled. The y-axis is expanded in the upper 500 m. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

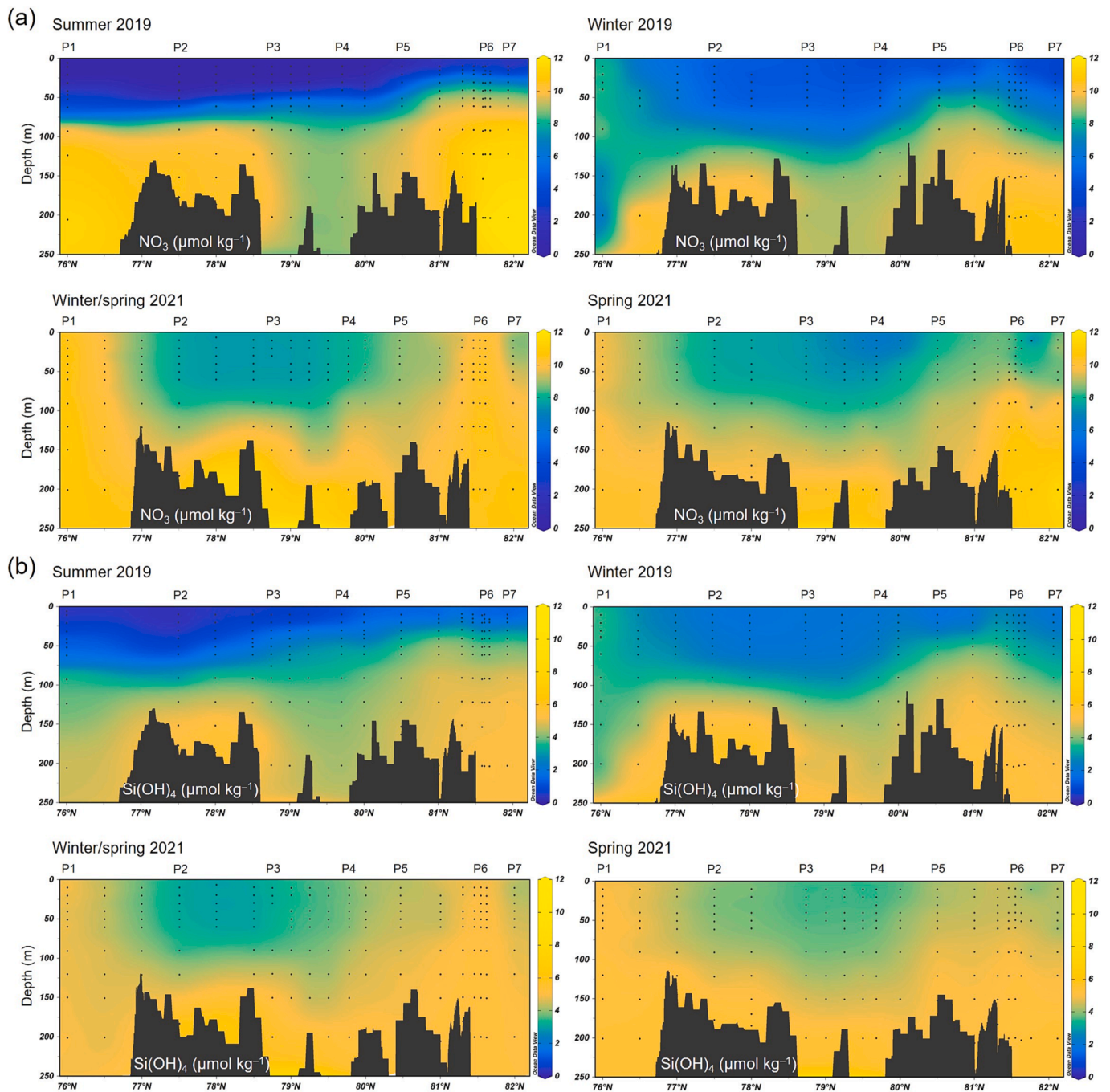
(Fig. 5). Lowest average surface layer  $A_T$  occurred in the Polar Water (Table 2). The water column at and around P5 was dominated by Polar Water with low  $A_T$  and  $C_T$  and exhibited little variability for all seasons. By winter (December 2019), the water column was becoming replenished with nutrients and with higher  $A_T$  and  $C_T$  south of the ice edge, with an imprint of summer meltwater inputs and primary production in the surface layer (Fig. 5 – 6). Nutrients,  $C_T$  and  $A_T$  generally increased with depth to the Intermediate Water, except for decreases in CBSDW (at P3 and P4). High  $C_T$  in Polar Water and Intermediate Water occurred at and around P2. Compared with the other seasons, Atlantic Water concentrations of  $\text{NO}_3$  and  $\text{Si}(\text{OH})_4$  were generally lowest in winter.

In winter/spring (March 2021), the well-mixed water column was relatively homogenous and rich in nutrients and  $C_T$  and with higher  $A_T$  (Fig. 5 – 6, S1). Relative to all seasons, highest average values in warm Polar Water of  $\text{NO}_3$ ,  $A_T$  and  $C_T$  and Polar Water concentrations of  $\text{NO}_3$ ,  $A_T$  and  $C_T$  occurred (Table 2). Highest average values of Atlantic Water concentrations of  $\text{NO}_3$ ,  $\text{Si}(\text{OH})_4$  and  $C_T$  also occurred during winter/spring. Highest average  $\text{NO}_3 \sim 15 \mu\text{mol kg}^{-1}$  and  $\text{Si}(\text{OH})_4 \sim 13 \mu\text{mol kg}^{-1}$  concentrations were found in IW/EBDW in the Nansen Basin (Table 2). By spring (April–May 2021), the water column was still largely homogenous, nutrient replete and with high  $A_T$  and  $C_T$  (Fig. 5 – 6, S1). Nutrient concentrations were similar in the surface and Atlantic Water layers, increased in Intermediate Water and  $A_T$  and  $C_T$  were highest ( $A_T \sim 2313 \mu\text{mol kg}^{-1}$ ;  $C_T \sim 2184 \mu\text{mol kg}^{-1}$ ) in the Atlantic Water layers. Values for  $\text{NO}_3$ ,  $C_T$  and  $A_T$  were slightly reduced in the surface layer compared with winter/spring. In contrast,  $\text{Si}(\text{OH})_4$  in warm Polar Water and Polar Water were the highest in all seasons (Fig. 5b; Table 2).

### 3.3. Biogeochemical-physical coupling and ocean acidification state

For each season, variability in  $A_T$  and  $C_T$  in the water column is closely linked to the distribution to different water masses that exhibit different  $A_T$  and  $C_T$  and thus varying  $A_T:C_T$  from physical, biogeochemical and biological processes. Largest variations in  $A_T$  and  $C_T$  during summer coincided with a wider range of salinities from inputs of fresh sea-ice meltwater (Fig. 7a). The variability in  $A_T$  was more closely coupled to salinity as expected from the near-conservative behaviour of  $A_T$ , showing lower  $A_T$  in low salinity Polar Water and higher  $A_T$  with saline Atlantic Water. Increased variability in  $C_T$  shows effects of other processes driving changes in  $C_T$ , particularly in summer.

The  $\Omega_{\text{aragonite}}$  varied between 0.76 and 2.30 in the full water column across all seasons controlled by changes in  $A_T$  and  $C_T$ , as reflected in the  $A_T:C_T$  ratio that ranged between 1.04 and 1.12 with effects of temperature (coldest and low  $\Omega_{\text{aragonite}}$  in winter/spring) and pressure (high pressure and low  $\Omega_{\text{aragonite}}$  in the deep Nansen Basin). Generally high  $\Omega_{\text{aragonite}}$  (mean 1.57–1.76) occurred in the surface warm Polar Water and Polar Water layers in summer, driven by reduced  $C_T$  (Table 2; Fig. 7b). Highest  $\Omega_{\text{aragonite}}$  of 2.25–2.30, and most buffered waters with mean  $A_T:C_T$  up to 1.08, occurred in warm Polar Water (at P1). The saline and high- $A_T$  Atlantic Water and modified Atlantic Water typically had high  $\Omega_{\text{aragonite}}$  with seasonal mean values of 1.34–1.64 and moderate buffer capacity with mean  $A_T:C_T$  of 1.06–1.07. Over the Barents Sea shelf, dense and cold CBSDW had  $\Omega_{\text{aragonite}}$  around 1.23 and had low buffer capacity (mean  $A_T:C_T$  ratio of 1.05) driven by elevated  $C_T$  in summer and winter. Generally low  $\Omega_{\text{aragonite}}$  of 1.12–1.14 (mean  $A_T:C_T$  ratio of 1.04–1.05) was found at 150–180 m in



**Fig. 5.** Transect section plots of (a)  $\text{NO}_3$  ( $\mu\text{mol kg}^{-1}$ ) and (b) silicic acid ( $\text{Si(OH)}_4$ ) ( $\mu\text{mol kg}^{-1}$ ) in the upper 250 m during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April-May 2021). The locations of the process (P) stations are labelled. Ocean Data View was used for data visualisation (Schlitzer, 2022).

the lower- $A_T$  and higher- $C_T$  Polar Water and Intermediate Water (at and around P2) in summer and winter. Aragonite undersaturation and lowest  $\Omega_{\text{aragonite}}$  of 0.76–0.78 occurred at 2800–3600 m in the lower- $A_T$  and nutrient-rich IW/EBDW in the Nansen Basin (at and around P7).

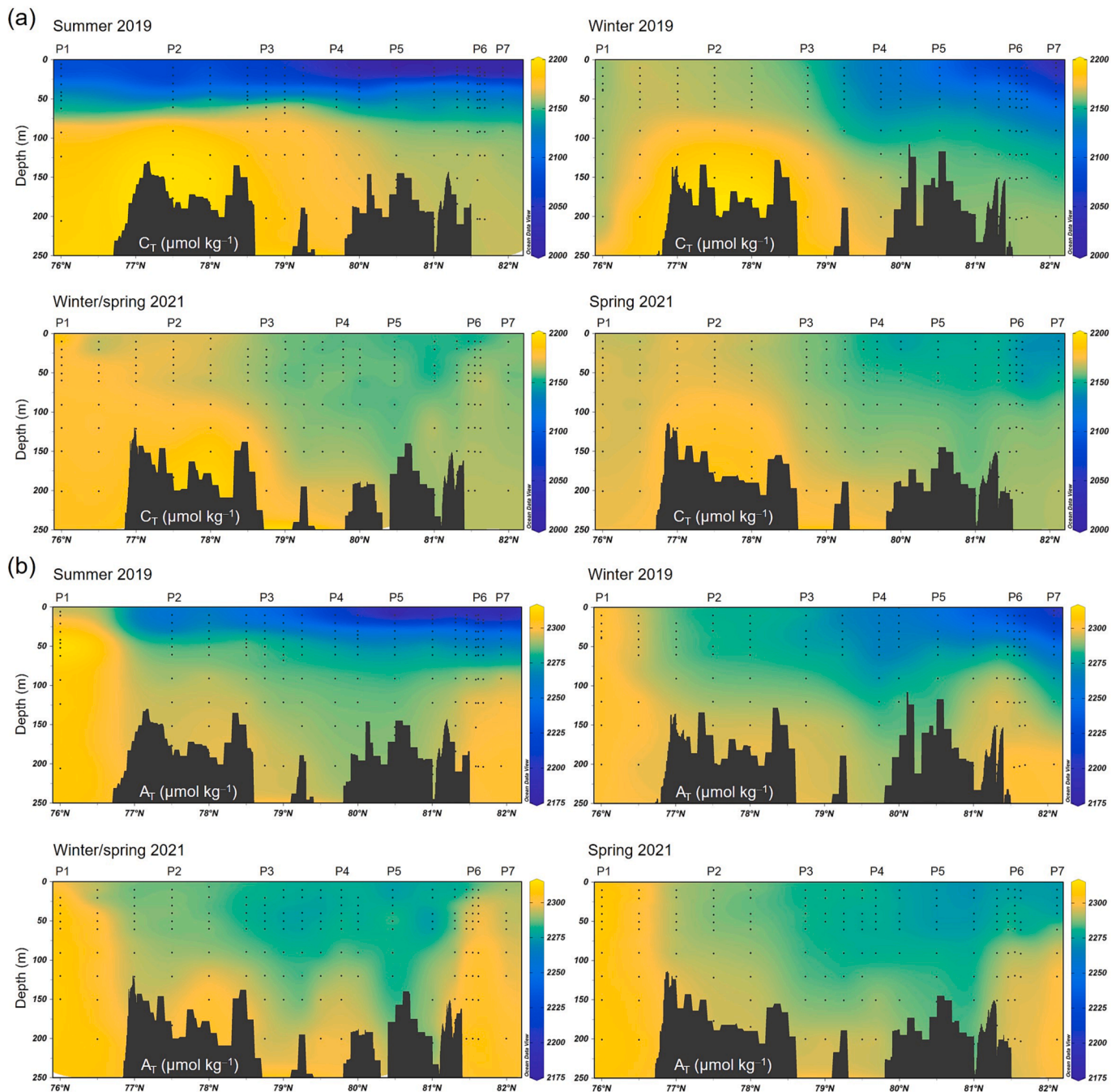
All water column nutrient and  $C_T$  data yielded close coupling and significant relationships between the cycling of  $C_T$  and  $\text{NO}_3$ ,  $\text{NO}_3$  and  $\text{PO}_4$ , and  $\text{Si(OH)}_4$  and  $\text{NO}_3$  (Table 3; Fig. 8). Nutrient uptake ratios for C: N (Fig. 8a) ranged from highest values of  $10.3 \pm 0.6$  in summer to lowest of  $2.6 \pm 0.4$  in winter/spring. Uptake ratios for N:P (Fig. 8b) ranged from  $17.2 \pm 0.3$  in winter/spring to  $16.3 \pm 0.3$  in winter. Uptake ratios for Si:N (Fig. 8c) ranged from  $0.94 \pm 0.03$  in winter/spring to  $0.50 \pm 0.02$  in summer. These inorganic carbon and nutrient ratios reveal

different stoichiometry between the seasons.

