1	Complete validation according to current international criteria of a confirmatory
2	quantitative method for the determination of nitrofuran metabolites in seafood by liquid
3	chromatography isotope dilution tandem mass spectrometry
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19 Abstract

Despite the ban of nitrofurans (NFs) for use in food production in many countries in the 20 1990s, NF metabolites in food are still regularly detected during import control testing. We 21 22 have developed a confirmatory routine method for the detection and quantification of NF 23 metabolites in seafood using LC-MS/MS and validated the method according to the strict criteria in European legislation and Codex Alimentarius. Method characteristics were found to 24 25 fulfill the criteria. We report for the first time a new false positive for 1-amino-2,4imidazolidinedione (AHD), the metabolite of Nitrofurantoin (NFT). By using optimized 26 washing procedures, the non tissue bound false positives can be minimized. The results from 27 28 the validation on both lean and fatty fish and crustaceans, results from proficiency tests and routine use over many years, demonstrates that the method is fit for purpose to determine NF 29 metabolites in the seafood category. 30

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32 Keywords: nitrofurans; LC-MS/MS; seafood; false positive

1. Introduction

During World War II it was discovered that Schiff base derivatives of nitrofuraldehyde 35 were effective against pathogenic bacteria, leading to extensive research and synthesis of such 36 37 compounds (McCalla, 1983). Nitrofurans (NFs) are comprised of a Sciff base of a nitrofuraldehyde and are part of the large family of hydrazide compounds characterized by the 38 RC(=O)NRNR₂ functional group (Moss, Smith, & Tavernier, 1995). NFs are broad-spectrum 39 synthetic antibiotics, effective against both Gram-negative and Gram-positive bacteria. Under 40 the World Health Organisation (WHO) Anatomical Therapeutic Chemical (ATC) 41 classification system they are categorization both as antiinfectives for systemic use (J01XE), 42 dermatologicals - antiseptics and disinfectants (D08AF) and antiparasitic products, 43 insecticides and repellents - antiprotozoals - agents against leishmaniasis and trypanosomiasis 44 45 (P01CC) (WHO). One of the advantages of using NFs is that resistance appears to develop slowly towards this class of antibiotics. NFs have been used as feed additives for growth 46 promotion in domesticated animals and have been used both prophylactically and 47 therapeutically to treat gastrointestinal infections, such as bacterial enteritis caused by 48 Escherichia coli and Salmonella. 49

The main NFs used in animal production are furazolidone (FZD, 3-{(E)-[(5-nitro-2-50 furyl)methylene]amino}-1,3-oxazolidin-2-one), furaltadone (FTD, 5-(4-morpholinylmethyl)-51 3-{(E)-[(E)-(5-nitro-2-furyl)methylene]amino}-1,3-oxazolidin-2-one), nitrofurazone (NFZ, 52 (2E)-2-[(5-nitro-2-furyl)methylene]hydrazinecarboxamide) and nitrofurantoin (NFT, 1-{(E)-53 [(5-nitro-2-furyl)methylene]amino}-2,4-imidazolidinedione). Upon ingestion, these NFs are 54 metabolized within minutes to their respective NF metabolites; 3-amino-1,3-oxazolidin-2-one 55 (AOZ), 3-amino-5-(4-morpholinylmethyl)-1,3-oxazolidin-2-one (AMOZ), 56 hydrazinecarboxamide (usually referred to as semicarbazide (SEM)) and 1-amino-2,4-57

58 imidazolidinedione (usually referred to as 1-aminohydantoin (AHD)). These NF metabolites

are more persistent than the parent NFs and can prevail for weeks after treatment, covalentlybound to muscle protein in treated animals.

In 1991 the U.S. Food and Drug Administration (FDA) withdrew the approvals for 61 62 FZD and NFZ, and in 1995 the use of NFs in livestock production was banned in the EU due to concerns regarding the carcinogenic and mutagenic properties of NFs and potential harmful 63 effects on human health (EC, 1993, 1995). Following their ban, a minimum required 64 performance level (MRPL) was established for methods to be used in official control of NF 65 metabolites at 1 µg/kg in the EU. The MRPL is implemented in the analytical method as the 66 decision limit (CC α) which is the limit at and above it can be concluded with an error 67 probability of α that a sample is non-compliant. The implications of this is that any detections 68 of NF metabolites, above the $CC\alpha$ of the confirmatory method used for the determination, is 69 70 evidence of illegal use of NFs. Codex Alimentarius Commission (CAC) has developed risk management recommendations for FZD and nitrofural (another name for NFZ) to prevent 71 72 residues of these NFs in food (CAC, 2018).