### 3.4. Seasonal changes in $C_T$ and $A_T$

#### 3.4.1. Seasonality and effects on ocean acidification state

Seasonal variability in  $C_T$  and  $A_T$  in the whole water column can be illustrated using theoretical trend lines to depict the impact of key processes, with potential synergy, and on the ocean acidification state,  $\Omega_{\text{aragonite}}$  (Fig. 9). Salinity-normalising  $C_T$  and  $A_T$  ( $C_{T \text{ sal}}$  and  $A_{T \text{ sal}}$ ) removes variability in  $C_T$  and  $A_T$  that result from salinity changes through physical processes, such as freshwater inputs, advection and mixing. The variability that remains in the  $A_{T \text{ sal}} - C_{T \text{ sal}}$  space results



**Fig. 6.** Transect section plots of (a)  $C_T$  ( $\mu\text{mol kg}^{-1}$ ) and (b)  $A_T$  ( $\mu\text{mol kg}^{-1}$ ) in the upper 250 m during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021). The locations of the process (P) stations are labelled. Ocean Data View was used for data visualisation (Schlitzer, 2022).

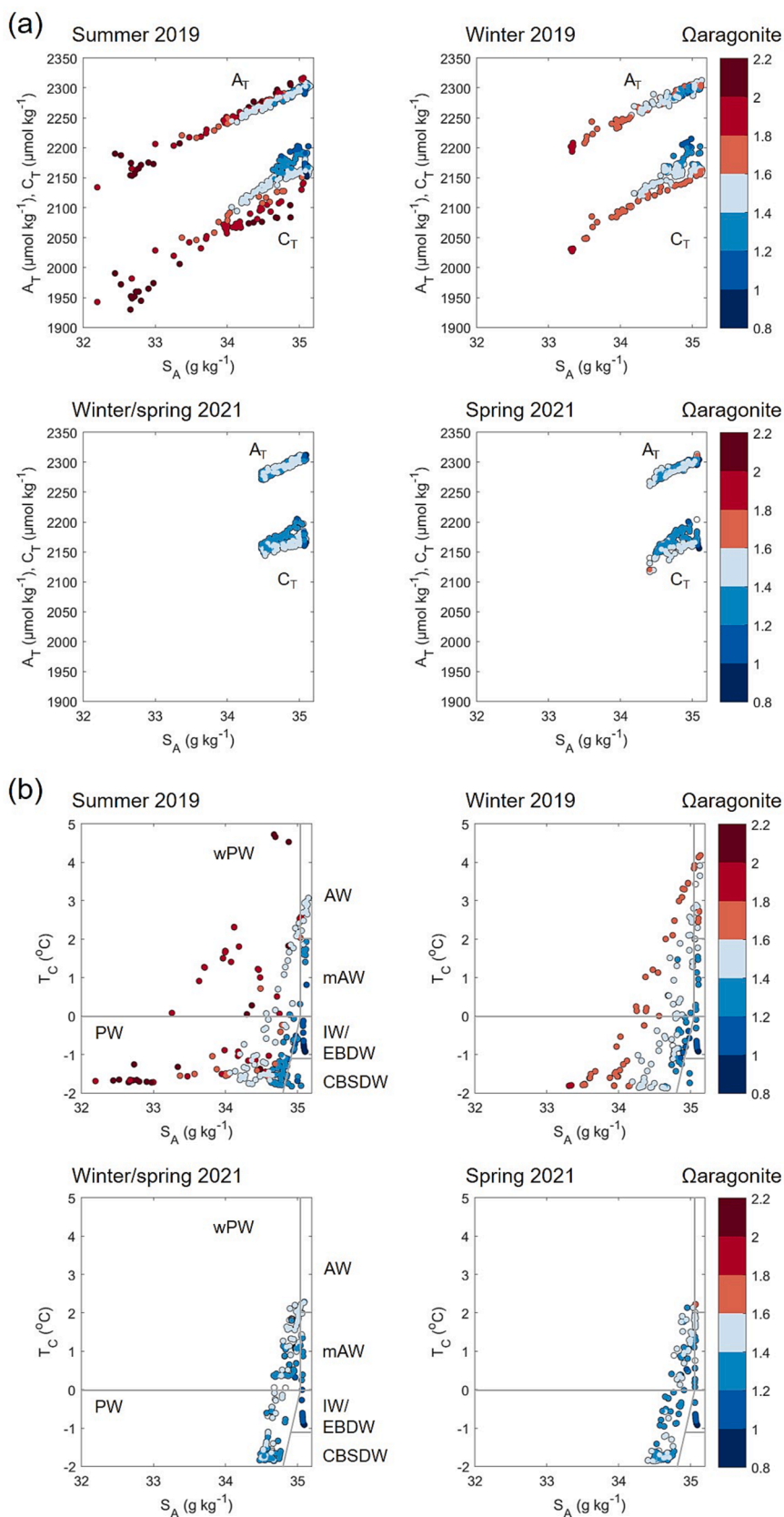
from biological processes of primary production and organic matter remineralisation (respiration) and biogeochemical processes, such as  $\text{CaCO}_3$  formation/dissolution and air-sea  $\text{CO}_2$  uptake/release.

Primary production reduced  $C_{T \text{ sal}}$ , with slight increases in  $A_{T \text{ sal}}$ , in summer, associated with elevated  $\Omega_{\text{aragonite}} > 1.8$ . The imprint of biological processes on  $C_{T \text{ sal}}$  likely contributed to the variability observed in the following winter (pre-conditioned from the growing season) and also in spring (at the start of the spring bloom). Remineralisation of organic matter contributed to higher values of  $C_{T \text{ sal}}$ , associated with lower  $\Omega_{\text{aragonite}} < 1.4$ , during the seasonal transitions. Reduced  $A_{T \text{ sal}}$  and  $C_{T \text{ sal}}$  in summer may also result from formation of biotic  $\text{CaCO}_3$  by calcifiers in the water column. Increased  $A_{T \text{ sal}}$  and  $C_{T \text{ sal}}$  and higher  $\Omega_{\text{aragonite}}$  could include contributions from dissolution of biotic  $\text{CaCO}_3$  and abiotic  $\text{CaCO}_3$  (ikaite). Reduced  $A_{T \text{ sal}}$  and  $C_{T \text{ sal}}$  and

lower  $\Omega_{\text{aragonite}}$  from winter/spring to spring could include a signal of ikaite formation in sea ice. The relative increase in  $C_{T \text{ sal}}$  at constant  $A_{T \text{ sal}}$  could result from atmospheric  $\text{CO}_2$  uptake in ice-free surface waters from summer to winter to spring due to an increase in  $\text{CO}_2$  solubility in cooler surface waters (decreased  $\Omega_{\text{aragonite}}$ ).

#### 3.4.2. Seasonal drivers in the upper ocean

The value of  $\Delta C_T$  and  $\Delta A_T$  represent 100 % of the total change as the sum of all seasonal drivers of  $C_T$  ( $\Delta C_{T \text{ sal}} + \Delta C_{T \text{ bio}} + \Delta C_{T \text{ CaCO}_3} + \Delta C_{T \text{ flux+res}}$ ) and  $A_T$  ( $\Delta A_{T \text{ sal}} + \Delta A_{T \text{ bio}} + \Delta A_{T \text{ CaCO}_3}$ ) changes, thus the absolute value of each term is taken to represent a fraction (%) of  $\Delta C_T$  and  $\Delta A_T$ , irrespective of the direction of change. The main driver of  $C_T$  and  $A_T$  changes from the surface layer to 120 m depth was salinity changes that accounted on average for 37–55 % of  $\Delta C_T$  and 78–85 % of  $\Delta A_T$



**Fig. 7.** Relationships between (a)  $A_T$  and  $C_T$  ( $\mu\text{mol kg}^{-1}$ ) and salinity ( $S_A$ ;  $\text{g kg}^{-1}$ ) and (b) temperature ( $T_C$ ;  $^{\circ}\text{C}$ ) and salinity ( $S_A$ ;  $\text{g kg}^{-1}$ ) during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021). Data are coloured by  $\Omega$ aragonite. Water masses warm Polar Water (wPW), Polar Water (PW), Atlantic Water (AW), modified Atlantic Water (mAW), Intermediate Water (IW), Eurasian Basin Deep Water (EBDW) and Cold Barents Sea Dense Water (CBSDW) are labelled in (b).

**Table 3**

Linear regression relationships for inorganic carbon (C), nitrate (N), phosphate (P) and silicic acid (Si) in the full water column for each season; summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021). Regression analysis information determined with Matlab (version 2017b) linear regression model fit:  $\text{gradient} \pm \text{standard error (se)}$ ,  $r^2$ , p value (statistically significant if  $p \leq 0.05$ ) and n number of observations.

| Linear regression | Season (year)        | Gradient $\pm$ se | $r^2$ | p      | n   |
|-------------------|----------------------|-------------------|-------|--------|-----|
| C:N               | Summer (2019)        | $10.3 \pm 0.6$    | 0.59  | < 0.01 | 200 |
|                   | Winter (2019)        | $7.3 \pm 0.7$     | 0.38  | < 0.01 | 178 |
|                   | Winter/spring (2021) | $2.6 \pm 0.4$     | 0.18  | < 0.01 | 182 |
|                   | Spring (2021)        | $2.9 \pm 0.5$     | 0.18  | < 0.01 | 190 |
| N:P               | Summer (2019)        | $16.8 \pm 0.2$    | 0.97  | $\ll$  | 200 |
|                   | Winter (2019)        | $16.3 \pm 0.3$    | 0.96  | $\ll$  | 176 |
|                   | Winter/spring (2021) | $17.2 \pm 0.3$    | 0.96  | $\ll$  | 190 |
|                   | Spring (2021)        | $17.0 \pm 0.2$    | 0.96  | $\ll$  | 188 |
| Si:N              | Summer (2019)        | $0.50 \pm 0.02$   | 0.76  | < 0.01 | 200 |
|                   | Winter (2019)        | $0.73 \pm 0.03$   | 0.81  | < 0.01 | 204 |
|                   | Winter/spring (2021) | $0.94 \pm 0.03$   | 0.83  | < 0.01 | 191 |
|                   | Spring (2021)        | $0.81 \pm 0.03$   | 0.83  | < 0.01 | 191 |

during all seasons (Fig. 10). Changes due to primary production/respiration accounted for an average of 23–37 % of  $\Delta C_T$  accompanied by change of 8–12 % of  $\Delta A_T$ . Formation/dissolution of  $\text{CaCO}_3$  represented an average 5–13 % of  $\Delta A_T$  with associated 1–3 % of  $\Delta C_T$ .

For  $\Delta C_{T \text{ flux+res}}$ , air-sea  $\text{CO}_2$  fluxes potentially represented an average of 4–37 % of  $\Delta C_T$ , which also included residual changes (e.g. input of  $\text{CO}_2$ -rich brines from sea ice) and uncertainties from the other  $\Delta C_T$  terms. Surface waters were undersaturated in  $\text{CO}_2$  relative to the atmosphere in all seasons (Fig. S3a), with strong undersaturation persisting in the Nansen Basin. Ice-scaled  $\text{CO}_2$  fluxes generally showed increased  $\text{CO}_2$  uptake, corresponding to higher  $\Delta C_{T \text{ flux}}$ , in surface waters south of the seasonal ice edge, and less  $\text{CO}_2$  uptake and thus lower  $\Delta C_{T \text{ flux}}$  in ice-covered waters over the slope and in the Nansen Basin (Fig. S3b – c). From average  $\Delta C_{T \text{ flux}}$  values across the Atlantic and Arctic regimes (Fig. S3), surface waters were an annual  $\text{CO}_2$  sink. Higher  $\Delta C_{T \text{ flux+res}}$  values in all seasons occurred beneath the sea ice, implying input of  $\text{CO}_2$  from other processes, e.g. from sea-ice brines, and that uncertainties in the other terms likely increased in ice-influenced waters.

Summer (August 2019) exhibited the largest  $\Delta C_T$  and  $\Delta A_T$  that coincided with the sea-ice cover. The main drivers were salinity changes (56 % of  $\Delta C_T$ ; 83 % of  $\Delta A_T$ ) and primary production (37 % of  $\Delta C_T$ ; 9 % of  $\Delta A_T$ ) following the seasonal ice melt and growing season. Highest  $\Delta C_T$  of  $-13.1 \text{ mol m}^{-2}$  and highest  $\Delta A_T$  of  $-8.5 \text{ mol m}^{-2}$  occurred over the central shelf (around P5). Sea-ice meltwater diluted  $A_T$  and  $C_T$  in the with  $\Delta A_{T \text{ sal}}$  of  $-8.2 \text{ mol m}^{-2}$  and  $\Delta C_{T \text{ sal}}$  of  $-8.9 \text{ mol m}^{-2}$  over the shelf (P5). Primary production was the main driver south of the ice edge, with  $\Delta C_{T \text{ bio}}$  of  $-4.3 \text{ mol m}^{-2}$  ( $\Delta A_{T \text{ bio}}$  of  $0.7 \text{ mol m}^{-2}$ ) south of the Polar Front (P1). Formation/dissolution of biotic and/or abiotic  $\text{CaCO}_3$  accounted for on average 3 % of  $\Delta C_T$  and 10 % of  $\Delta A_T$ , with  $\Delta A_{T \text{ CaCO}_3}$  up to  $1.4 \text{ mol m}^{-2}$  ( $\Delta C_{T \text{ CaCO}_3}$  of  $0.7 \text{ mol m}^{-2}$ ) beneath the sea ice. Changes due to air-sea  $\text{CO}_2$  fluxes and residual changes accounted for  $\sim 5$  % of  $\Delta C_T$  and varied from loss of  $C_T$  with  $\Delta C_{T \text{ flux+res}}$  of  $-1.4 \text{ mol m}^{-2}$  over the slope to inputs of  $C_T$  with  $\Delta C_{T \text{ flux+res}}$  of  $2.3 \text{ mol m}^{-2}$  over the shelf. The  $\Delta f\text{CO}_2$  values exhibited strong seasonality with substantial  $\text{CO}_2$  undersaturation in surface waters with  $\Delta f\text{CO}_2$  around  $-250 \mu\text{atm}$  in summer (Fig. S3). Ice-scaled  $\Delta C_{T \text{ flux}}$  showed maximum seasonal atmospheric  $\text{CO}_2$  uptake of  $0.7$ – $0.9 \text{ mol m}^{-2}$  in ice-free surface waters (Fig. S3b – c).

By winter (December 2019), changes due to salinity change

remained the main driver (45 % of  $\Delta C_T$ ; 78 % of  $\Delta A_T$ ) with an imprint of summertime primary production (31 % of  $\Delta C_T$ ; 9 % of  $\Delta A_T$ ) being eroded with increasing respiration/remineralisation (Fig. 10). A shift to positive  $\Delta C_T$  and a net increase in  $C_T$  of  $0.2 \text{ mol m}^{-2}$  occurred just north of the Polar Front. Salinity-driven changes were diminished relative to summer, with  $\Delta A_{T \text{ sal}}$  of  $-4.9 \text{ mol m}^{-2}$  and  $\Delta C_{T \text{ sal}}$  of  $-5.3 \text{ mol m}^{-2}$  in the Nansen Basin. Respiration reduced the summer  $C_T$ -uptake signal with  $\Delta C_{T \text{ bio}}$  of around  $-2.6 \text{ mol m}^{-2}$  in the Nansen Basin. The contribution from  $\text{CaCO}_3$  processes accounted for 3 % of  $\Delta C_T$  and 13 % of  $\Delta A_T$  with  $\Delta A_{T \text{ CaCO}_3}$  between  $-0.5 \text{ mol m}^{-2}$  to  $0.2 \text{ mol m}^{-2}$  over the shelf. Wintertime  $\Delta C_{T \text{ flux+res}}$  showed a general shift to net increases in  $C_T$  accounting for up to 21 % with highest  $\Delta C_{T \text{ flux+res}}$  of  $1.9 \text{ mol m}^{-2}$ . Ice-scaled  $\Delta C_{T \text{ flux}}$  showed atmospheric  $\text{CO}_2$  uptake up to  $0.5 \text{ mol m}^{-2}$  in surface waters south of the ice edge (Fig. S3).

In winter/spring (March 2021), salinity changes (39 % of  $\Delta C_T$ ; 85 % of  $\Delta A_T$ ) was the dominant driver (Fig. 10) The well mixed water column had seasonally low  $\Delta C_T$  and  $\Delta A_T$  of  $-1.5 \text{ mol m}^{-2}$  and  $-2.1 \text{ mol m}^{-2}$ , respectively, and slight positive  $\Delta C_T$  and  $\Delta A_T$  over the central shelf.  $\Delta C_{T \text{ sal}}$  and  $\Delta A_{T \text{ sal}}$  were more variable with  $\Delta C_{T \text{ sal}}$  ranging between  $-1.8$  and  $0.0 \text{ mol m}^{-2}$  and  $\Delta A_{T \text{ sal}}$  ranging between  $-1.6$  and  $0.0 \text{ mol m}^{-2}$ . Largest  $\Delta C_{T \text{ sal}}$  and  $\Delta A_{T \text{ sal}}$  occurred with varying sea-ice cover in the slope and Nansen Basin.  $\Delta C_{T \text{ bio}}$  was seasonally low ranging between  $-1.2$  and  $0.0 \text{ mol m}^{-2}$  (23 % of  $\Delta C_T$ ). The  $\Delta A_{T \text{ CaCO}_3}$  was seasonally low (6 % of  $\Delta A_T$ ) ranging between  $-0.7 \text{ mol m}^{-2}$  over the shelf and  $0.4 \text{ mol m}^{-2}$  in the Nansen Basin. By late spring (April–May 2021), the main driver was salinity changes (37 % of  $\Delta C_T$ ; 84 % of  $\Delta A_T$ ) alongside contributions from primary production (31 % of  $\Delta C_T$ ; 12 % of  $\Delta A_T$ ). Similar to winter/spring, smaller seasonal  $\Delta C_T$  of  $-1.9 \text{ mol m}^{-2}$  and  $\Delta A_T$  of  $-1.7 \text{ mol m}^{-2}$  occurred with positive  $\Delta C_T$  and  $\Delta A_T$  over the central shelf. The ranges of  $\Delta C_{T \text{ sal}}$  and  $\Delta A_{T \text{ sal}}$  increased, with greatest changes with  $\Delta C_{T \text{ sal}}$  of  $-2.0$  and  $\Delta A_{T \text{ sal}}$  of  $-2.1 \text{ mol m}^{-2}$  in the Nansen Basin. The onset of the spring bloom decreased  $C_T$  with  $\Delta C_{T \text{ bio}}$  of  $-1.3 \text{ mol m}^{-2}$  in the central shelf. The  $\Delta A_{T \text{ CaCO}_3}$  remained seasonally low (1 % of  $\Delta C_T$ ; 5 % of  $\Delta A_T$ ), ranging between  $-0.5 \text{ mol m}^{-2}$  as a signal of  $\text{CaCO}_3$  formation and  $0.6 \text{ mol m}^{-2}$  as a signal of  $\text{CaCO}_3$  dissolution over the central shelf. In winter/spring and spring, the  $\Delta C_{T \text{ flux+res}}$  ranged between  $-0.3$  and  $1.7 \text{ mol m}^{-2}$  and accounted for 32–37 % of  $\Delta C_T$ . The  $\Delta f\text{CO}_2$  values were minimum around  $-50 \mu\text{atm}$  in winter/spring and ice-scaled  $\Delta C_{T \text{ flux}}$  showed atmospheric  $\text{CO}_2$  uptake in surface waters south of the ice edge was up to  $0.5$ – $0.6 \text{ mol m}^{-2}$  (Fig. S3).

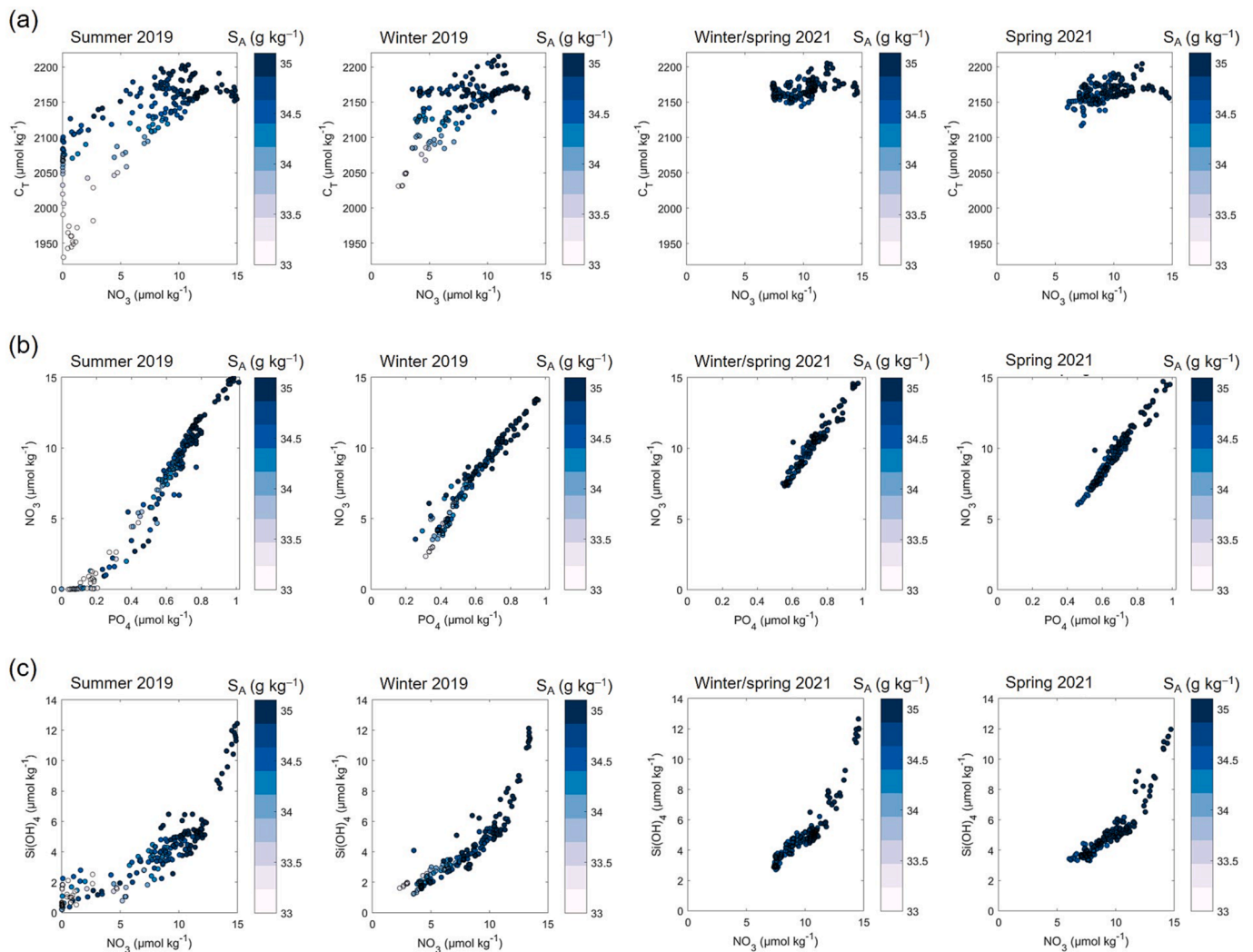
### 3.5. Net community production

Estimates of NCP from changes in  $\text{NO}_3$  in the upper 50 m from winter/spring to spring (NCP<sub>spr</sub>) showed that some biological carbon uptake had occurred by the spring but higher NCP from winter/spring to summer (NCP<sub>sum</sub>) occurred across the Atlantic and Arctic regimes (Table 4). There was a general northward decrease in NCP<sub>sum</sub> from  $1.3 \text{ mol C m}^{-2}$  ( $14 \text{ g C m}^{-2}$ ) at P1 to  $0.4 \text{ mol C m}^{-2}$  ( $5 \text{ g C m}^{-2}$ ) at P7. The highest NCP<sub>spr</sub> was  $0.3 \text{ mol C m}^{-2}$  at P6.