73 The number of detections of NF metabolites has decreased since the peak years at the beginning of the millennium, possibly due to a combination of increased food import control 74 and decreased use. Residues of NFs are still regularly detected during EU import control of 75 aquaculture products such as *e.g.* fish and crustaceans as reported in the Rapid Alert System 76 for Food and Feed (RASFF) (RASFF). The main NF metabolites detected are SEM and AOZ, 77 whereas AMOZ was more prominent in the early 2000s compared to present days. For SEM 78 there is a known problem of false positives in foods due to the use of additives in both food 79 80 packaging and food, which will result in identical analytical response (de la Calle & Anklam, 2005). 81

82 While early methods focused on the determination of the parent NFs, the lack of83 stability of the parent NFs has forced a change of analytical methodology focus towards

methods which determine NF metabolites. Because NF metabolites have high polarity and 84 85 low molecular mass the detection and quantification is usually carried out after derivatization with 2-nitrobenzaldehyde (2-NBA). Following derivatization of the metabolites, 3-[(E)-(2-86 Nitrobenzylidene)amino]-1,3-oxazolidin-2-one (NP-AOZ), 5-(4-Morpholinylmethyl)-3-[(E)-87 (2-nitrobenzylidene)amino]-1,3-oxazolidin-2-one (NP-AMOZ), (2E)-2-(2-88 Nitrobenzylidene)hydrazinecarboxamide (NP-SEM) and 3-[(E)-(2-Nitrobenzylidene)amino]-89 90 2,4-imidazolidinedione (NP-AHD) are obtained. Sample workup and derivatization of NF metabolites are usually carried out based on the published methods from Hoogenboom et. al. 91 and Leitner et. al. (Hoogenboom, Vankammen, Berghmans, Koeman, & Kuiper, 1991; 92 93 Leitner, Zollner, & Lindner, 2001). 94 Different instrument setups for the detection and quantification of NF metabolites have been published using LC with either ultraviolet (UV) or mass spectrometry (MS) as 95 detectors (Conneely, Nugent, O'Keeffe, Mulder, van Rhijn, Kovacsics, et al., 2003; Hu, Xu, & 96

Yediler, 2007). Whereas UV is more cost beneficial, the use of MS is more selective and may
fulfill requirements for unequivocal identification of the metabolites. Both MS, MS/MS and
ion-trap MS have been used for the determination of NF metabolites. The use of MS/MS is
often preferred due to higher selectivity compared to single quadrupole MS and better
reproducibility compared to ion-trap-MS (Saari & Peltonen, 2004). Using two transitions for
the identification and quantification of NF metabolites in MS/MS, fulfills the point system for
unequivocal detection of an illegal substance (CAC, 2014; EC, 2002).

Several methods for the determination of various NF metabolites in foods such as egg,
chicken, pork, poultry, dairy products, feed, bakery products and baby food have been
developed (Vass, Hruska, & Franek, 2008). No method for the determination of all NF
metabolites demonstrated to be applicable to seafood as a food category has previously been
published, but methods for the LC-MS/MS determination of AOZ in prawns, SEM in

109	crayfish, prawns, shrimp and fish, FZD and AOZ in fish and AOZ, AMOZ, AHD and SEM in
110	shrimp has been published (Chu & Lopez, 2005; Conneely, et al., 2003; Hoenicke,
111	Gatermann, Hartig, Mandix, & Otte, 2004; Hu, Xu, & Yediler, 2007; Saari & Peltonen, 2004;
112	Valera-Tarifa, Plaza-Bolanos, Romero-Gonzalez, Martinez-Vidal, & Garrido-Frenich, 2013).
113	Important validation parameters such as <i>e.g.</i> recovery and/or $CC\alpha/CC\beta$ are also missing in
114	several of the methods published earlier. We hypothesized that it would be possible to
115	develop a horizontal method applicable to seafood as a food category and report here our
116	findings after comprehensive validation of a method for the determination of NF metabolites
117	in seafood, carried out according to the current EU decision 657/2002/EC and Codex
118	Guideline 71-2009 (CAC, 2014; EC, 2002).
119	
120	2. Material and methods
121	2.1 Samples
122	Samples of seafood were selected to represent both variety in matrix composition and seafood
123	where NF metabolites are often detected, such as crustaceans. Atlantic salmon (Salmo salar)
124	was used as a matrix representing fish with high fat content and atlantic cod (Gadus morhua)
125	represented fish with low fat content. Shrimp (Pandalus borealis) was used as a proxy for
126	crustaceans. Validation samples were obtained from regular monitoring samples collected

routinely by our institute and the Norwegian Food Safety Autority. Samples did not contain

128 NF metabolites. The wild cod and shrimp were collected by fishermen authorized for

129 commercial fishing by the Norwegian Directorate of Fisheries. Both wild cod, shrimp and

130 farmed salmon were treated according to Norwegian laws regulating handling and euthanasia

131 fish and seafood.

133 2.2 Standards

Unlabeled NF metabolites AOZ, AMOZ, AHD and SEM (VETRANAL, \geq 98% purity) were 134 purchased from Sigma Aldrich (St. Louis, MO). Labelled NF metabolites AOZ-D₄, AMOZ-135 D₅, AHD-¹³C₃ and SEM-¹⁵N₂¹³C were purchased from Witega (Berlin, DE). Stock solutions 136 used for establishing calibration curves and for spiking of control samples were prepared 137 138 independently. Stock solutions of unlabeled analytes used in calibration curves and control samples, and labeled analytes used as internal standards, were prepared by weighing 15-25 139 mg of the compounds. The weighed amount of unlabeled AMOZ and AOZ was transferred 140 into a 20 ml volumetric flask followed by diluting to the mark with methanol (Chromasolv, 141 142 Sigma Aldrich). Stock solution of unlabeled AHD was prepared by transfering the weighed amount into a 100 ml volumetric flask, followed by diluting to the mark with methanol. For 143 unlabeled SEM the weighed amount was transferred into a 250 ml volumetric flask, followed 144 by diluting to the mark with Milli Q water. The weighed amount of the labeled internal 145 standards were all transferred into 100 ml volumetric flasks and diluted to the mark with 146 147 methanol, except for SEM which was diluted with Milli Q water. All stock solutions were diluted to intermediate mixture solutions by adding an appropriate amount of each stock 148 solution to a 10 ml volumetric flask and diluting to the mark with methanol. Nominal 149 concentrations (in parenthesis) for intermediate mixture solution were; calibration curves (10 150 μ g/ml), internal standards (2-10 μ g/ml) and control solutions (4-6 μ g/ml). Working standard 151 solutions in the ng/ml range were prepared by adding an appropriate amount of the 152 intermediate solutions to 100 ml volumetric flasks and diluting to the mark with methanol. 153 154

155 2.3 Reagents

The extraction solution, 0.2 M hydrochloric acid (HCl), was prepared by adding 16.7 ml of 156 concentrated HCl (12 M, 37%, Merck, Darmstadt, DE) to 300 ml Milli Q water in a 1 L 157 volumetric flask followed by diluting to the mark with Milli Q water. Derivatization reagent 158 (100 mM) was prepared by weighing 76 mg 2-NBA (p.a., Sigma Aldrich) into a 5 ml 159 volumetric flask and diluting to the mark with methanol (Chromasolv, Sigma-Aldrich). The 2 160 M sodium hydroxide solution was prepared by weighing 20 g NaOH (Merck) in a 250 ml 161 162 volumetric flask, followed by dissolving the NaOH in Milli Q water and finally diluting to the mark. The 0.3 M trisodiumfosfate buffer solution was prepared by weighing 11.4 g 163 Na₃PO₄·12H₂O (Merck) into a 100 ml volumetric flask followed by diluting to the mark with 164 165 milli Q water. Mobile phase A for the LC was prepared by adding 1 ml concentrated acetic 166 acid (100%, glacial, anhydrous for analysis EMSURE®, Merck) to a 1 L mixture of 900 ml acetonitrile and 100 ml Milli Q water. Mobile phase B was prepared by adding 1 ml 167 168 concentrated acetic acid to 1 L of Milli Q water. In addition to the aforementioned Milli Q water and methanol, hexane (Merck) was used for sample cleanup and ethanol (Kemetyl, 169 Kolbotn, NO) and diethyl ether (Merck) was used for the washing procedures (2.4). End 170 solution was prepared by mixing 25 ml acetonitrile with 225 ml Milli Q water and adding 250 171 µl concentrated acetic acid. 172

173

174 **2.4 Washing procedure for tissue bound NF metabolites**

175 Two different procedures were used for determination of tissue bound residues of NF

176 metabolites. Both washing procedures were performed after weighing of the sample, but prior

to the sample preparation (supplementary Figure 1) described in section 2.5.

178 Washing procedure 1 was based on the report from the 2004 Joint FAO/WHO

179 Technical Workshop on Residues of Veterinary Drugs without ADI/MRL (FAO/WHO,

2004). The sample was first homogenised with 5 ml ethanol/water (1/1; v/v), followed by 5
min 4000 rpm centrifugation. The supernatant was then discarded. The sample was further
washed three times with 5 ml of methanol, two times with 5 ml of ethanol and two times with
5 ml of diethyl ether. Each time the sample was washed by addition of the solvent, turning the
sample tube three times, and decanting off the solvent before the next washing step.
Following washing and decanting of the last washing solvent, the procedure described under
sample preparation (2.5) was carried out.