### 3.6. Interannual variability

With a one-year gap between the seasonal cruises in 2019 and 2021, year-to-year variations were superimposed onto the seasonal changes in the northwestern Barents Sea. The Barents Sea shelf and slope, Atlantic Water inflow and adjoining Nansen Basin exhibit strong interannual variability in sea ice and hydrographic conditions (Lundsgaard et al., 2022; Renner et al., this issue).

A strong seasonal cycle in sea-ice cover and temperature in the surface layer and in deeper waters in the southern part of the shelf (P2) persisted throughout 2019–2021 (Van Engeland et al., 2023). The variability in the surface layer in the central region (P4) showed a relatively regular seasonal pattern throughout 2019–2021, however year-to-year variations were evident in deeper waters during this period (Van Engeland et al., 2023) that resulted in shifts between cold IW and



**Fig. 8.** Relationships between (a)  $C_T$  ( $\mu\text{mol kg}^{-1}$ ) and  $\text{NO}_3$  ( $\mu\text{mol kg}^{-1}$ ) C:N, (b)  $\text{NO}_3$  ( $\mu\text{mol kg}^{-1}$ ) and  $\text{PO}_4$  ( $\mu\text{mol kg}^{-1}$ ) C:P, and (c)  $\text{Si(OH)}_4$  ( $\mu\text{mol kg}^{-1}$ ) and  $\text{NO}_3$  ( $\mu\text{mol kg}^{-1}$ ) Si:N, during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021). Data are coloured by salinity ( $S_A$ ;  $\text{g kg}^{-1}$ ).

CBSDW (summer to winter 2019) to relatively warm wPW (winter/spring to spring 2021). The changes in water mass composition overlying the shelf impacted the carbonate system, particularly  $C_T$ , and showed that interannual variability was stronger than seasonal variability (Table 2). Across the shelf slope (P6), year-to-year variability was strong in the surface layer with higher sea-ice cover throughout 2019 and early 2020, shifting to less sea ice and warmer waters in the summer seasons of 2020 and 2021. A more regular seasonal pattern dominated the variability in deeper waters (Van Engeland et al., 2023). Therefore, the interannual variability introduced as a result of the sampling gap between 2019 and 2021 is likely to exert a greater impact on the perceived seasonality for waters most influenced by ice-ocean interactions, i.e. the surface layer and water column of the marginal ice zone (north of P2) across the Barents Sea shelf and slope.

To assess the impact of these different temporal scales on the seasonal dynamics, the biogeochemical conditions in the surface layer and at  $\sim 120$  m depth of the northwestern Barents Sea ( $77\text{--}81^\circ\text{N}$ ,  $34^\circ\text{E}$ ; from P2 to P5) were evaluated relative to a 5-year average (Table S1) (Jones et al., 2020; Jones et al., 2019; Jones et al., 2022c; Skjelvan et al., 2021) during consecutive summer seasons from 2018 to 2022 (one year either side of the study period). The summer season represented the greatest changes in total change in  $C_T$  and  $A_T$  (Fig. 10) and therefore can provide a good assessment of interannual variability in the upper water

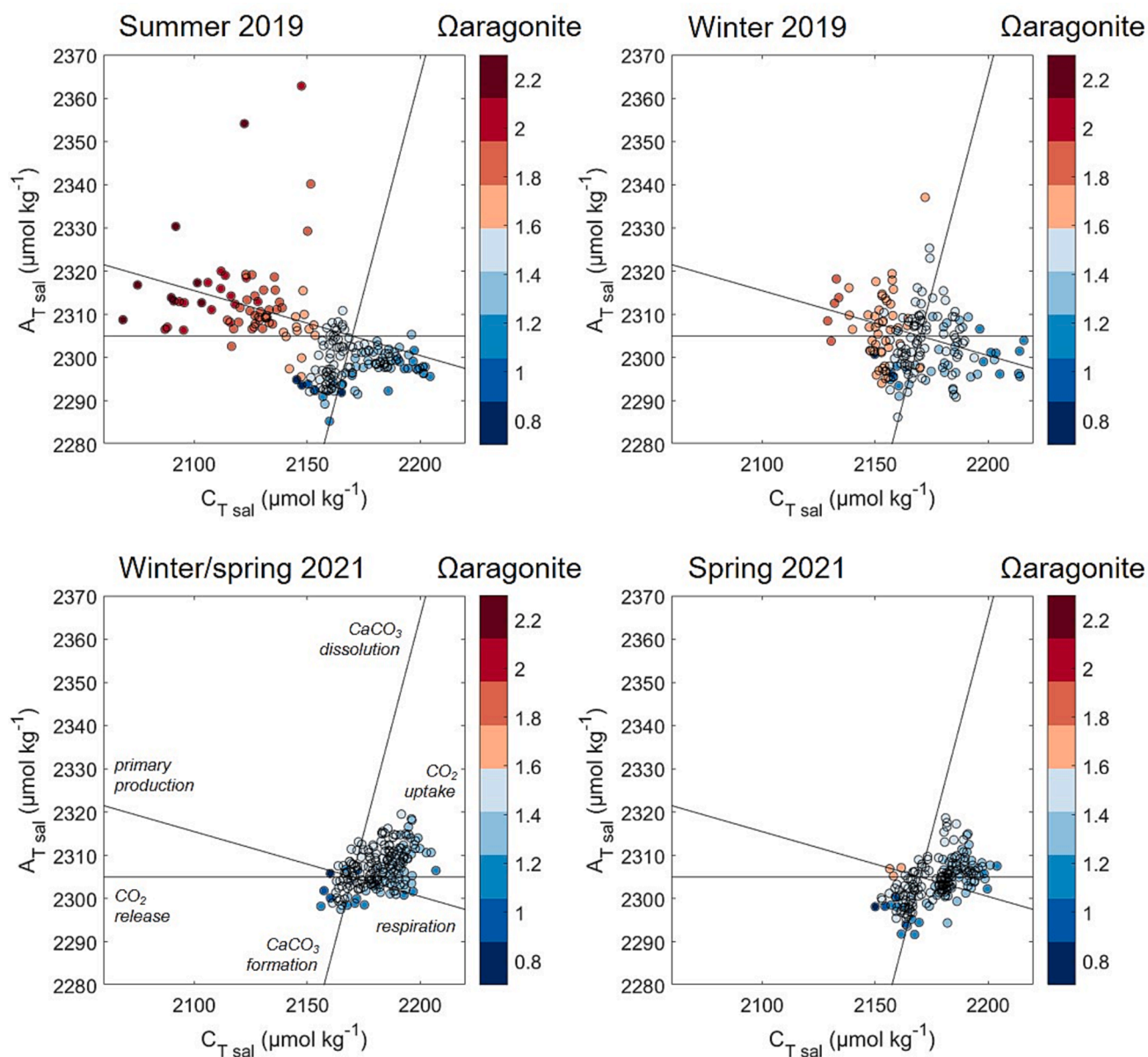
column and concurrent impacts on  $\Delta C_T$  and  $\Delta A_T$  that yield uncertainties in single-year estimates of  $\Delta C_T$  and  $\Delta A_T$ . Relative to the 5-year average  $\Delta C_T$  ( $\Delta C_{T\text{ ave}}$ ; average surface – average 120 m), the difference in average  $\Delta C_T$  in 2019 was  $8 \mu\text{mol kg}^{-1}$  and yielded an estimated error on the 2019 total  $\Delta C_T$  of 9 % due to interannual variability. The corresponding difference in  $\Delta A_T$  (relative to  $\Delta A_{T\text{ ave}}$ ) was  $2 \mu\text{mol kg}^{-1}$  and gives an estimated error on the 2019 total  $\Delta A_T$  of 4 % due to interannual variability. The  $\Delta C_T$  and  $\Delta A_T$  determined in summer 2019, and NCP estimates relative to winter/spring and spring 2021 (section 3.6), are conservative with respect to the 5-year average from this region and can be considered as largely representative with regards to interannual variability in the northwestern Barents Sea.

## 4. Discussion

### 4.1. Summer to winter transition

#### 4.1.1. Sea-ice meltwater and primary production drive reductions in carbon and nutrients in summer

Seasonal changes in  $C_T$ ,  $A_T$  and inorganic nutrients in the upper layer of the Barents Sea were strongly influenced by meltwater released during the seasonal retreat of the ice pack and biological uptake following the progression of the growing season. Greatest seasonal

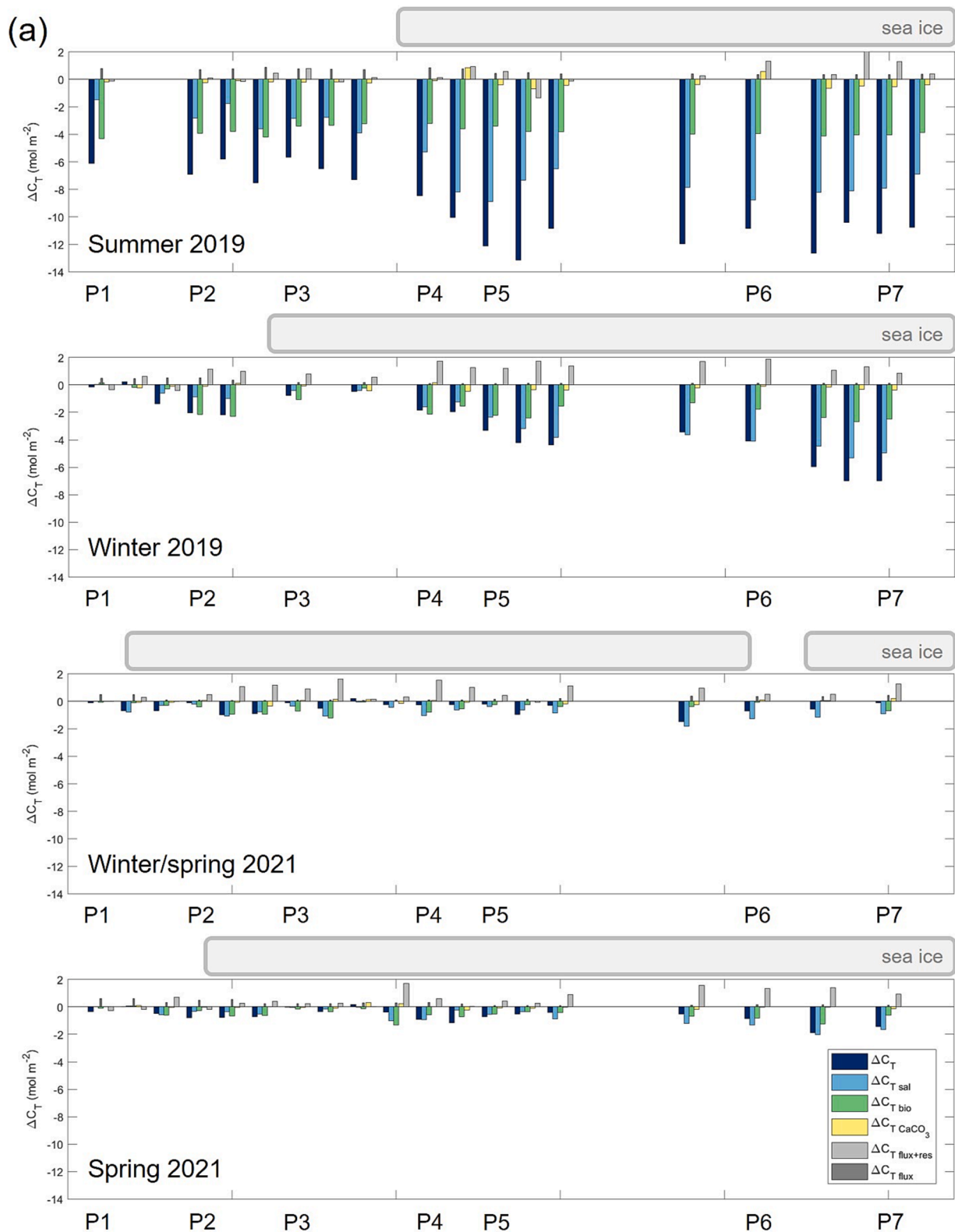


**Fig. 9.** Relationship between salinity-normalised  $A_T$  and  $C_T$  ( $A_{T\text{ sal}}$  and  $C_{T\text{ sal}}$ ;  $\mu\text{mol kg}^{-1}$ ), following Friis et al. (2003), for (a) summer (August 2019), (b) winter (December 2019), (c) winter/spring (March 2021) and (d) spring (April–May 2021) coloured according to the saturation state of aragonite ( $\Omega_{\text{aragonite}}$ ). Trend lines showing the theoretical slope for changes in  $A_{T\text{ sal}}$  and  $C_{T\text{ sal}}$  due to primary production/respiration,  $\text{CaCO}_3$  dissolution/formation, and  $\text{CO}_2$  uptake/release are shown from an arbitrary point in the  $A_{T\text{ sal}}-C_{T\text{ sal}}$  space.

reductions in  $C_T$  and nutrients occurred in the relatively warm, fresh and ice-free waters with near-depleted  $\text{NO}_3$  and low  $\text{Si}(\text{OH})_4$  in the surface layer (Fig. 11). Sea ice inflows from the Arctic Ocean and local sea ice formation and melt cycles contribute meltwater that forms the main freshwater source to the Barents Sea and wider Atlantic Water inflow region (Renner et al., 2018; Jones et al., 2021; Lundesgaard et al., 2021; Lundesgaard et al., 2022). Lowest  $C_T$  and  $A_T$  in the fresher Polar Water yielded largest changes in  $C_T$  and  $A_T$  in the upper 120 m with  $\Delta C_T$  of  $-13.1 \text{ mol m}^{-2}$  ( $232 \mu\text{mol kg}^{-1}$ ) and  $\Delta A_T$  of  $-8.5 \text{ mol m}^{-2}$  ( $150 \mu\text{mol kg}^{-1}$ ) driven by meltwater inputs that accounted for 55 % and 81 % of the summer  $\Delta C_T$  and  $\Delta A_T$ , respectively. The resultant elevated  $A_T:C_T$  and  $\Omega_{\text{aragonite}}$  in surface waters (Fig. 11g–h) were strongly driven by biological  $C_T$  uptake in the ice-free and meltwater-stabilised water column, which slightly counteracted effects of  $A_T$  dilution, resulting in biologically-driven increases in  $\Omega_{\text{aragonite}}$  and a buffer against acidification in summer surface waters.