Washing procedure 2 was based on the method used by the European Reference 187 Laboratory (EURL) for residues of veterinary medicines and contaminants in food of animal 188 189 origin (Eric Verdon, Couedor, & Sanders, 2007). In this procedure the sample was washed four times, first with 6 ml of methanol/water solution (1/1; v/v), then with 6 ml of 190 methanol/water solution (3/1; v/v), followed by 6 ml of methanol and finally 2 ml of Milli Q 191 water. Each time the addition of washing solvent was followed by a 15 min rotary 192 homogenization at 100 rpm, before centrifugation for 10 min at 4000 rpm and disposal of the 193 194 supernatant by decantation. Following the washing and decanting of the milli Q water, the procedure described in sample preparation (2.5) was carried out. 195

196

197 **2.5 Sample preparation**

Sample weighing for validation and later routine use was identical. Homogenized samples were weighed $(1.00\pm0.02 \text{ g})$ into polypropylene (PP)-tubes. It is important to make sure that the analytical test portion is sufficiently large and homogeneous to produce representative results of the sample. During validation matrix blank for positive controls were spiked with appropriate volumes of a 10 ng/ml working standard mixture of NF metabolites. During later routine use, when the CC α of the method had been established, the matrix blank for positive

control was spiked with 100 µl of a 4-6 ng/ml working standard mixture of NF metabolites. 204 205 Calibration curve solutions were prepared by adding 10-1000 µl of a 10 ng/ml working 206 standard mixture to PP-tubes. Calibration curves with 5-6 concentrations in the range 0.1-10 ng of each NF metabolite were prepared fresh from the stock solutions every day. All samples 207 and calibration curve solutions were then spiked with 100 µl of a 20-100 ng/ml internal 208 standard mixture solution. The derivatization agent, 2-NBA, (50 µl of the 100 mM solution) 209 210 was added to all samples, matrix blanks and calibration curve solutions . Then 5 ml of 0.2 M HCl were added to samples and matrix blanks and 1 ml 0.2 M HCl was added to calibration 211 curve solutions. A procedural blank was prepared by adding 50 µl of 2-NBA and 1 ml of 0.2 212 213 M HCl to an empty tube. All tubes were vortex mixed for 1 minute and placed in a heating cabinet at 37±3 °C overnight. The next morning samples were removed from the heating 214 cabinet and allowed to cool down to room temperature. Then 0.3 M Na₃PO₄ solution was 215 216 added to each sample and matrix blank (500 μ l) and calibration curve solution (100 μ l) followed by swirl mixing. The pH was adjusted to 7 ± 0.5 by adding 2 M NaOH to the 217 samples and matrix blanks (340-370 µl), the procedural blank (340-370 µl) and calibration 218 curve solutions (80 µl), followed by swirl mixing and waiting 5 minutes before the pH was 219 220 checked using pH-paper. The amount of 2 M NaOH required to adjust the pH to 7 was found 221 to be matrix dependent. Typical volumes of 2 M NaOH required to reach pH 7 was 340-360 µl for salmon, 350 µl for shrimp, 350-360 µl for cod and 370 µl for the procedural blank. 222 After the pH adjustment, 4 ml ethyl acetate was added to the samples and matrix blanks and 223 224 the content of the tubes were mixed for 20 minutes using a rotator. Samples and matrix blanks were then centrifuged at 4000 rpm for 5 minutes, and the ethyl acetate was transferred to a 225 226 new PP-tube. The extraction was repeated with 4 ml ethyl acetate and the combined extracts were filtered through a 0.2 µm nylon filter. The solvent was removed by evaporation at 30 °C 227 228 using a flow of nitrogen. The samples and matrix blanks were reconstituted in 250 µl of end

solution. Further clean up of the samples was done by adding 2 ml of hexane to the

reconstituted sample, vortex mixing and centrifugation at 4000 rpm for 5 minutes. The bottom

231 layer was transferred to vials for instrument determination.

232

233 **2.6 Instrument determination**

234 Samples were analysed using an Agilent 1200 LC, fitted with the large volume injecton kit 235 allowing for volumes up to 100 µl to be injected. The LC was connected to an Agilent 6410B 236 MS/MS fitted with a Zorbax Eclipse Plus C18 column (600 Bar), with particle size of 1.8 µm, 237 internal diameter 2.1 mm and length 150 mm. The instrument sequence was set up according to Commission decision 2002/657/EC with minor adjustments. The validation sequence 238 started with calibration curve solutions followed by procedural blanks, negative matrix 239 blanks, (spiked) samples and calibration curve solutions. The sequence used for later routine 240 followed the same setup except that the negative matrix blank and spiked sample at CCa were 241 242 also analysed after the samples in the sequence. Injection volumes were 50 µl for all solutions expect calibration curve solutions at or above 1 ng, where 20-5 µl injection volumes were 243 used. Flow rate was set at 0.25 ml/min and the column temperature was held at 45 °C. The 244 245 gradient started at 17 % A, which was held for 0.1 minutes before changing rapidly to 40% A in 0.01 min. This composition was held until 3.5 minutes. Then the composition was rapidly 246 returned to initial conditions during 0.01 min, which was held until 10 minutes. The MS was 247 operated in positive ESI MRM mode. Gas temperature was held at 350 °C, gas flow was held 248 249 at 6 L/min, nebulizer pressure was held at 15 psi and capillary was held at 3500 (V). Detailed 250 MS-parameters are given in Table 1.

251

252 2.7 Validation setup and calculation of validation results

253	Validation was carried out according to Codex guideline 71-2009 and EU Commission
254	decision 2002/657/EC which lays down the parameters to be checked in the validation and the
255	criteria for the evaluation of results and fitness for purpose of the method. The outline of the
256	validation is shown in Table 2.
257	Selectivity/specificity was evaluated by investigating chromatographic separation of
258	the analytes, comparing peak shape in standards and samples. Matrix effects which could give
259	interferences were evaluated by analyzing >20 blank samples of each matrix.
260	Linear range of the method was evaluated by analysis of calibration curves and by
261	determining the decision limit (CC α). The upper limit of quantification (ULOQ) was set equal
262	to the concentration of the highest calibration curve standard.
263	$CC\alpha$ was determined using the three methods described in Commission decision
264	2002/657/EC. When using the first method CC α was determined by analyzing > 20 blank
265	samples of each matrix (sample blank method) and using three times signal to noise (S/N)
266	given by the formula
267	
268	$CC\alpha = 3 * S/N$
269	
270	CCα was also determined using linear regression of the matrix blanks spiked at 1.0 x MRPL,
271	1.5 x MRPL and 2.0 x MRPL as stated in Commission decision 2002/657/EC (calibration
272	curve method). The CC α was determined by calculating > 20 calibration curves and
273	determining the actual mean zero value of analyte and adding 2.33 standard deviations (SD)
274	of the intercept as given in the formula
275	
276	$CC\alpha = zero \ value_{standard \ curve} + 2.33SD_{Intecept \ standard \ curve}$

The CCa was also determined using the ISO 11843 procedure referred to in Commission decision 2002/657/EC (ISO 11843 method) using the formula from Verdon et. al (E. Verdon, Hurtaud-Pessel, & Sanders, 2006).

282
$$CC\alpha = t_{(\alpha,IJ-2)}\frac{\hat{\sigma}}{\hat{b}}\sqrt{\frac{1}{K} + \frac{1}{IJ} + \frac{\bar{x}^2}{\sum(x_{ij} - \bar{x})^2}}$$

284	where t_{α} is the Student's t-value at risk α and IJ-2 degrees of freedom, $\hat{\sigma}$ is the estimated
285	residual standard deviation of the regression function, \hat{b} is the estimated slope of the
286	calibration curve, \bar{x} is the mean of the x _{ij} values, K is the number of replicates of the real
287	state, I is the number of calibration levels and J is the number of replicates per level.
288	Detection capability (CC β) was determined by analyzing > 20 matrix blanks spiked at
289	CCα and adding 1.64 SDs as specified in the formula
290	
291	$CC\beta = CC\alpha + 1.64 SD_{[CC\alpha]}$
292	
293	Ion ratios of quantifier and qualifier in standards and samples were determined every day.
294	Recovery was evaluated by spiking matrix blanks at 1.0 x MRPL, 1.5 x MRPL and 2.0
295	x MRPL and calculating the recovery using the formula
296	