In the wake of the retreating sea ice, meltwater inputs created a shallow meltwater lens that strengthens stratification in the surface layer (Leu et al., 2011; Koenig et al., this issue b). Surface waters were exposed increased light levels with a longer productive period and likely represented later-stage bloom conditions (in August) with contributions from regenerated production (Randelhoff et al., 2018; Henley et al., 2020). Increased biological uptake of  $C_T$  and nutrients was the major driver with seasonal maxima in  $\Delta C_{T\text{ bio}}$  of  $-4.3 \text{ mol m}^{-2}$  ( $\Delta A_{T\text{ bio}}$  of  $0.7 \text{ mol m}^{-2}$ ) in Atlantic-regimes south of the Polar Front. Surface waters were substantially undersaturated in  $\text{CO}_2$  relative to the atmosphere and with  $\Delta C_{T\text{ flux}}$  of  $0.3\text{--}0.9 \text{ mol m}^{-2}$  the region was a strong sink for atmospheric  $\text{CO}_2$ , which was also shown for the wider Barents Sea region in summer (Ericson et al., 2023).

Higher C:N uptake ratios ( $>10$ ) in summer were indicative of ongoing  $C_T$  uptake under low  $\text{NO}_3$  conditions in meltwater-influenced surface waters. The cycling of  $C_T$ ,  $\text{NO}_3$  and  $\text{PO}_4$  became un-coupled



**Fig. 10.** Seasonal drivers of changes in (a)  $C_T$  ( $\Delta C_T$ ;  $\text{mol m}^{-2}$ ) and (b)  $A_T$  ( $\Delta A_T$ ;  $\text{mol m}^{-2}$ ) in the upper ocean (surface to 120 m) during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021). Total change ( $\Delta C_T$ ,  $\Delta A_T$   $\text{mol m}^{-2}$ ; dark blue bars) is partitioned into different processes of salinity changes ( $\Delta C_{T \text{ sal}}$ ,  $\Delta A_{T \text{ sal}}$ ;  $\text{mol m}^{-2}$ ; light blue bars), primary production/respiration ( $\Delta C_{T \text{ bio}}$ ,  $\Delta A_{T \text{ bio}}$ ;  $\text{mol m}^{-2}$ ; green bars),  $\text{CaCO}_3$  formation/dissolution ( $\Delta C_{T \text{ CaCO}_3}$ ,  $\Delta A_{T \text{ CaCO}_3}$ ;  $\text{mol m}^{-2}$ ; yellow bars),  $\text{CO}_2$  fluxes and residual changes ( $\Delta C_{T \text{ flux+res}}$ ;  $\text{mol m}^{-2}$ ; light grey bars) and air-sea  $\text{CO}_2$  fluxes in the upper 10 m ( $\Delta C_{T \text{ flux}}$ ;  $\text{mol m}^{-2}$ ; dark grey lines). Positive  $\Delta C_T$  and  $\Delta A_T$  indicate an increase in  $C_T$  and  $A_T$ , respectively, in the surface layer relative to 120 m depth; negative  $\Delta C_T$  and  $\Delta A_T$  indicate a decrease in  $C_T$  and  $A_T$ , respectively, in the surface layer relative to 120 m depth. Seasonal sea-ice cover is indicated by horizontal grey bars. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)



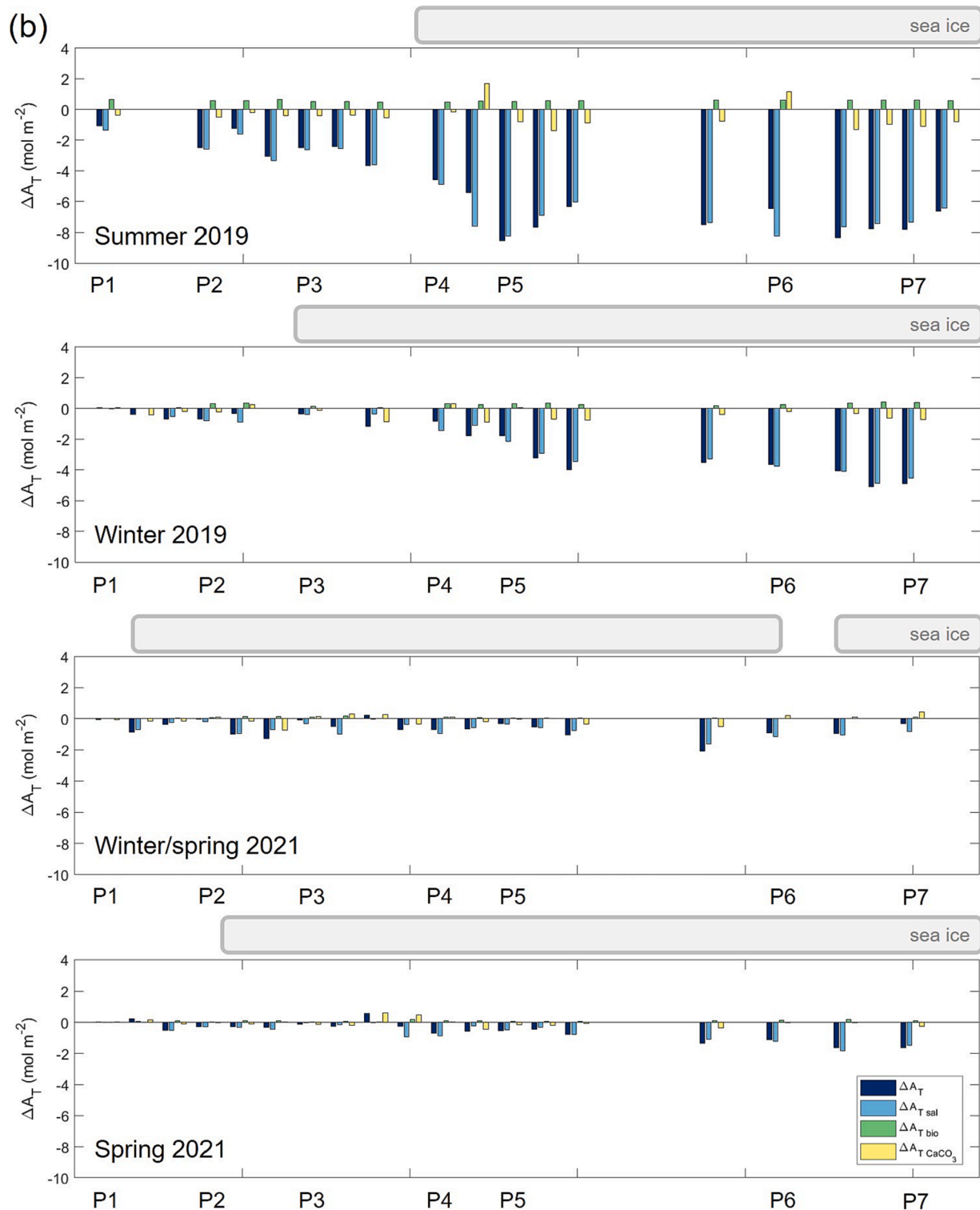


Fig. 10. (continued).

resulting in overconsumption of  $C_T$  at depleted  $\text{NO}_3$  concentrations following the growing season. This occurred typically in open waters during the progression of summer with a likelihood of subsurface chlorophyll *a* maxima appearing as surface nutrients get depleted (Koenig et al., this issue b). High C:N uptake ratios and overconsumption of  $C_T$  relative to  $\text{NO}_3$  has also been reported for the summer in the Atlantic Water inflow (Randelhoff et al., 2018), across the northern Svalbard shelf (Jones et al., 2021) and in Svalbard fjords (Ericson et al., 2019). Surface water  $C_T$  can also be affected by ice-algal production (Leu

et al., 2011), which may contribute to production and  $C_T$  uptake signals in the pelagic blooms in the MIZ. Concurrent lower Si:N uptake ratios ( $\sim 0.5$ ) reflected  $\text{Si}(\text{OH})_4$  uptake that followed the species succession from spring to late summer. Concentrations of  $\text{Si}(\text{OH})_4$  were near or at depletion as a result of  $\text{Si}(\text{OH})_4$  uptake in diatom blooms following depletion of  $\text{NO}_3$  in the surface layer due to earlier uptake by non-siliceous species, such as the haptophyte *Phaeocystis pouchetii* with (Reigstad et al., 2002; Assmy et al., 2017; Henley et al., 2020).

The northward decrease in  $\text{NCP}_{\text{sum}}$  from highest seasonal values of

**Table 4**

Net community production (NCP; mol C m<sup>-2</sup>) estimated from changes in salinity-normalised nitrate concentrations ( $\Delta\text{NO}_3_{\text{sal}}$ ;  $\mu\text{mol kg}^{-2}$ ) from winter/spring (March 2021) to spring (April–May 2021) NCP<sub>spr</sub>, and from winter/spring (March 2021) to summer (August 2019) NCP<sub>sum</sub>.  $\Delta\text{NO}_3_{\text{sal}}$  was integrated in the upper 50 m and converted to carbon using C:N stoichiometric ratio 106:16 (Redfield et al., 1963).

| Station        | Regime/ location                               | NCP <sub>spr</sub>             | NCP <sub>sum</sub>       |
|----------------|--|--------------------------------|--------------------------|
|                |  | (mol C m <sup>-2</sup> )       | (mol C m <sup>-2</sup> ) |
|                |  | <i>March to April–<br/>May</i> | <i>March to August</i>   |
| P1             | Atlantic Water inflow/ south of Polar Front    | 0.1                            | 1.3                      |
| P2             | Arctic/ Barents Sea shelf                      | 0.0                            | 1.0                      |
| P3             | Arctic/ Barents Sea shelf                      | 0.0                            | 0.8                      |
| P4             | Arctic/ Barents Sea shelf                      | 0.2                            | 0.9                      |
| P5             | Arctic/ Barents Sea shelf                      | 0.1                            | 0.6                      |
| P6             | Barents Sea shelf slope/ Atlantic Water inflow | 0.3                            | 0.6                      |
| P7             | Nansen Basin                                   | 0.0                            | 0.4                      |
| <i>average</i> |  | <i>0.1 ± 0.1</i>               | <i>0.8 ± 0.3</i>         |

1.3 mol C m<sup>-2</sup> (14 g C m<sup>-2</sup>) in the Atlantic regime at P1 to 0.4 mol C m<sup>-2</sup> (5 g C m<sup>-2</sup>) in the Nansen Basin at P7 followed the seasonal retreat of the sea ice and the comparative increased primary production in ice-free waters. Using the same nitrate-derived carbon uptake approach, spring NCP (January to May) of 26 g C m<sup>-2</sup> and summer NCP (January to August) of 36 g C m<sup>-2</sup> were determined along the Atlantic Water inflow in eastern Fram Strait (Chierici et al., 2019), which are close to the range 28–32 g C m<sup>-2</sup> estimated for annual export production in the Barents Sea (Fransson et al., 2001). These estimates are approximately 50 % larger than the highest NCP<sub>sum</sub> of 14 g C m<sup>-2</sup> in the Atlantic Water inflow from this study, which is likely due time period of seasonal biological production (March taken as wintertime pre-bloom conditions versus January as in Chierici et al., 2019) and the more northerly location with seasonal sea-ice cover delaying primary production. Furthermore, the NCP<sub>sum</sub> estimates determined from changes in NO<sub>3</sub>, referred to as new production, do not consider biological production using other inorganic nitrogen sources (Chierici et al., 2019) and reveal the range of NCP estimates for the Arctic Ocean region depending on the approach used and the temporal and spatial scales selected. The NCP<sub>spr</sub> and NCP<sub>sum</sub> values corresponded to daily rate of change since winter/spring (March) of approximately  $20 \pm 27 \text{ mg C m}^{-2} \text{ d}^{-1}$  by spring (April–May) and  $64 \pm 24 \text{ mg C m}^{-2} \text{ d}^{-1}$  by summer (August) in the upper 50 m. The NCP<sub>spr</sub> and NCP<sub>sum</sub> estimates of organic matter production from biological carbon (NO<sub>3</sub>) uptake align with the general trend of organic carbon export rates in the upper 30–200 m, with high and spatially variable fluxes ( $178 \pm 202 \text{ mg C m}^{-2} \text{ d}^{-1}$ ) from diatom blooms in May and less variable fluxes ( $159 \pm 79 \text{ mg C m}^{-2} \text{ d}^{-1}$ ) with greater contribution from small flagellates in August (Bodur et al., 2023).

Decreased A<sub>T</sub> up to 1.4 mol m<sup>-2</sup> in the upper layer ascribed to CaCO<sub>3</sub> processes suggested formation of CaCO<sub>3</sub>, which could result from biotic CaCO<sub>3</sub> formation by calcifying organisms or precipitation of ikaite from sea ice (Fransson et al., 2013). The changes in A<sub>T</sub> indicated that CaCO<sub>3</sub> processes were seasonally important in decreasing A<sub>T</sub> (accounting for 10 % of  $\Delta A_T$ ) in the surface layer in summer. The formation of CaCO<sub>3</sub> shells and skeletons in waters of the Barents Sea occurs by phytoplankton and zooplankton both locally and/or signals of CaCO<sub>3</sub> production advected in the Atlantic Water inflow (Chierici et al., 2019; Oziel et al., 2020; Jones et al., 2021). The calcifying phytoplankton *Emiliania huxleyi* has calcitic shells (coccoliths) and has expanded its geographic range northwards in the Atlantic Water inflow and has been found to maintain growth under nutrient limited conditions in the Barents Sea (Hegseth and Sundfjord, 2008; Oziel et al., 2020). Additionally, the calcifying zooplankton pteropods (aragonitic shells) and foraminifera and have been found with highest seasonal abundances in summer

(Zamelczyk et al., 2021; Anglada-Ortiz et al., 2023). Pteropods and foraminifera are key producers of CaCO<sub>3</sub>, contributing to A<sub>T</sub> and C<sub>T</sub> changes through calcification or CaCO<sub>3</sub> dissolution, and seasonal carbon export and fluxes in the Barents Sea (Anglada-Ortiz et al., 2023). Another possible mechanism of reduced A<sub>T</sub> from CaCO<sub>3</sub> processes in seasonally-ice covered waters is from ikaite formation in sea ice (Rysgaard et al., 2007). Meltwater release may transfer a signal of ikaite formation to surrounding surface waters (Rysgaard et al., 2007; Fransson et al., 2013; Chierici et al., 2019).

#### 4.1.2. Winter surface waters imprinted by meltwater dilution and biological processes

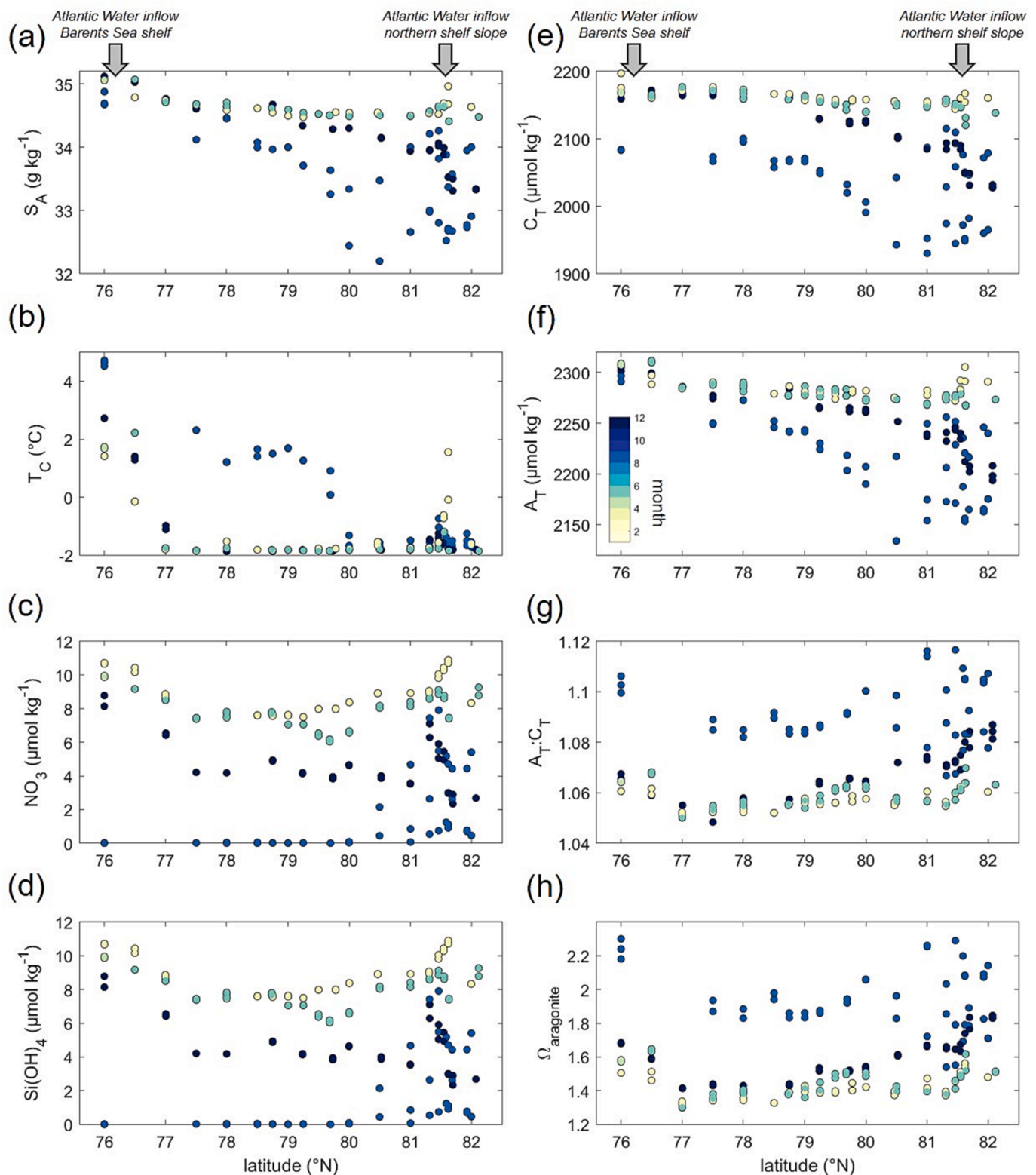
Increased remineralisation of organic matter and mixing from late summer to early winter shifted the conditions in the Barents Sea from substantial biological carbon uptake and a strong atmospheric CO<sub>2</sub> sink towards areas of net increases in C<sub>T</sub> ( $\Delta C_T$  up to 0.2 mol m<sup>-2</sup>) in the upper layer. Atmospheric cooling of surface waters increased the extent of Polar Water and the pack ice expanded southward during the transition to winter. The southern part of the region was ice-free (Fig. 2) for a longer period and showed smaller seasonal changes in C<sub>T</sub> and A<sub>T</sub> from vertical mixing into subsurface Atlantic Water (and Atlantic-influenced waters). The biologically-driven decreases in C<sub>T</sub> and NO<sub>3</sub> at the end of the growing season had been eroded and wind-driven mixing dispersed the lower C<sub>T</sub> and NO<sub>3</sub> signature and lowered C:N and N:P ratios deeper into the water column. Replenishment of C<sub>T</sub> and NO<sub>3</sub> occurred in the upper layers (Fig. 5 – 6) from subsurface waters alongside inputs from organic matter remineralisation. This was most notable at southern locations, with higher C<sub>T</sub> (and A<sub>T</sub>) in the surface layer south of the ice edge that contributed to variability in NO<sub>3</sub> and Si(OH)<sub>4</sub> with lower concentrations below 100 m depth (Fig. 6). Negating effects of inputs of A<sub>T</sub> from intrusions of Atlantic-influenced waters, these processes reduced the A<sub>T</sub>:C<sub>T</sub> ratio and  $\Omega_{\text{aragonite}}$  in the upper layer. Since the preceding summer, nutrient-recycling and remineralisation (Randelhoff et al., 2018; Henley et al., 2020) and advective fluxes from Atlantic Water (Koenig et al., this issue a) could re-supply NO<sub>3</sub> to alleviate any nutrient limitation in the presence of late-stage phytoplankton communities (Randelhoff et al., 2015; Slagstad et al., 2015; Tremblay et al., 2015) to prolong C<sub>T</sub> uptake through the autumn and contribute to a wintertime biological-uptake signal with  $\Delta C_{T \text{ bio}}$  around  $-2.7 \text{ mol m}^{-2}$  in the Nansen Basin.

The fresher Polar Water surface layer in Arctic regime in the Nansen Basin reflected effects of relative freshening in the upper layers, whilst vertical mixing increased the salinity, C<sub>T</sub> and A<sub>T</sub> in the water column to counteract the effects of dilution south of the winter ice edge. Decreases in A<sub>T</sub> up to 0.9 mol m<sup>-2</sup> around the ice edge could be attributed to CaCO<sub>3</sub> precipitation, which follows the lowered-A<sub>T</sub> signal of biotic calcification and/or abiotic ikaite formation from the summer. A decoupling between C<sub>T</sub> and A<sub>T</sub> variability was evident in the water column as C<sub>T</sub> had greater increases than A<sub>T</sub> (and salinity), suggesting different driving processes or sources of C<sub>T</sub> and A<sub>T</sub> during the summer to winter transition. The relative increase in C<sub>T sal</sub> at near-constant A<sub>T sal</sub> indicated that atmospheric CO<sub>2</sub> uptake in the cooling surface water had a key role in increasing C<sub>T</sub> since the summer (Fig. 9). The uptake of atmospheric CO<sub>2</sub> that was driven by primary production was slightly counteracted by mixing with subsurface waters and late-stage blooms, with an average ice-scaled  $\Delta C_{T \text{ flux}}$  of 0.2 mol m<sup>-2</sup> dampened from an average summertime  $\Delta C_{T \text{ flux}}$  of 0.6 mol m<sup>-2</sup> (Fig. S3) across the region. Concomitant thermodynamic reductions in surface water  $\Omega_{\text{aragonite}}$  also occur during winter cooling, alongside increased C<sub>T</sub> from CO<sub>2</sub> inputs (Fig. 11), as the solubility of aragonite increases with decreasing temperature (Mucci, 1983).

#### 4.2. Winter to spring transition

##### 4.2.1. Mixing, organic matter remineralisation and sea-ice processes enrich carbon and nutrients below the ice pack

Vertical mixing, remineralisation of organic matter and CO<sub>2</sub> fluxes



**Fig. 11.** Surface water (<15 m) (a)  $S_A$  ( $\text{g kg}^{-1}$ ), (b)  $T_C$  ( $^{\circ}\text{C}$ ), (c)  $\text{NO}_3$  ( $\mu\text{mol kg}^{-1}$ ), (d)  $\text{Si}(\text{OH})_4$  ( $\mu\text{mol kg}^{-1}$ ), (e)  $C_T$  ( $\mu\text{mol kg}^{-1}$ ), (f)  $A_T$  ( $\mu\text{mol kg}^{-1}$ ), (g) the  $A_T:C_T$  ratio and (h)  $\Omega_{\text{aragonite}}$  along the transect. Data are colour coded by month of sampling with the legend in (f). The general location of the Atlantic Water inflow in the southern Barents Sea shelf and northern shelf slope is marked by grey arrows.

enriched the water column with  $C_T$ ,  $A_T$  and nutrients beneath the sea ice by winter/spring (March). This resulted in highest concentrations of  $C_T$ ,  $A_T$  and  $\text{NO}_3$  in the Polar Water layer and highest  $C_T$ ,  $\text{NO}_3$  and  $\text{Si}(\text{OH})_4$  in Atlantic-influenced water masses across the region out of all seasons. Mixing of Atlantic Water and replenishment of  $C_T$  and  $\text{NO}_3$  was

particularly evident along the inflow pathways (Fig. 5). The well mixed water column had low  $\Delta C_T$  and  $\Delta A_T$  with lowest variability and small range out of all the seasons. The water column was more homogenous and mixing with subsurface Atlantic Water (salinity-driven changes) was a key driver accounting for 39 % and 85 % of  $\Delta C_T$  and  $\Delta A_T$ ,

respectively. Small changes as a result of meltwater inputs ( $\Delta C_{T\text{ sal}}$  and  $\Delta A_{T\text{ sal}}$ ) occurred with localised erosion of the sea-ice cover through shoaling Atlantic-derived water masses over the slope and in the Nansen Basin. Deep mixed layers and low chlorophyll *a* concentrations (Koenig et al., this issue b) reflected the low biological activity and resulted in small contributions from biological and  $\text{CaCO}_3$  processes. This coincided with lowest seasonal abundances of the calcifying zooplankton foraminifera and pteropods that were found in winter/spring (Anglada-Ortiz et al., 2023).