297
$$Recovery(\%) = \frac{Measured \ content}{Fortification \ level} * 100\%$$

299	Since there were no certified reference materials available at the time of validation, the
300	trueness was evaluated by determining a proficiency test of shrimp containing SEM.
301	Following validation, the method has participated regularly in proficiency tests.
302	Precision, as repeatability, was determined by analyzing six replicates of salmon
303	muscle spiked at 1.0 x MRPL, 1.5 x MRPL and 2.0 x MRPL on three different days.
304	Intermediate precision as within-laboratory reproducibility was determined by analyzing two
305	replicates of each matrix spiked at 1.0 x MRPL, 1.5 x MRPL and 2.0 x MRPL on three
306	different days. Results from both the repeatability experiments and the intermediate precision
307	experiments were combined to calculate the within-laboratory reproducibility.
308	Measurement uncertainty (MU) was evaluated since this is an important requirement
309	of ISO 17025 and accreditation. MU was calculated using the relative standard deviation
310	(RSD) from the within-laboratory reproducibility using a coverage factor of two followed by
311	adding five percent points due to the limited traceability of the trueness, using the formula
312	$MU(\%) = 2 \ x \ RSD(\%) + 5(\%)$
313	
314	and finally rounding up to the nearest 5%.
315	Stability of the analytes in neat solutions and in matrix was determined by storing 10
316	standards at -20 °C in the dark, 10 standards at +4 °C in the dark, 10 standards at room
317	temperature in the dark and 10 standards at room temperature in daylight. In addition 5 blanks

of each matrix was spiked at 5 ng and stored at -20 °C in the dark. One standard and one

spiked matrix blank from each storage condition was analyzed after one day, one week, two

weeks etc. until the validation study was finished after eight weeks.

Ruggedness/robustness was evaluated using different batches of reagents, solutions and personnel during the eight weeks of the the validation study. Experience from later routine use of the method, and participation in proficiency tests, allows for a more long term evaluation of the robustness of the method.

325

326 3. Results

327 **3.1 Selectivity/specificity**

Full chromatographic separation of the four analytes was achieved with equivalent peak 328 shapes in sample and standards. A typical chromatogram of standard is shown in Figure 1. 329 Isotope labelled internal standards were used in the determination to ensure robust 330 331 determinations since they are known to equalize matrix effects and they make it possible to unambiguously identify an analyte's retention time . Ion ratios were found to be quite similar 332 from day to day and all the ratios were within the maximum permitted tolerances fulfilling the 333 334 criteria for unequivocal identification of all the NF metabolites (Supplementary Table 1). During validation no interfering signals were detected in the blank samples. Later routine 335 analysis revealed that e.g. scampi could contain possible false positive signals for NP-AHD 336 with the same quantifier and qualifier ion within the expected ratio-interval as the analyte 337 (Figure 2). However, the use of the mass labeled internal standard for AHD proved that the 338 339 retention time for these false positive signals were not due to AHD. The signals could possibly be due to other compounds with similar structures. This demonstrates the importance 340 of using isotope labeled internal standards since retention times may vary from injection to 341 342 injection depending on the system stability and matrix load on the column. The use of isotope labelled internal standards is particularly important when determining illegal substances such 343 as NF metabolites, or other compounds where false positives may lead to serious economic 344

damage for the producer. Preferably ¹³C or ¹⁵N labeled internal standards should be used since
they have more similar retention time to the unlabeled analyte compared with ²H labeled
internal standards (see *e.g.* the chromatogram of NP-AMOZ in Figure 1). It is well known that
SEM can contain false positive signals due to the legitimate use of food packaging or food
additives, but to our knowledge, this is the first time a false positive for AHD is reported.

350

351 **3.2 Calibration curves and working range**

Linear working ranges for the NF metabolites spanned from CC α to the highest level in the calibration curve (10 ng/g). Higher concentration levels were not anticipated to be necessary since samples with a concentration of any NF metabolite above CC α would be in violation of the legal limit. Calibration curves were calculated using linear regression analysis with 5-7 concentration levels by forcing the curve through origo and weighting each observation with 1/x. RSDs for the calibration curves were better than or equal to 12% for all analytes, and correlation coefficients were better than 0.99 on all days (Supplementary Table 2).

359

360 3.3 Decision limit (CCα)

Results for CCa determined using the three methods outlined in section 2.7; sample blank 361 362 method, calibration curve method and ISO 11843 method are shown in Table 3. Using the sample blank method, the CC α was found to be between 0.1 ng/g wet weight (w.w.) and up to 363 0.3 ng/g w.w. for AOZ in salmon. The calibration curve method gave CCα values in the range 364 between 0.2 ng/g w.w. for AHD in shrimp up to 0.5 ng/g w.w. for SEM in both salmon and 365 shrimp and AHD in salmon. The ISO 11843 method gave CCa values between 0.2 ng/g w.w. 366 and up to 0.8 ng/g w.w. for AOZ in shrimp. CCa was always found to be under the MRPL (1 367 ng/g w.w.) for all analyte-matrix combinations using all calculation methods. The three 368

methods used to determine CCa had small variations for most analyte-matrix combinations, 369 370 but the difference was larger for *e.g.* AOZ in shrimp where the difference was 0.6 ng/g w.w. between the sample blank method (0.2 ng/g w.w.) and the ISO 11843 method (0.8 ng/g w.w.). 371 $CC\alpha s$ found using the calibration curve approach and the ISO 11843 method were always 372 higher than the CCas found using the sample blank method. This could in part be due to the 373 very low background noise in modern MS/MS-instruments, resulting in very low S/N in 374 375 sample blanks. Since the instrument sensitivity will vary from day to day and from matrix to matrix, the determination of $CC\alpha$ in every instrument sequence has been suggested (E. 376 Verdon, Hurtaud-Pessel, & Sanders, 2006). However, as our results show, CCa is also 377 378 dependent on the method used to calculate it. Nevertheless, all results for CCa for this method 379 will also fulfill the future criteria for reference points for action (RPA) proposed at 0.5 ng/g, except when using the ISO 11843 method to calculate the CC α for AOZ in shrimp (0.8 ng/g 380 381 w.w.) (EC, 2019).

382

383 3.4 Detection limit (CCβ)

384 CC β was determined following the determination of CC α by spiking 24 matrix blanks 385 of each matrix at CC α . Since the values of CC α was found to be both matrix, analyte and 386 calculation dependent, samples were spiked at a selected level in the ranges of the found 387 CC α s. CC β results were then calculated for each matrix and analyte combination and the 388 results are shown in Table 3. CC β was found to be between 0.3 ng/g w.w. for AHD in shrimp 389 and 0.9 ng/g w.w for AHD in cod.

390

391 **3.5 Recovery**

Recovery was evaluated by spiking experiments. Samples of salmon, shrimp and cod were 392 393 spiked at three levels and the results are shown in Supplementary Table 3. Recoveries for all analytes were found to be in the ranges 89-101% for salmon, 93-115% for shrimp and 84-394 112% for cod. The results were within the criteria of 50-120% for concentrations at or below 395 1 ng/g and mostly within 70-110% for concentrations above 1 ng/g and up to 10 ng/g. The 396 exceptions were 2 ng/g of AHD in shrimp (115%) and 2 ng/g AHD in cod (112%). 397 Nevertheless, such concentrations are well above the CCa and any detection at this 398 concentration would therefore be illegal. 399

400

401 **3.6 Trueness**

To evaluate the trueness of the method a proficiency test of shrimp (SEM consensus value 402 2.77 ng/g w.w.) was also analyzed during validation. We found 2.7 ng/g w.w. of SEM in this 403 sample which gave a z-score of 0.0. The method has participated in many proficiency tests 404 405 since it was first validated and the results have shown that other analyte-matrix combinations also result in z-scores less than |2| (Figure 3). In 2016 we had an exceptional high value for 406 tissue bound AOZ in shrimp with a z-score +3.3. When this result was reported we had used 407 408 washing procedure 1 to report the amount of tissue bound NF metabolites. Due to our high zscore in this proficiency test we decided to try to modify our washing procedure by adding a 409 stone for better mixing during shaking. At the same time we also tested a different washing 410 procedure 2 for comparison, published by the EURL for veterinary drugs (see 2.4 for details), 411 412 and also modified this washing procedure by adding the stone for better mixing. Following 413 the modification with the stone both washing procedures 1 and 2 produced satisfactory results with z-scores of -1.4 for washing procedure 1 and -1.7 for washing procedure 2. However, we 414 noticed that our results were now on the lower side of the z-score scale, and we suspected this 415 to be due to the increased loss of sample/analyte due to adsorption to the stone. We therefore 416

417 compared the old washing procedure 1 and the new washing procedure 2 without adding the 418 stone (Supplementary Table 4 and 5). We found that the new washing procedure 2 was more 419 efficient in removing particularly SEM and AOZ, which could explain why we got the high z-420 score for AOZ when using the old washing method 1. We therefore changed to the new 421 washing procedure 2 for future analyses when determining tissue bound NF metabolites.

422

423 **3.7 Precision as repeatability**

Repeatability (Supplementary Table 6) determined by analyzing salmon spiked at 1, 1.5 and 2 ng/g w.w using six replicates on three different days gave RSDs between 0.7% for AMOZ at 1 ng/g w.w. and 10% for AHD at 1 ng/g w.w.. Hence, repeatability of the method was within the 12% criteria for half the value predicted for reproducibility by the Horwitz equation, and well within the upper criteria of two thirds (15%) of the value predicted by the Horwitz equation.