The cooled surface waters and increased  $C_{T\text{ sal}}$  indicated that atmospheric  $\text{CO}_2$  uptake was also driving changes in  $C_T$  prior to substantial sea-ice cover, with  $C_T$  increases in the surface layer (ice-scaled  $\Delta C_{T\text{ flux}}$ ) of  $0.5\text{ mol m}^{-2}$  that contributed to the greatest seasonal change due to  $\Delta C_{T\text{ flux+res}}$  (37 % of  $\Delta C_T$ ) with average  $\Delta C_{T\text{ flux+res}}$  values of  $0.7\text{ mol m}^{-2}$ . This likely included inputs of  $\text{CO}_2$  from the rejection of  $\text{CO}_2$ -rich brines from forming sea ice where ikaite formation has taken place and would partition  $A_T$  in the sea ice and enrich  $C_T$  in the brine (Rysgaard et al., 2007). These processes enabled  $C_T$  to become enriched in the upper layers and drive thermodynamic reductions in  $\Omega_{\text{aragonite}}$  and lowering the  $A_T:C_T$  ratio in the water column. Relative to the other seasons, lowest C:N and highest N:P ratios in winter/spring resulted from organic matter remineralisation and mixing with Atlantic Water that enriched the water column with  $\text{NO}_3$  relative to  $C_T$ . Lower C:N and higher N:P have been previously reported for Atlantic-influenced waters in the Barents Sea region (Frigstad et al., 2014; Jones et al., 2021). Concurrent higher Si:N ratios resulted from dissolution of diatom frustules and/or mixing with  $\text{Si(OH)}_4$ -rich Intermediate Water over the shelf. Warm Polar Water occupied the deep layer of the shelf, rather than CBSDW that was found in summer and winter 2019 (discussed further in section 4.3) and represented a nutrient (and  $A_T$ ) source to overlying Polar Water if vertical mixing processes occurred. This showed inter-annual variability in ice-ocean processes influencing water mass structure in the deeper water column across the shelf.

#### 4.2.2. Spring blooms, sea-ice melt and seasonal reductions in carbon and nutrients

The general northward retreat of the sea ice and degraded sea ice along the Atlantic Water inflow (Fig. 2) exposed the nutrient-rich surface layer to increasing light levels and meltwater stabilisation in spring (April-May). Smaller seasonal changes in  $\Delta C_T$  of  $-1.9\text{ mol m}^{-2}$  and  $\Delta A_T$  of  $-1.7\text{ mol m}^{-2}$  reflected conditions of the well-mixed column with relatively little change since early spring (March). The high  $C_T$  and  $A_T$  content in the upper and Atlantic-influenced layers was pre-conditioned by vertical mixing, organic matter remineralisation,  $\text{CO}_2$  fluxes and inputs from  $\text{CO}_2$ -rich brines since the winter. The variability in  $C_T$  and  $A_T$  closely followed that of the early spring with values similar to those at the end of winter as a starting point for the changes in concentration that followed.

Excess  $A_T$  in the surface layer relative to subsurface waters indicated a surface source of  $A_T$  ( $\Delta A_{T\text{ CaCO}_3}$  up to  $0.6\text{ mol m}^{-2}$ ; only accounting for 5 % of average  $\Delta A_T$ ). Sources of  $\text{CaCO}_3$  include shells and skeletons of phytoplankton and zooplankton in the water column (Anglada-Ortiz et al., 2023) or advection in the Atlantic Water inflow (Chierici et al., 2019; Oziel et al., 2020). Another possible source of  $A_T$  and  $C_T$  from  $\text{CaCO}_3$  in seasonally-ice covered waters is from the dissolution of ikaite released from sea ice (Rysgaard et al., 2007) during early ice melt in spring. Potential minor contributions of  $A_T$  could also result from glacial meltwater originating from Svalbard (Ericson et al., 2019), river runoff from Svalbard and Arctic rivers (Cooper et al., 2008) and precipitation, which is expected to be insignificant compared to contributions from sea-ice (local and advected) meltwater (Koenig et al., this issue b).

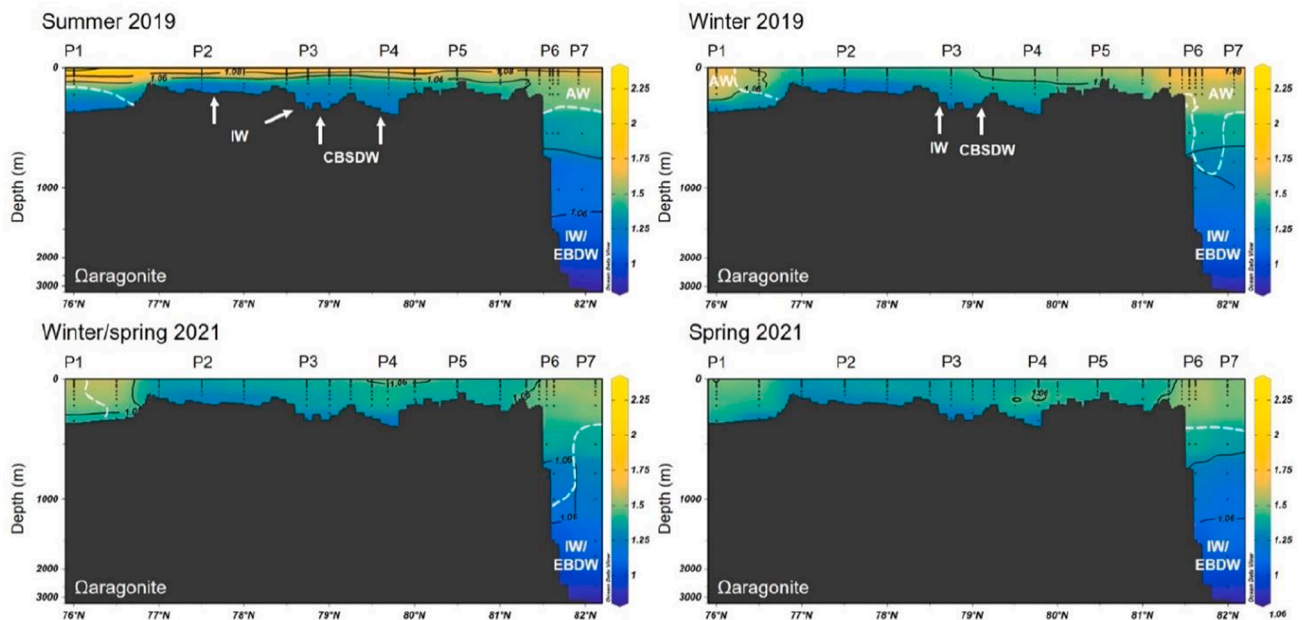
Nutrient and  $C_T$  concentrations in surface waters showed a slight northward decrease relative to earlier spring (March) concentrations indicating uptake (prior to significant dilution) in the Polar Water over the central shelf. Pelagic phytoplankton in Arctic waters typically start to bloom following significant melt and retreat of the seasonal ice pack

to increase light availability (Reigstad et al., 2002; Leu et al., 2011). The onset of seasonal sea-ice melt in spring and favourable light levels was sufficient to promote phytoplankton production and an early spring bloom despite deep mixed layers in ice-covered waters (Koenig et al., this issue b). Additionally, nutrients, phytoplankton and upstream blooms could be advected along the Atlantic Water inflow to enhance the spring nutrient and  $C_T$  uptake signal despite the presence of seasonal sea ice (Assmy et al., 2017; Henley et al., 2020). This resulted in a reduction in surface layer concentrations of  $\text{NO}_3$ ,  $C_T$  and  $A_T$  compared with winter/spring values. Dilution in the upper layer lowered  $C_T$  and  $A_T$ , particularly over the northern shelf and Nansen Basin. Increased C:N ratios and decreased N:P ratios traced reductions in  $C_T$  following uptake of  $\text{NO}_3$ , consistent with nutrient and carbon stoichiometry of biological uptake. Reductions in  $C_T$  in the upper layer ( $\Delta C_{T\text{ bio}}$ ) up to  $-1.3\text{ mol m}^{-2}$  increased during the seasonal transition, enhancing the  $A_T:C_T$  ratio and biologically-driven increases in  $\Omega_{\text{aragonite}}$ . Springtime NCP estimates were more spatially variable than the NCP estimates in summer with the highest value of  $0.3\text{ mol C m}^{-2}$  ( $4\text{ g C m}^{-2}$ ) in the Atlantic Water inflow across the northern shelf slope. The  $\text{NCP}_{\text{spr}}$  was about 25 % of the maximum summer NCP and was located in partially ice-covered waters and may have contributions from ice algae production (Leu et al., 2011). Higher NCP of  $16 \pm 6\text{ g C m}^{-2}$  and  $15 \pm 5\text{ g C m}^{-2}$  was previously derived from biological carbon uptake and equivalent nitrate-derived carbon consumption, respectively, in the upper 50 m further west in the Nansen Basin during the spring bloom period in May to June (Assmy et al., 2017; Fransson et al., 2017). The almost four-fold higher values in the west in the Nansen Basin likely result from later stage spring bloom and more nitrate uptake.

#### 4.3. Acidification of intermediate and deep waters in the Barents sea and Nansen Basin

In summer and winter 2019, localised deepening of the isolines over the shelf indicated the presence of cold ( $T_c \leq -1.1\text{ }^\circ\text{C}$ ) and dense CBSDW between  $78.7$  and  $80^\circ\text{N}$  and cool Intermediate Water at  $77.5$  and  $78.7^\circ\text{N}$  (Fig. 12). Both water masses occupied a greater area in summer and persisted into winter (Fig. 4). From winter to spring 2021, no Intermediate Water or CBSDW was found over the shelf, instead warm Polar Water occupied this depth range (about 100–300 m) below a water column dominated by Polar Water. The CBSDW showed slightly reduced  $\text{NO}_3$  and similar  $\text{Si(OH)}_4$  and  $A_T$  ( $\sim 2300\text{ }\mu\text{mol kg}^{-1}$ ) content relative to the warm Polar Water that occupied this depth and location from winter to spring. A notable difference was elevated  $C_T \sim 2190\text{ }\mu\text{mol kg}^{-1}$  with  $\Omega_{\text{aragonite}} \sim 1.2$  and lowest  $A_T:C_T$  ratio of 1.05 (Fig. 12) and the least buffered waters in the Barents Sea. The similar  $A_T$  content could be considered as a semi-conservative tracer to suggest that any meltwater dilution was largely counteracted by advected supply of high  $A_T$  water. Therefore, the elevated  $C_T$  showed that other (non-biological, non-mixing/advective) processes contributed to enrichment of  $C_T$  relative to  $\text{NO}_3$  and  $A_T$  in the CBSDW.