430

431 **3.8 Intermediate precision/within-laboratory reproducibility**

Intermediate precision for salmon was determined by combining the results from repeatability testing (3.7) with two replicates on three different days, giving a larger dataset spanning six days in total (Supplementary Table 7). For shrimp and cod two replicates were analyzed on six different days. Intermediate precision RSDs ranged from 3.1% for 2 ng/g w.w. SEM in shrimp up to 17% for 1 ng/g w.w. of both AHD in cod and AOZ in shrimp. Hence, the within-laboratory reproducibility of the method was within the maximum permitted tolerance of 23% predicted by the Horwitz equation.

439

440 **3.9 Measurement uncertainty (MU)**

A "worst case" approach was chosen for the estimation of measurement uncertainty by
selecting the highest within-laboratory reproducibility per matrix and using this RSD in the
calculation of MU as specified in section 2.7. Calculated values for MU are shown in
Supplementary Table 8. The within-laboratory reproducibility per matrix-analyte combination
was found to vary from 7.2% for SEM in cod to 17% for AHD in cod and AOZ in shrimp.
The associated calculated MU was estimated to vary between 20% for SEM in cod to 40% for
AHD in cod and AOZ in shrimp.

448

449 **3.10 Stability**

Stability of the standards were investigated as described in section 2.7. One of the samples 450 from each storage condition was worked up together with the other samples on every day of 451 the validation. Linear regression analysis of the results was performed following complete 452 453 validation after eight weeks and the results are shown in Supplementary Table 9. The analysis 454 revealed no significant negative trend (p-values > 0.05) during the validation period for the analytes with the exception of AOZ stored in the dark at room temperature (p-value 0.05). 455 456 However, this significant result could be due to chance, since AOZ stored at both -20 °C and +4 °C were not found to decrease significantly. The lack of significant trends should be 457 interpreted with caution since all analytes, except AHD (+0.01 ng/day) stored in the dark and 458 AMOZ in cod (0.00 ng/day), displayed negative correlation coefficients has high as -0.08 459 ng/day. The regression coefficients for SEM (average -0.05 ng/day) were significantly (p < 460 461 0.005) more negative than for the other NF metabolites (average -0.01 ng/day). This is consistent with previous investigations where stock solutions of NF metabolites were found to 462 be stable for at least 10 months if stored at -4 °C, whereas more dilute solutions were found to 463

be stable for three months with the exception of SEM which was found to decompose at a rate
of 5% during 3.9 months in dilute solutions (Cooper & Kennedy, 2007; Radovnikovic,
Moloney, Byrne, & Danaher, 2011). Our decomposition rate for SEM was faster with an
overall average of 1% per day. We also found that regression coefficients for SEM in matrix
(average -0,08 ng/day) were significantly more negative (p< 0.0005) than regression
coefficients in solution (average -0.04 ng/day), so care should be taken when storing samples
with SEM.

471

472 3.11 Ruggedness/robustness

The method was found to be robust during the eight weeks of validation and has also proved to be robust during routine use performed by different personnel, using different batches of regents, solutions and standards over several years. The main factors, which may influence the robustness, was found to be the preparation and storage of standards and the pH adjustment.

478

479 **4.** Conclusion

The method was successfully validated and implemented for the determination of both tissue 480 bound NF metabolites and total (free + tissue bound) NF metabolite residues and has been 481 used in routine analysis for many years. Washing procedure 2 turned out to be the most 482 efficient in removing free metabolites when determining tissue bound NF metabolites, 483 particularly for SEM and AOZ. The two washing procedures we investigated differs to some 484 extent in type and amount of solvent, but the main difference appears to be the more thorough 485 homogenization, followed by centrifugation, for washing procedure 2 compared to the more 486 gentle mixing and decanting of washing procedure 1. 487

A new false positive for AHD is reported for the first time and an isotope labeled internal
standard is required for identifying this new false positive. This is because the false positive
has identical response in the mass spectrometer compared to AHD, but the slight shift in
retention time, compared to the isotope labeled internal standard, makes it possible to identify
this false positive.

493 The CC α of the method was satisfactory with respect to the MRPL using all the three methods 494 for its calculation. Small differences in CC α were found when using the different calculation 495 methods. Calculations using the sample blank method generally gave the lowest CC α of the 496 investigated calculation methods.

497 No statistically significant negative trends were seen from the stability investigations,

although most NF metabolites had negative correlation coefficients. SEM had the highest

anegative correlation coefficient and decomposed at a rate of around 1% per day.

500 The analytical method was found fit for purpose to determine both bound and total NF

501 metabolites. The selection of validated matrixes, results from proficiency tests and routine use

over many years ensures that the method is horizontally applicable to seafood as a foodcategory.

504

505 Acknowledgements

We wish to thank the Norwegian Food Safety Authority for financing this study and artistMalin Igland Valdersnes for producing the graphical abstract.

508

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