Taking the hydrographic conditions to represent both seasonal and interannual variability (2019 vs. 2021; as also reported in Van Engeland et al. (2023)), the changes in water mass structure that followed indicated some key processes influencing carbonate chemistry (Table S1) (Jones et al., 2020; Jones et al., 2019; Jones et al., 2022c; Skjelvan et al., 2021), nutrient cycling and acidification state over the shelf. The 2018/19 winter period had strong surface cooling, sea-ice formation and deep convection, resulting in a wider expanse of the ice pack and the production of cold, saline Intermediate Water and CBSDW (Lundsgaard et al., 2022; Van Engeland et al., 2023). The Intermediate Water and CBSDW present over the shelf in summer and early winter 2019 were remnants of the cooling, deep convection and enhanced sea-ice formation of the preceding 2018/19 winter period. In 2019, sea-ice inflows into the Barents Sea covered a substantially wider areal extent and contributed more freshwater to the region (Lundsgaard et al., 2022; Renner et al., this issue). The elevated  $C_T$  and salinity of the CBSDW



**Fig. 12.** Transect section plots of  $\Omega$ aragonite during summer (August 2019), winter (December 2019), winter/spring (March 2021) and spring (April–May 2021). Contours are shown for the  $A_T:C_T$  ratio (thin black lines) and absolute salinity of  $35.06 \text{ g kg}^{-1}$  (dashed white line) that distinguishes modified Atlantic Water and Atlantic Water. Water masses Atlantic Water (AW), Cold Barents Sea Dense Water (CBSDW), Intermediate Water (IW) and Eurasian Basin Deep Water (IW/EBDW) are labelled. The locations of the process (P) stations are labelled. Ocean Data View was used for data visualisation (Schlitzer, 2022).

reflected the impact of deep wintertime convection and ice-ocean processes where  $\text{CO}_2$ -rich brines expelled from forming sea ice enriched subsurface water with  $C_T$  under the winter sea ice (Fransson et al., 2013). Following the period of cooling and convection, the southward expanding ice pack capped the water column to reduce vertical mixing and warm Polar Water was found encroaching the central shelf from the Atlantic Water inflow over the northern shelf. The acidified CBSDW layer became largely diminished showing seasonal variability.

In contrast, the 2020/21 winter period had less surface cooling and reduced convection, limiting production of cold and saline subsurface waters over the shelf (Van Engeland et al., 2023). Instead, warm Polar Water was found overlying the shelf, from recirculated Atlantic Water that enters the region from the northern inflow and is subsequently modified by cooling and freshening by mixing with Arctic waters (Renner et al., 2018). Modifications of water masses are likely reflected in the  $A_T$ ,  $C_T$  and nutrient content through a balance of organic matter remineralisation and silicic acid and carbonate dissolution versus dilution processes. The Barents Sea and Atlantic Water inflow shelves experiences significant interannual variability in sea ice (Renner et al., this issue) and winters of less sea ice imports, reduced cooling and convection can therefore lead to increased presence of Atlantic-influenced waters (Van Engeland et al., 2023) that represent an additional source of nutrients and  $A_T$  to pre-condition the water column over the central shelf for the following growing and meltwater seasons.

In the Nansen Basin, IW/EBDW occupied the intermediate and deep layers (below 1000 m deep) in all seasons (Fig. 4). Intermediate Water was relatively nutrient rich, especially with respect to  $\text{Si}(\text{OH})_4$ , with slightly lower  $A_T$  ( $\sim 2302 \mu\text{mol kg}^{-1}$ ) and slightly higher  $C_T$  ( $\sim 2169 \mu\text{mol kg}^{-1}$ ) compared with Atlantic Water. The average water mass characteristics showed some variability in  $C_T$ ,  $A_T$  and nutrients, and a low  $A_T:C_T$  ratio  $\sim 1.06$ . Values of  $\Omega$ aragonite were  $0.76\text{--}1.22$  at  $1000\text{--}3650 \text{ m}$  and the saturation horizon for aragonite ( $\Omega$ aragonite = 1) was located at  $1500\text{--}2000 \text{ m}$  depth in the Nansen Basin (Fig. 12). Reductions in  $\Omega$ aragonite of  $0.05\text{--}0.14$  during 1994–2011 (Ericson et al., 2014) and  $0.04\text{--}0.06$  during 1996–2015 (Ulfsbo et al., 2018) have been found in Atlantic Water and Intermediate Water layers ( $\sim 100\text{--}1500 \text{ m}$ ) in the Arctic Ocean. For initial  $\Omega$ aragonite values of  $1.2\text{--}1.4$  (Ericson

et al., 2014), the corresponding rate of change in  $\Omega$ aragonite was therefore  $-0.002$  and  $-0.003$  per year in the Intermediate Water layer. In the Nansen Basin at  $\sim 82^\circ\text{N}$ , Jones et al. (2021) reported  $\Omega$ aragonite of  $1.13 \pm 0.02$  in Intermediate Water at  $1500 \text{ m}$  in 2015 and this study found  $\Omega$ aragonite of  $1.11 \pm 0.03$  at  $1500 \text{ m}$  (at P7,  $\sim 82^\circ\text{N}$ ) for the years 2019 to 2021. The Atlantic Water inflow into the Arctic Ocean transports anthropogenic  $\text{CO}_2$ , from atmospheric uptake in the Atlantic Ocean, and inputs from organic matter remineralisation into Intermediate Waters of the Nansen Basin. The reported decline in  $\Omega$ aragonite Intermediate Waters results from the accumulation of  $C_T$  at intermediate depths that are isolated from the surface ocean, therefore limiting connectivity and ventilation in the Arctic Ocean (Ericson et al., 2014; Ulfsbo et al., 2018). Following initial  $\Omega$ aragonite conditions from Ericson et al. (2014), it is estimated that  $\Omega$ aragonite in Intermediate Water in the Nansen Basin has decreased by  $\sim 0.07$  in 20 years (Jones et al., 2021) and  $\sim 0.09$  in about 25 years (this study). This corresponds to a change in  $\Omega$ aragonite of  $-0.003$  and  $-0.004$  per year, respectively, and indicates that the rate of decrease of  $\Omega$ aragonite (enhanced risk of acidification) has slightly increased across the different study periods since 1994. Therefore, the shallower limit ( $\sim 1500 \text{ m}$ ) of Intermediate Water in the Nansen Basin could experience acidification with undersaturation of aragonite in the next 50–100 years.

#### 4.4. Atlantic Water connectivity alleviates surface water acidification along the inflow pathways in the Barents sea

A more Atlantic-like regime along the inflow pathways south of the Polar Front and in the upper  $\sim 500 \text{ m}$  across the slope and into the Nansen Basin shaped the biogeochemical characteristics and drivers of  $C_T$  and  $A_T$  in the upper water column (Fig. 4). Both regions of Atlantic Water inflow had highest  $A_T:C_T$  ratios and  $\Omega$ aragonite in surface waters in summer and elevated  $\Omega$ aragonite in the upper layer in the other seasons relative to other waters across the Barents Sea shelf (Fig. 12). The Atlantic Water inflow south of the Polar Front generally displayed the lowest seasonality in inorganic carbon and nutrient cycling. Seasonally ice-free, wind-induced mixing maintained a more homogeneous water mass structure during each season (Fig. 3) with small

changes in  $C_T$  and  $A_T$  in the upper water column relative to the rest of the Barents Sea shelf and Nansen Basin (Fig. 11). Primary production was the biggest driver of seasonal  $C_T$  change and increased NCP to a summer maximum of  $1.3 \text{ mol C m}^{-2}$  relative to the rest of the region with concurrent high standing stocks of chlorophyll *a* (Koenig et al., this issue b) and nutrient resupply (Koenig et al., this issue a) within a persistent Atlantic-like regime.

The Atlantic Water inflow over the northern shelf slope displayed strong seasonality in  $C_T$  and nutrients and some of the largest changes in  $C_T$  and  $A_T$  in the upper water column (through reduced  $C_T$  and  $A_T$  in the surface layer) relative to the rest of the Barents Sea. Seasonally covered in sea ice with varying conditions (Renner et al., this issue), reduced wind-driven mixing sustained a more stable water column typically capped with Polar Water. Shoaling of Atlantic-influenced waters into the surface layer over the slope and Nansen Basin likely contributed to degradation of the sea ice in March 2021 (Fig. 2). This region exhibits variable ice conditions (Renner et al., this issue) and is sometimes ice-free from winter to early spring due to Atlantic Water intruding the surface, with subsequent heat loss and melting of the seasonal sea ice (Onarheim et al., 2014; Duarte et al., 2020). Ice-free waters and subsequent wind-driven mixing can allow nutrients to enrich the upper layer (Randelhoff and Sundfjord, 2018), which was evident by increased temperature, salinity and  $\text{NO}_3$  in the surface waters. This was accompanied by localised increases in  $A_T$ , elevated  $A_T:C_T$  ratio and higher  $\Omega$  aragonite. In addition, springtime NCP was highest ( $0.3 \text{ mol C m}^{-2}$ ) relative to the rest of the region, fuelled by favourable light levels in patchy sea-ice cover and nutrient influx from below, in addition to advection of nutrients and phytoplankton along the inflow. Atlantic Water ventilation and connectivity along the northern shelf induced a transient Atlantic-like regime within the Arctic regime. The region is known to reflect signals from ice-ocean and biogeochemical processes that take place upstream in the inflow (Randelhoff et al., 2018; Renner et al., 2018; Lundesgaard et al., 2021) and shows that advective fluxes have an important influence on inorganic carbon and nutrient cycling and the state of ocean acidification downstream on the Barents Sea shelf and slope (Jones et al., 2021; Renner et al., this issue).

The Atlantic Water inflow pathways in the Barents Sea exhibited enhanced  $A_T$  from advection,  $\text{CaCO}_3$  dissolution and biological production (increases in  $\Omega$  aragonite). The concomitant increases in  $C_T$  (decreases in  $\Omega$  aragonite) from the carbon-rich Atlantic Water likely resulted in transient offsets to the physical, geochemical and biological mechanisms driving increases in  $\Omega$  aragonite. However, in the presence of active phytoplankton communities and favourable light levels, the resupply of  $\text{NO}_3$  facilitates  $C_T$  uptake and yields net increases in  $\Omega$  aragonite. These synergistic processes resulted in (i) generally elevated nutrients (except for depleted  $\text{NO}_3$  in summer),  $C_T$  and  $A_T$  and low seasonality in the ice-free, mixed water column south of the Polar Front and (ii) increased nutrients,  $C_T$  and  $A_T$  and high seasonality in the ice-influenced water column over the shelf slope and Nansen Basin. Atlantic-derived  $A_T$  supplied to productive surface waters enhanced the buffering capacity (increased  $A_T$  and decreased  $C_T$ ), resulting in seasonally and spatially elevated  $\Omega$  aragonite as a mechanism to counteract ocean acidification in the Atlantic Water inflow regions of the Barents Sea.

## 5. Conclusion

Variability in inorganic carbon and nutrient cycling in the Barents Sea was strongly coupled to sea-ice cover and meltwater release, different water masses, biological production/remineralisation and  $\text{CaCO}_3$  processes, which exerted seasonal controls on the state of ocean acidification ( $\Omega$ ) across the Atlantic and Arctic regimes. Seasonally elevated  $\Omega$  ( $\Omega$  aragonite up to 2.30) in the surface layer occurred along the Atlantic Water inflow pathways from connectivity to Atlantic-influenced waters that increased  $A_T$  alongside biological  $C_T$  uptake. Nitrate-derived NCP in the upper 50 m was enhanced in spring ( $0.3$

$\text{mol C m}^{-2}$ ) and summer ( $1.3 \text{ mol C m}^{-2}$ ) in the Atlantic-like regimes.

In summer,  $C_T$  and  $A_T$  changes were driven by meltwater dilution (55 % of  $\Delta C_T$ ; 81 % of  $\Delta A_T$ ) and biological carbon uptake during primary production (37 % of  $\Delta C_T$ ). This resulted in enhanced surface water  $\Omega$  south of the ice edge. Winter cooling, deep convection and sea-ice processes resulted in lower  $A_T$  and elevated  $C_T$ , driving low  $\Omega$  ( $\Omega$  aragonite 1.12–1.14) and risk of acidification in Arctic-like regimes over the shelf. From winter to early spring,  $C_T$  increases were explained by a combination of vertical mixing, remineralisation of organic matter, atmospheric  $\text{CO}_2$  uptake and release of  $\text{CO}_2$ -rich sea ice brines. The water column was replenished with inorganic nutrients prior to the following growing season. The onset of the spring blooms and ice melt initiated the meltwater- and biologically-driven reductions in  $A_T$  and  $C_T$  and uptake of nitrate in surface waters. Formation and dissolution of biotic and abiotic  $\text{CaCO}_3$  contributed to 5–13 % of  $\Delta A_T$ . This is the first study investigating a full seasonal cycle of  $\text{CO}_2$  fluxes in the marginal ice zone of the Barents Sea. Surface waters across Atlantic and Arctic regimes acted as a net annual sink of atmospheric  $\text{CO}_2$ . The intermediate and deep waters in the Nansen Basin revealed ongoing acidification trends from inflowing carbon-rich Atlantic-influenced waters. Understanding the seasonal dynamics and drivers of inorganic carbon and nutrient cycling in the Barents Sea is essential to understand the state of ocean acidification and determine the future impacts of Atlantification, which can have cascading consequences for the ecosystem along the Atlantic Water inflow and into the Arctic Ocean.

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## Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Data availability

Data are reported in Chierici et al. (2021a), Chierici et al. (2021b), Gerland (2022), Jones et al. (2022a), Jones et al. (2022b), Ludvigsen (2022), Reigstad (2022) and Søreide (2022) and are available from the Norwegian Marine Data Centre.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.pocean.2023.103131>.

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